



Cavity Heat Treatments

Ari Palczewski

USPAS Course:

SRF Technology: Practices and Hands-On Measurements

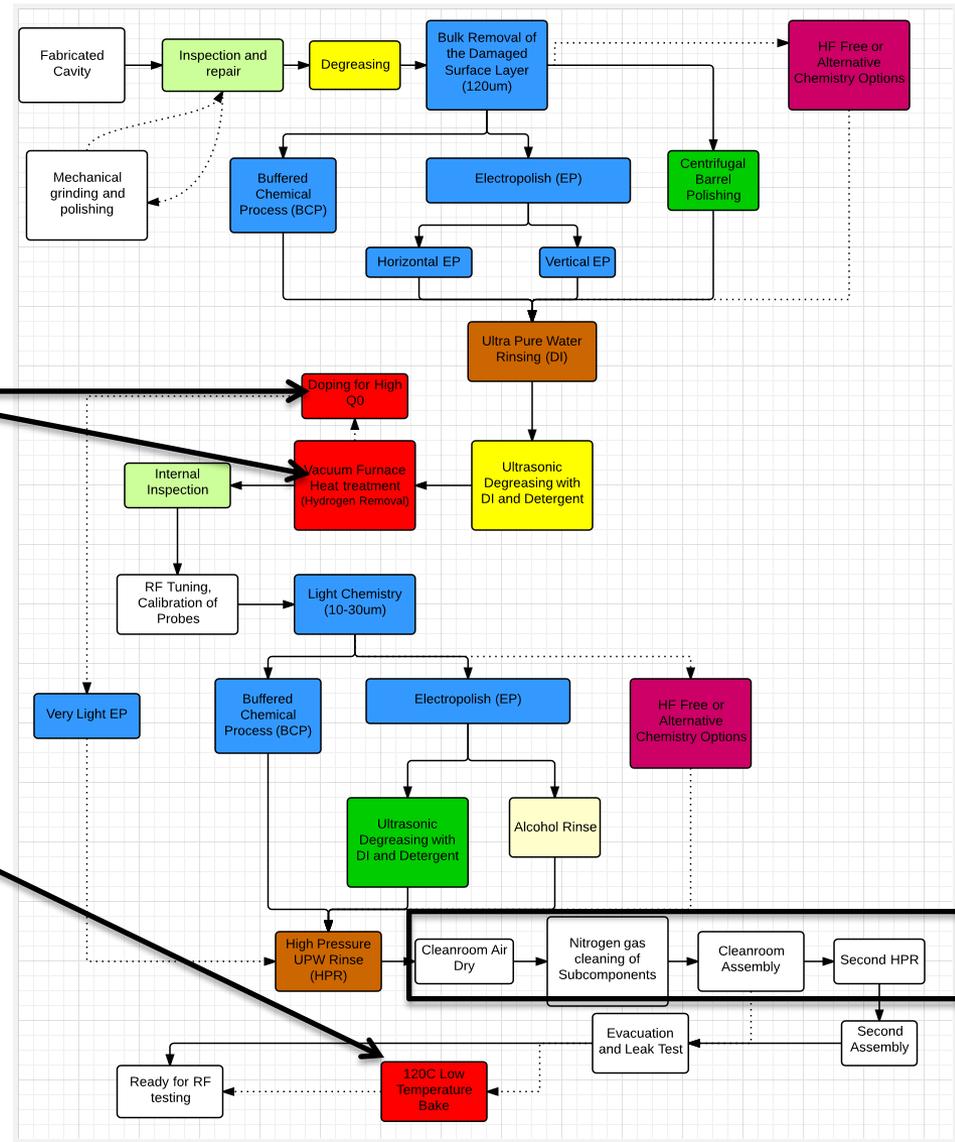
January 2015

Cavity preparation – heat treatments

Back to John's Talk – 9T SRF cavity preparation

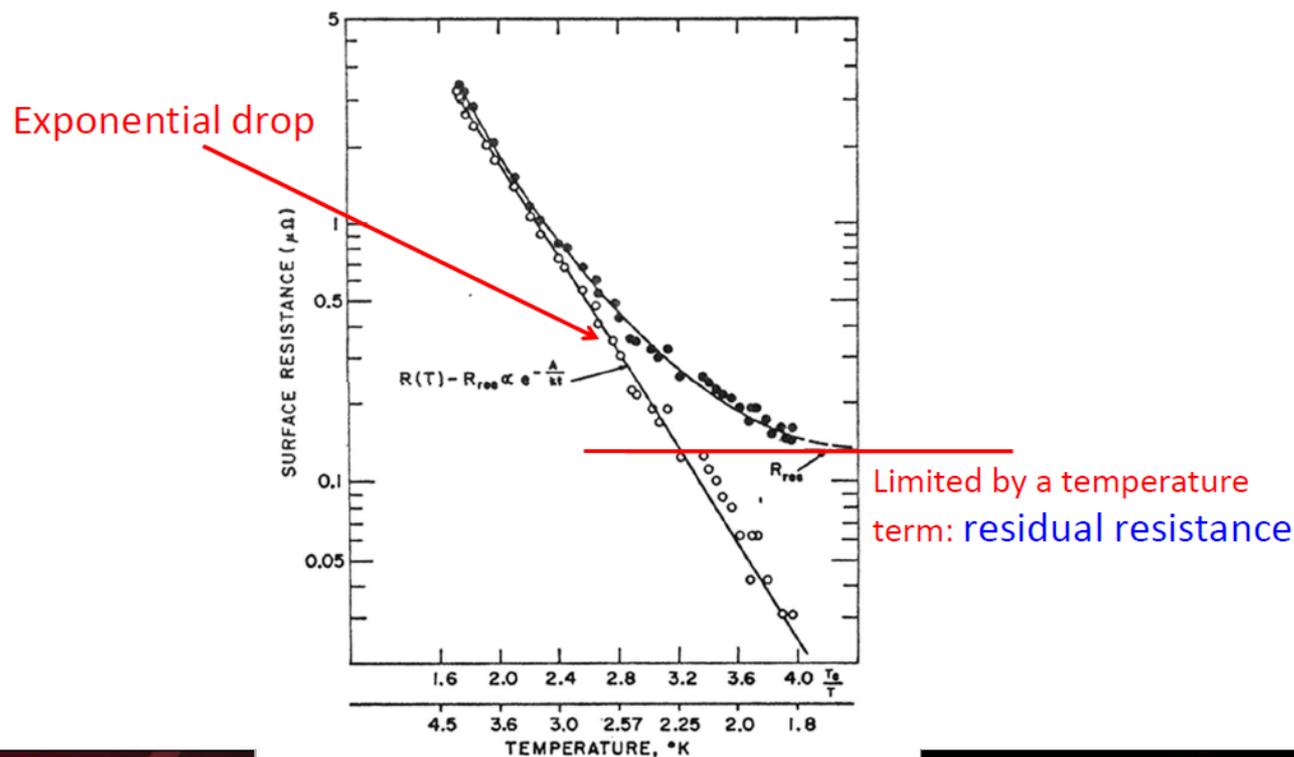
Furnace treatment either with nitrogen doping or without

120C bake



Reminder from Geng T1 and T3 - losses

Losses in Superconductor: measurements



We will see how the BCS as well as residual terms are effected by different heat treatments

Exponential drop part is what we call “BCS” resistance – it is the temperature dependant part

Residual term is not temperature dependant.

Heat treatments

Cavity Bakeing

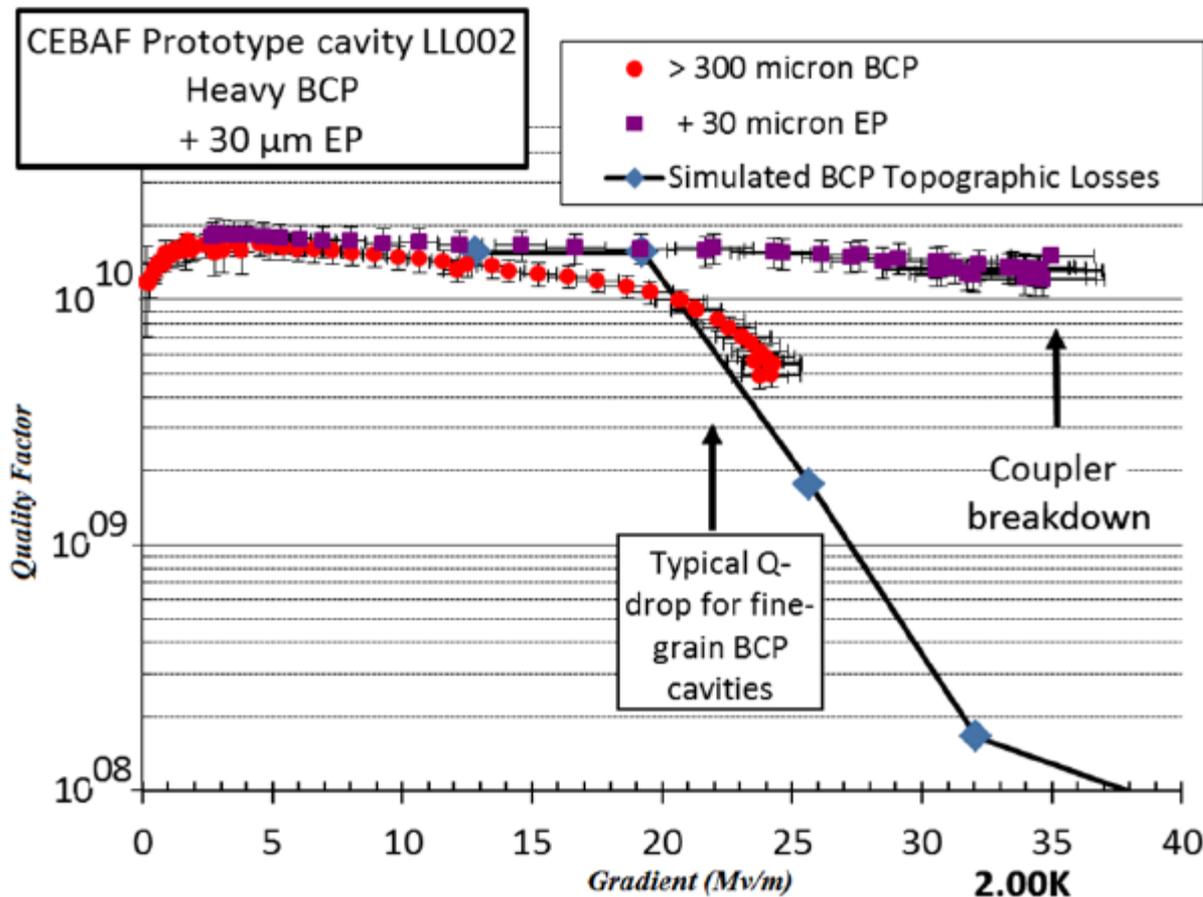
- Temperatures from 50C to 300C – usually 120C
- Usually performed after final chemistry
- Done on test stand while the cavity is the vacuum vessel
- Used primarily to remove high field Q-slope – and enhance Q0 @ 2k in certain cavities
- Used to removing residual water in cavity
- Used to reduce multi-packing by changing secondary yield coefficient

Cavity furnace/heat treatment

- Temperatures between 400C to 1800C (600C to 1400C modern)
- Usually done before final chemistry
- Usually with cavity open in large vacuum furnaces
- Primarily to remove hydrogen from manufacturing (welding and bulk chemistry)
- Sometimes used to purify niobium (T>1000C)
- Sometimes used to “Soften” niobium (large grain stress from stamping)
- Used to dope cavities with Nitrogen and Titanium for high Q0

“Bakeing”

From Reece 13T – topography not “Q-slope”

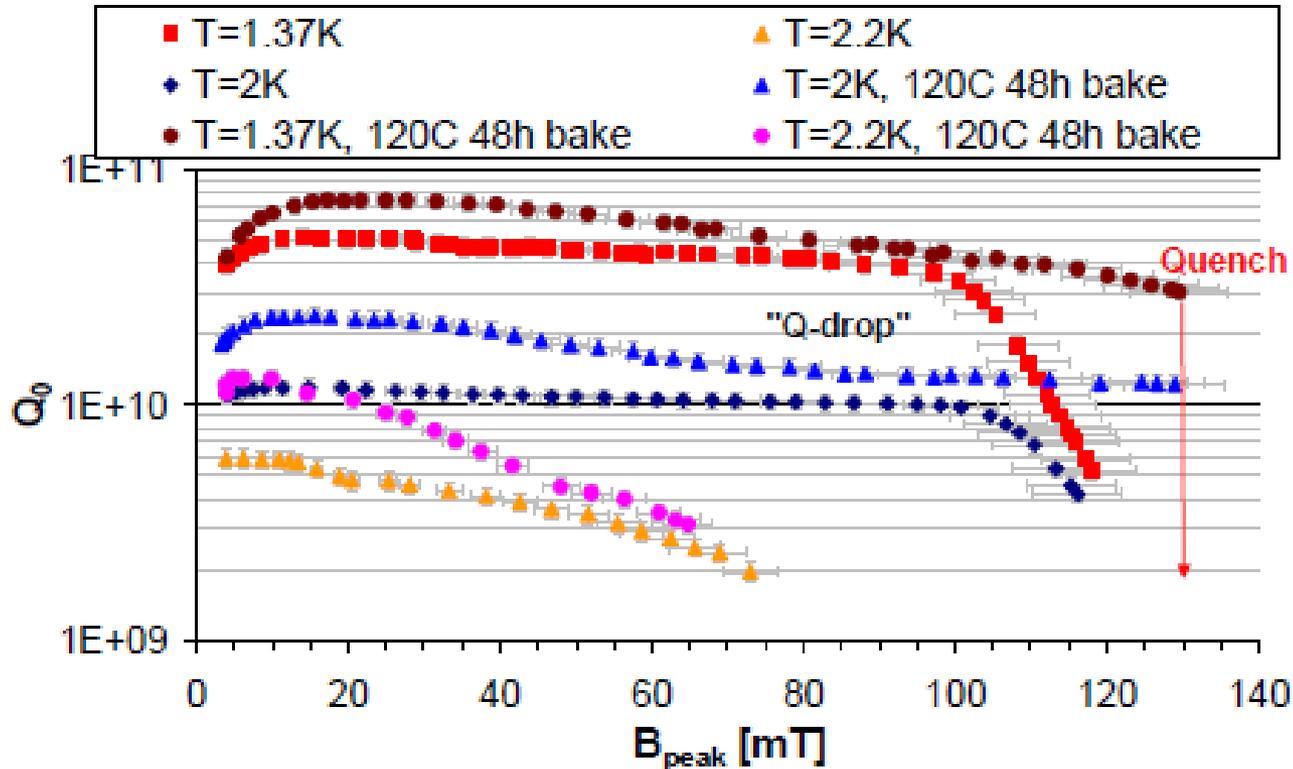


The BCP surface 600C for 10H, but still shows a Q-slope – this if from topography (mostly)

The Ep'ed surface was also baked @ 120C for 24 hours, removing the high field Q-slope

C. Xu, C. E. Reece and M. J. Kelley, "Simulation of non-linear SRF losses derived from characteristic Nb topography: comparison of etched and electropolished surfaces," <http://arxiv.org/abs/1406.7276>, 2014.

Q-slope and bake BCP - LG cavity

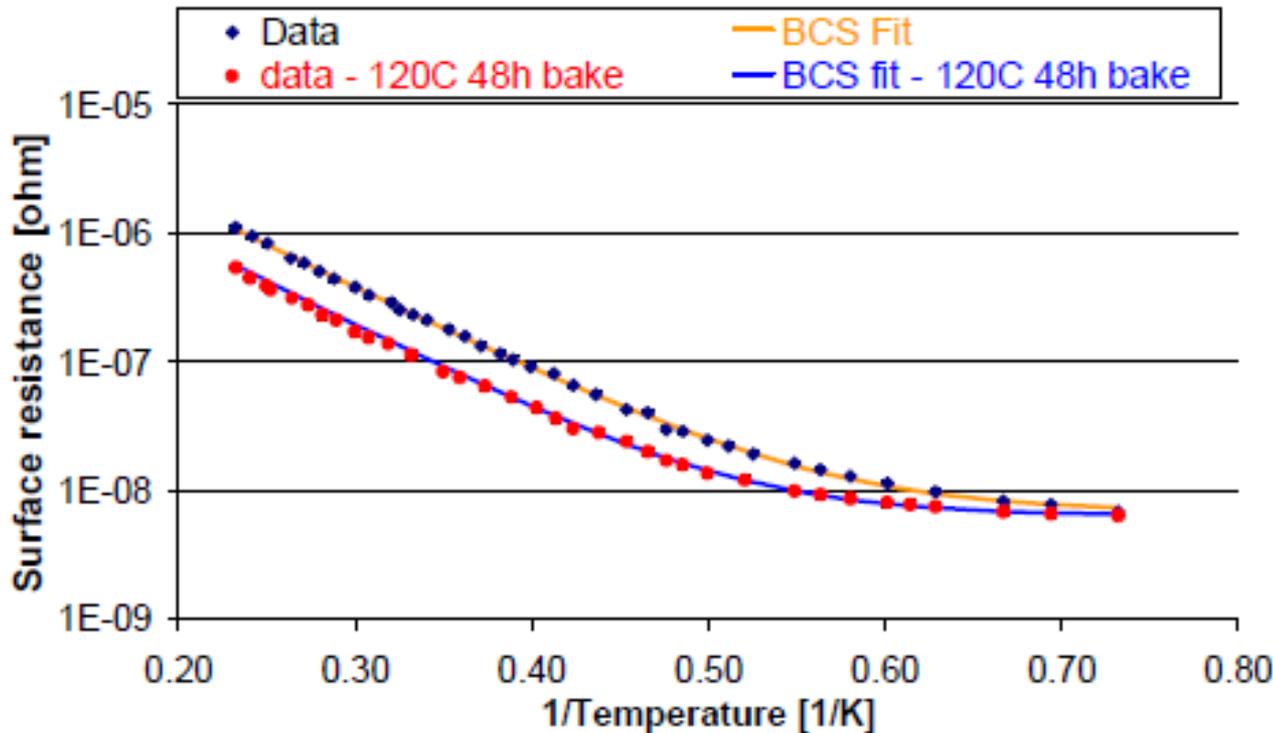


Q-slope from BCP'ed cavity is because cavity was not baked, not a surface roughness like the fine grain cavity on previous slide

Figure 14: Q_0 vs. B_{peak} before and after 120°C, 48h baking.

G. Ciovati et al. Effects of low temperature baking on niobium cavities
<http://srf2003.desy.de/fap/paper/WeO14.pdf>

“Residual” vs. “BCS” before and after bake



Cavity shown is Large grain from previous slide, but effect is the same for fine grain cavities

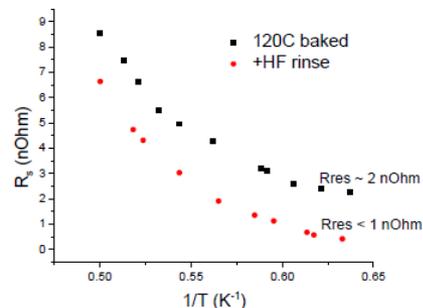
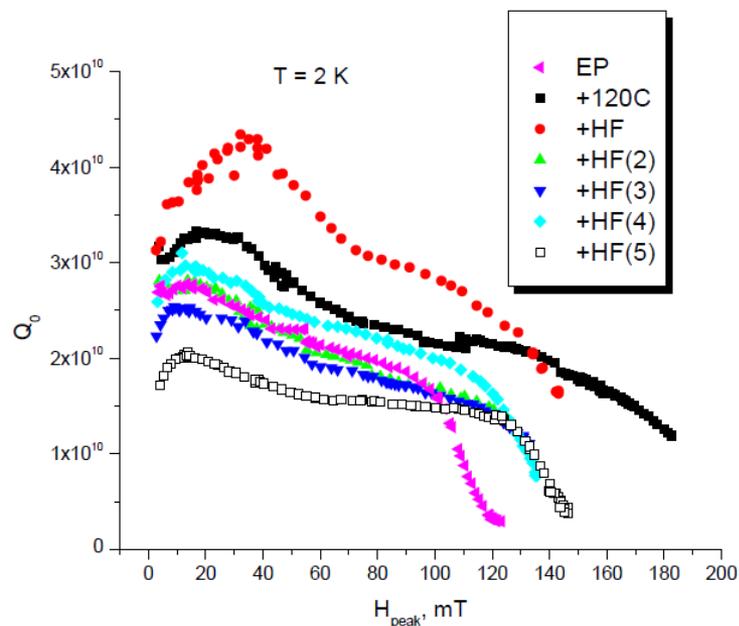
- BCS resistance goes down with bake
- Residual resistance goes up
- @ 2.0K and 1.3 to 1.5 Ghz Q0 goes up

Figure 15: Surface resistance vs. 1/temperature before and after 120°C, 48h baking.

G. Ciovati et al. Effects of low temperature baking on niobium cavities
<http://srf2003.desy.de/fap/paper/WeO14.pdf>

Side note – HF rinsing on baked cavities

Results on EP fine grain (tumbled)



- ✓ Single HF rinse after mild baking significantly improves medium field Q_0
- ✓ Multiple HF rinse cycles do bring the high field Q_0 slope back
- ✓ Onset field is still higher than before baking by ~ 25 mT after total 5 HF rinse cycles
 - ✓ Further rinses in queue

HF rinsing does not change BCS term from 120C bake, but lowers residual term – Q_0 in mid field goes up

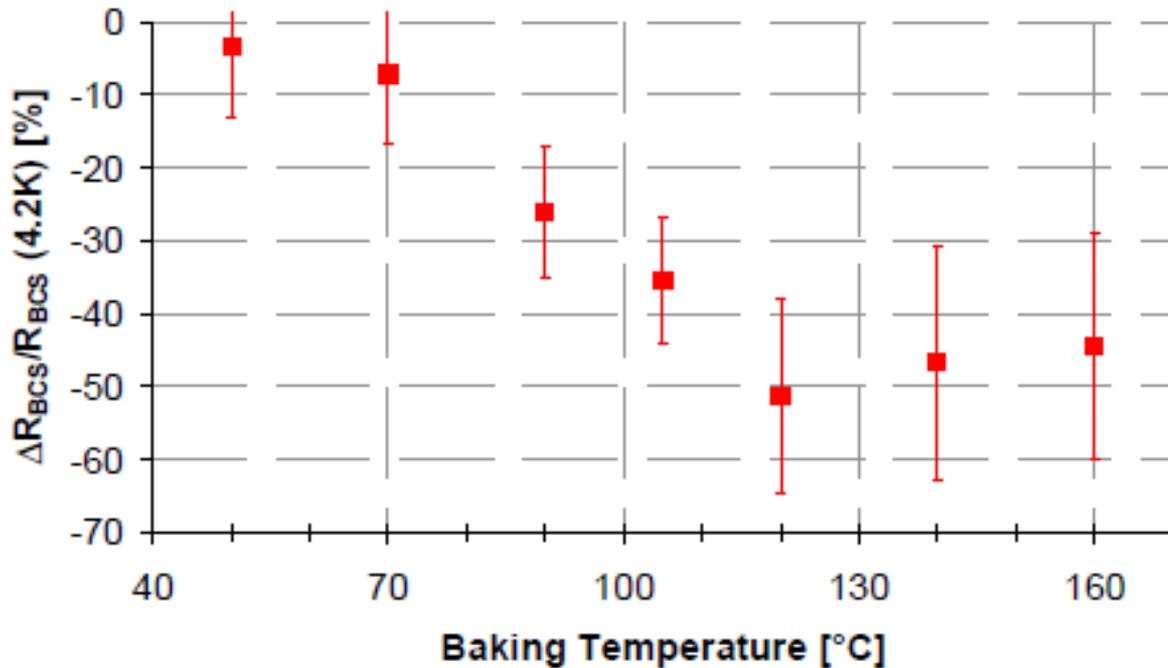
2/6/12

A. Romanenko - All Experimenters Meeting

13

https://www.fnal.gov/directorate/program_planning/all_experimenter_meetings/special_reports/Romanenko_SCRF%20Cavities_02_06_12.pdf

Optimal temperature for bake – Large grain



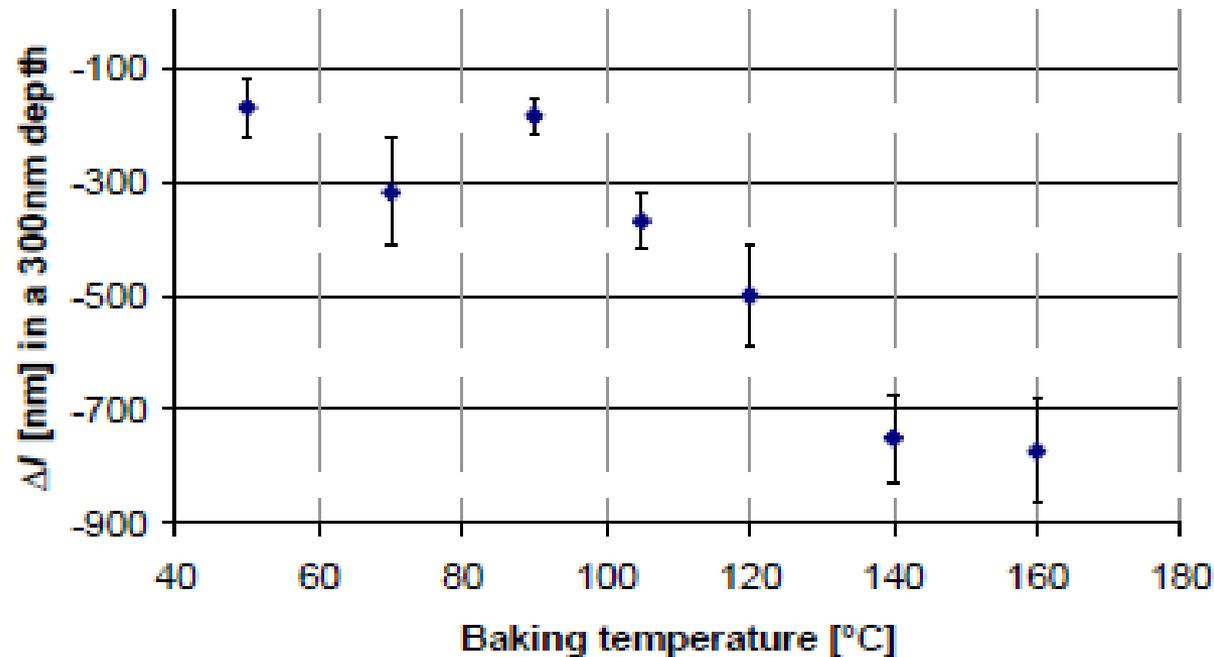
From coupons, BCS change is because mean free path changes, coupled to cavity data

120C to slightly above is the sweet spot for best Q0 @2K @~1.3 to 1.5

Figure 8: Variation of BCS surface resistance at 4.2K as a function of the baking temperature.

G. Ciovati et al. Effects of low temperature baking on niobium cavities
<http://srf2003.desy.de/fap/paper/WeO14.pdf>

Optimal temperature for bake – Large grain

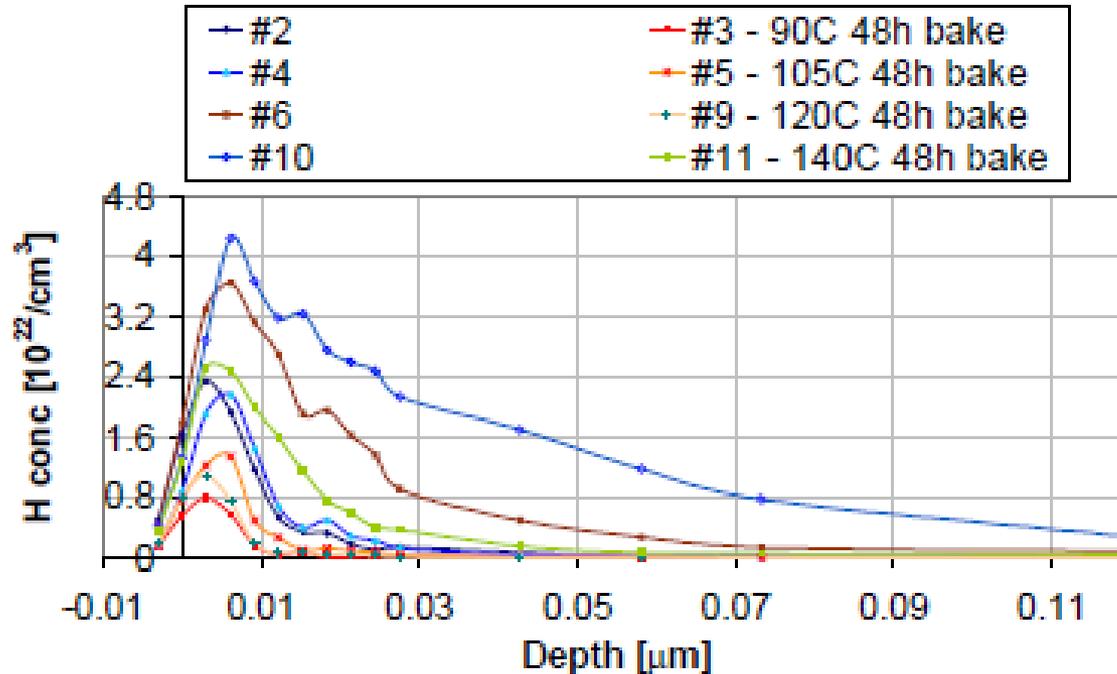


Baking changes the mean free path at the surface

Figure 9: Variation of mean free path as a function of the baking temperature.

G. Ciovati et al. Effects of low temperature baking on niobium cavities
<http://srf2003.desy.de/fap/paper/WeO14.pdf>

Optimal temperature for bake – Large grain



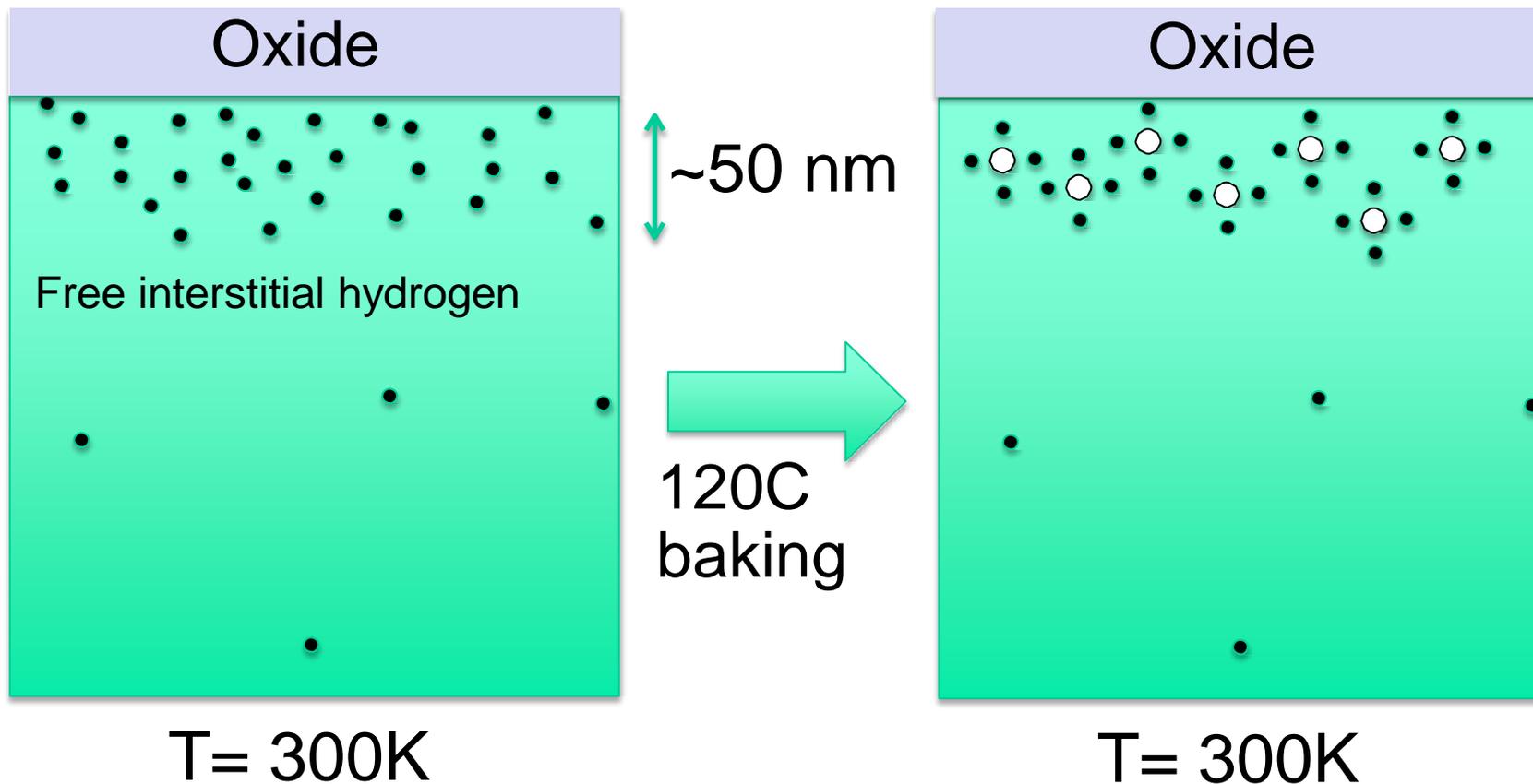
Hydrogen content at surface is greatly reduced by bake, room temperature!

Figure 13: Hydrogen concentration vs. depth for samples baked and not baked.

G. Ciovati et al. Effects of low temperature baking on niobium cavities
<http://srf2003.desy.de/fap/paper/WeO14.pdf>

120 C Baking Effect

Vacancies trap H, Prevent Nb-H formation

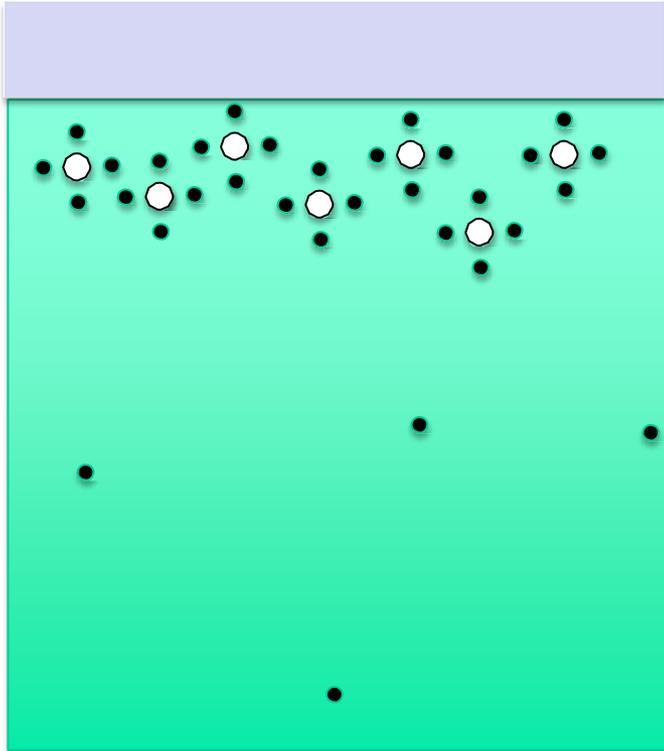


Alexander Romanenko

A. Romanenko, C. J. Edwardson, P. G. Coleman, P. J. Simpson, *Appl. Phys. Lett.* **102**, 232601 (2013)

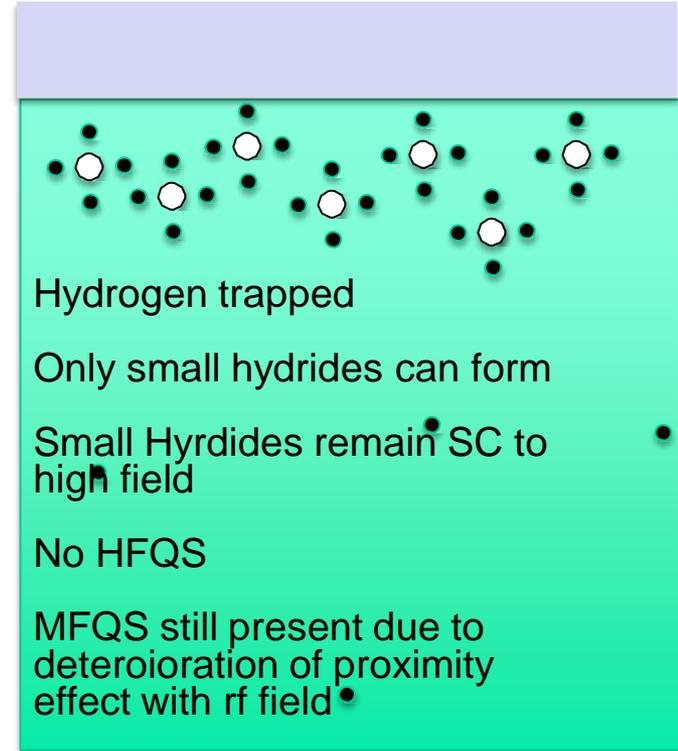
Cool down of 120C baked niobium

Oxide



T= 300K

Oxide



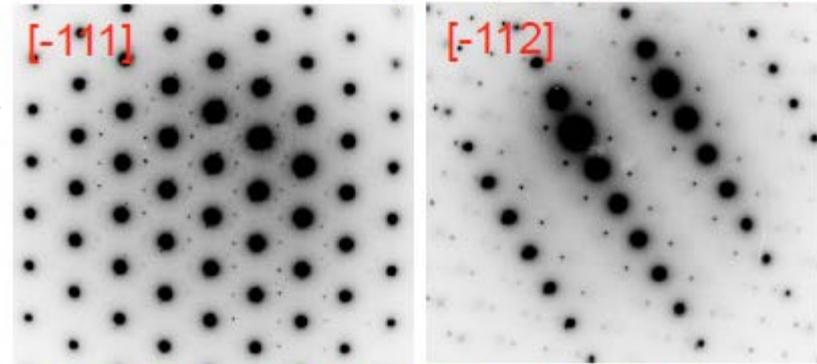
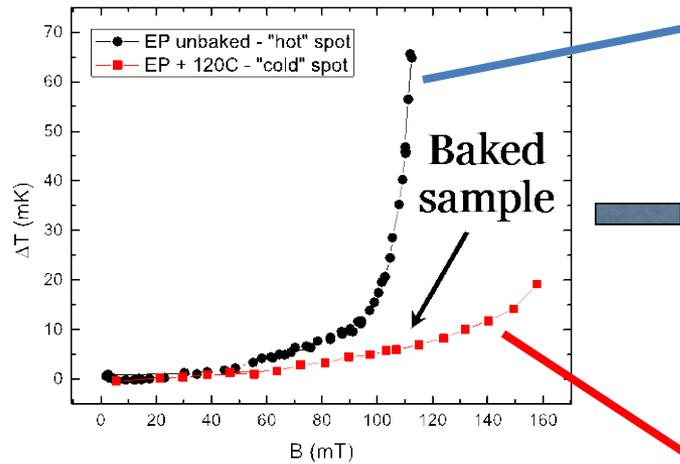
T= 2K

Alexander Romanenko

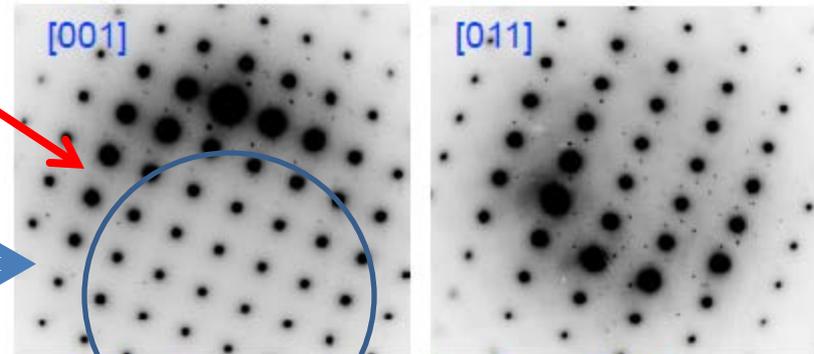
120 C Bake Inhibits Nb-H formation Romanenko (SRF 13)

Cold: 120C in situ bake for 48hours

Hot: no such bake



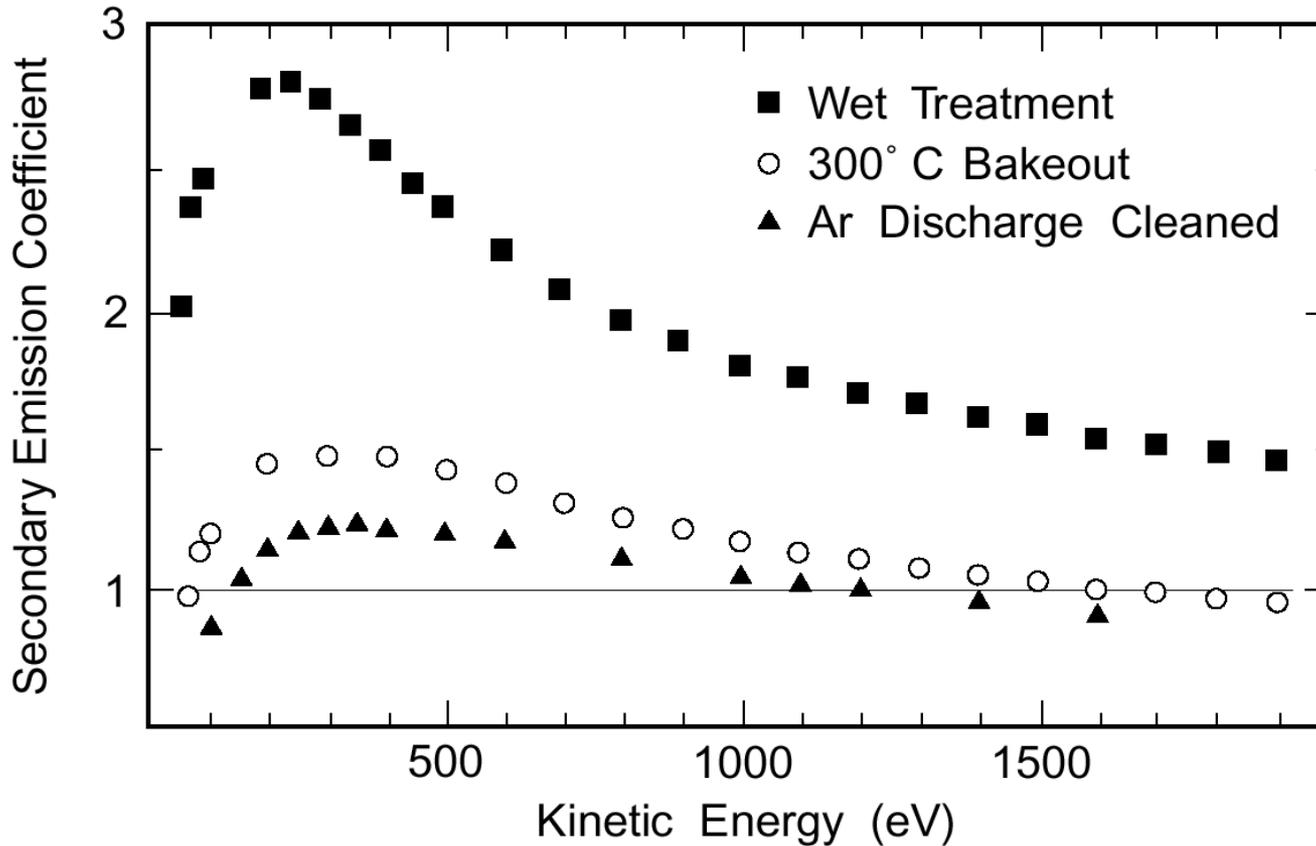
Hot spot: 44-68% of probed spots



Baked spot: 26-29% of probed spots

- Substantial reduction of Hydride formation after 120 C Bake

Multipacting reduction by bake



Baking can reduce secondary emission coefficient so that Multipacting is less prevalent – also has been shown to work for 120C

http://uspas.fnal.gov/materials/08UMD/SRF_Limitations.pdf

High Temperature heat treatments

1970s → ~1800 °C UHV HT for ~10 hrs.

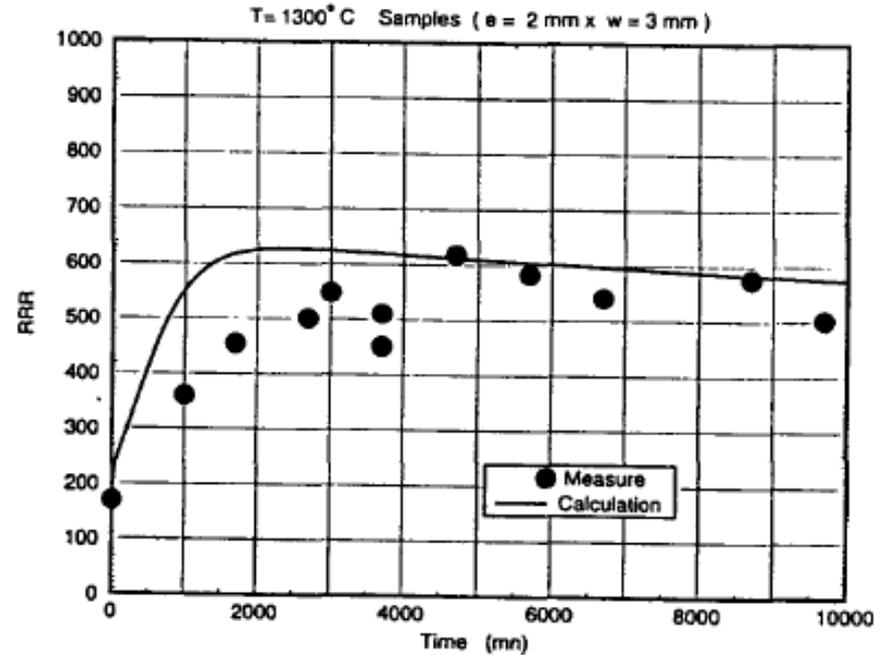
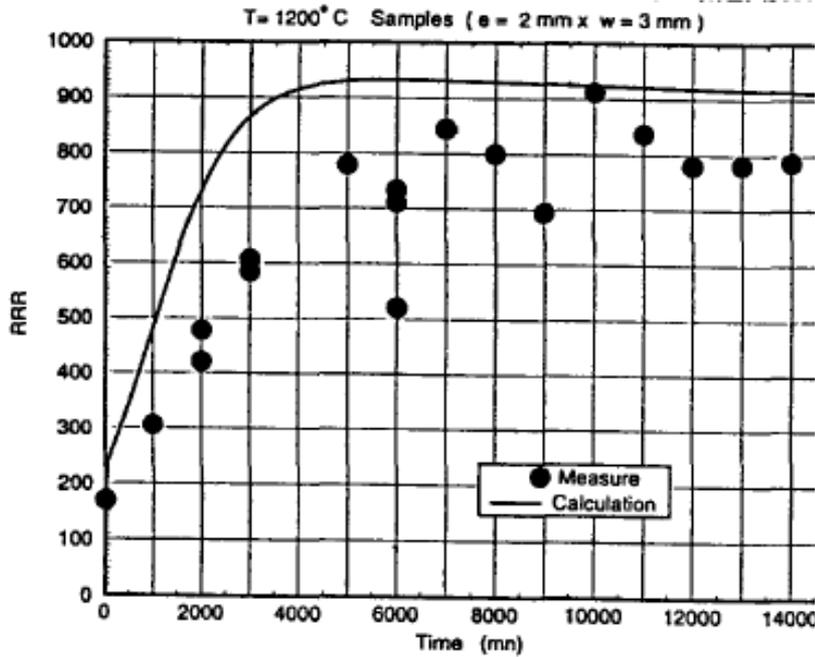
1980s → ~1300 °C solid state getter, such as Titanium, was used in-side the furnace to "post-purify".

2000s → **600 10h -800 °C 2-3h**, mainly just to degas hydrogen absorbed by the Nb during cavity fabrication and surface treatments.

2010's → clean furnace studies from 600 to 1400C to reduce need for final chemistry

2012 → "doping" "polluting" "contaminating" cavity @ 800 to 1400C with titanium and Nitrogen – Extended Q-rise

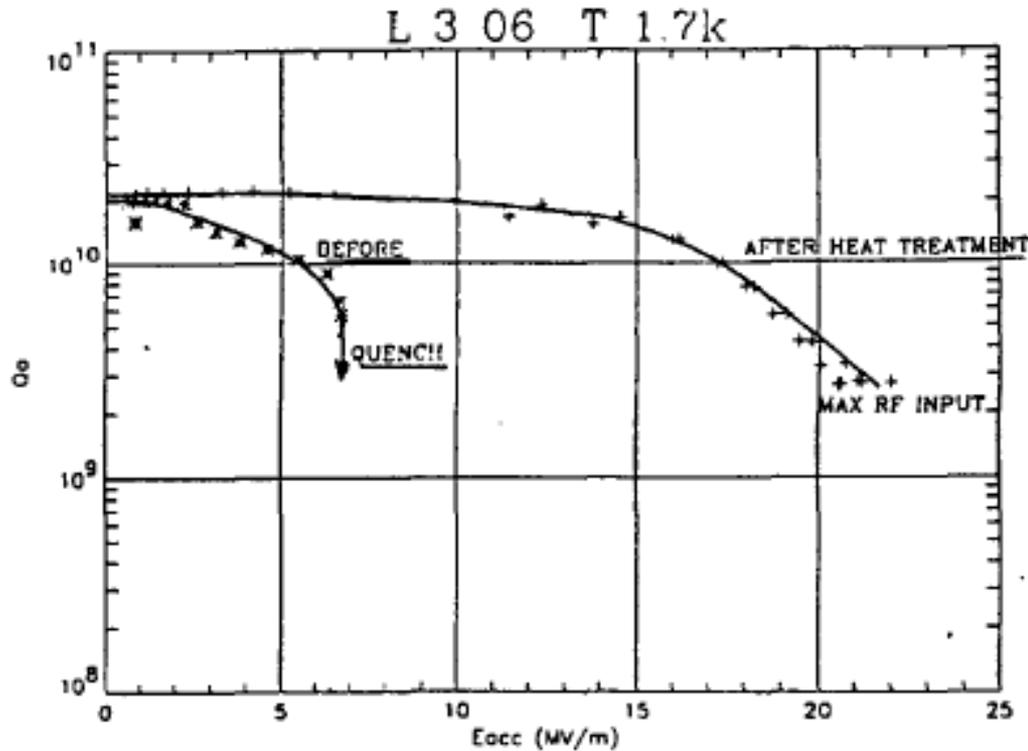
Nb PURIFICATION by Ti GETTERING



Titanium gettering to improve RRR of cavities , remove impurities

H. Safa, Proceedings of the 1995 Workshop on RF Superconductivity

Nb PURIFICATION by Ti GETTERING



Gettering increases RR of cavity with also increases thermal conductivity γ at low temperatures.

Improves quench field from localized defects

Figure 5 – A 3-cell 1.5GHz cavity having a quench at an accelerating field of 6.5MV/m due to an identified defect. After heat treatment, its quench level was pushed up higher than 22MV/m.

H. Safa, Proceedings of the 1995 Workshop on RF Superconductivity

From Geng T1 and T3 – Q-Disease

Symptom of Residual Losses

- Q-disease
 - Q_0 at low field degrades when cavity parked at a temperature 70-150 K for extended period of time
 - Similar effect when cavity cool down rate is slower than 1K/min in passing 70-150 K

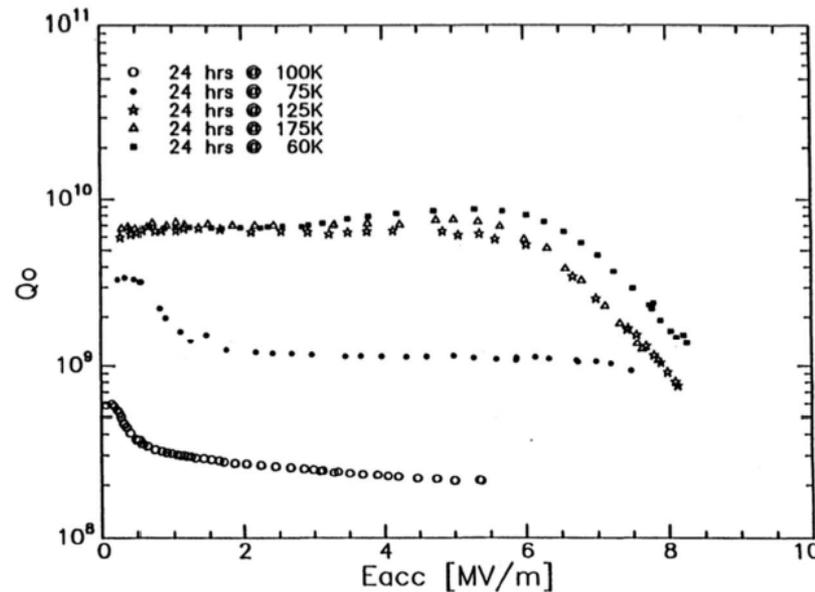


Figure 1 : Eacc – Dependence of Q – Degradation on "Holding" Temperature

J. Halbritter, P. Kneisel, K. Saito, SRF1993

Cavities which are not high temperature heat treated after heavy weld manufacturing, after bulk chemistry or mechanical polishing All show Q-Disease

From Geng talk – minimize Q-disease

Overcoming Residual Losses

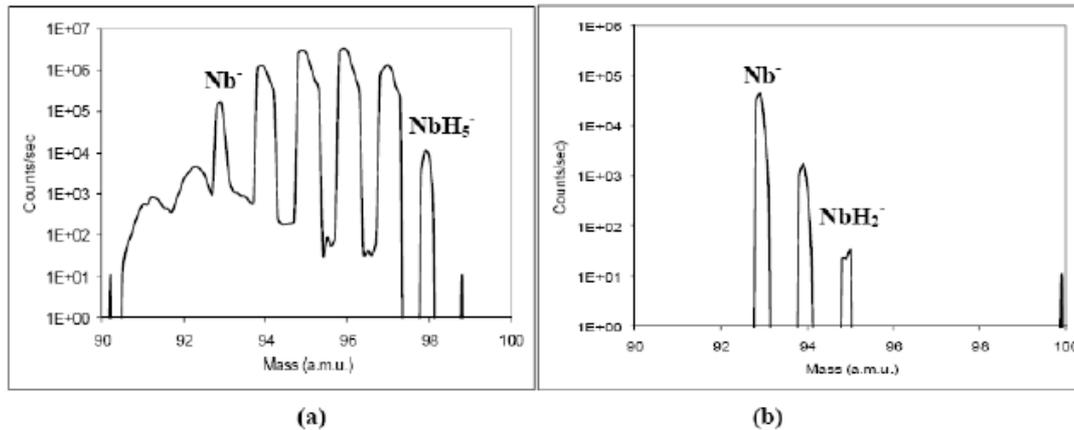
- Minimize H uptake from processing
 - BCP etching at $< 15\text{ }^{\circ}\text{C}$
 - “H free” EP
- Hydrogen out-gassing in vacuum furnace
 - $800\text{ }^{\circ}\text{C}$ x 2hr
 - Or at lower temperature for longer time
- Minimum or no chemistry after out-gassing



Standard (600-800 °C) Furnace Treatment



The standard furnace used for the high-temperature heat treatment of SRF cavities is an ultra-high-vacuum furnace with molybdenum hot-zone; molybdenum (or tungsten) resistive heating elements and cavities are heated by radiation from the heating elements.



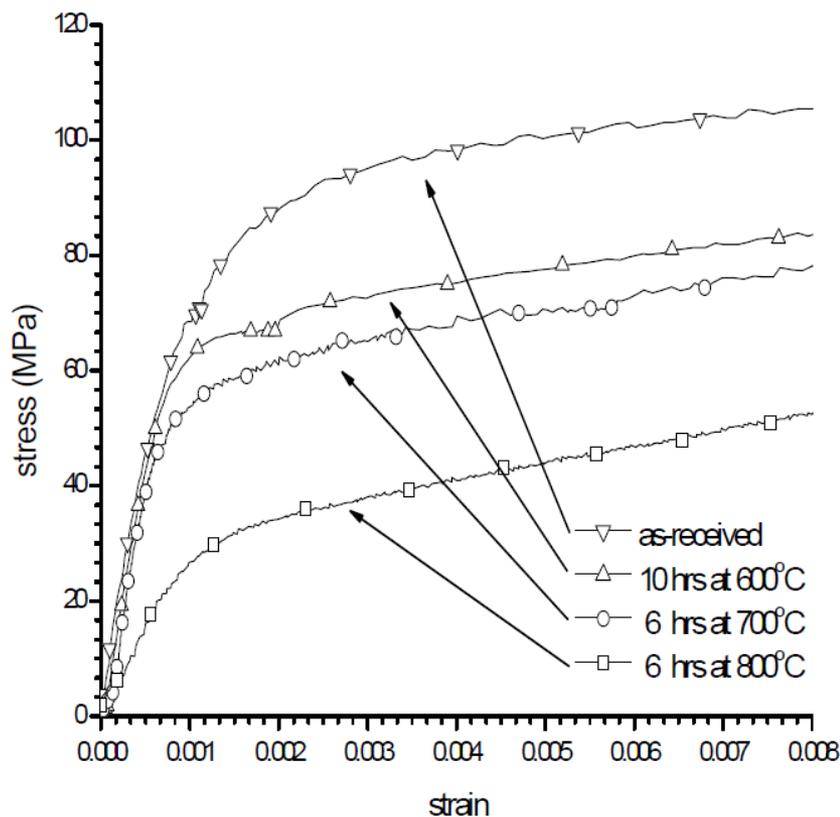
High temperature annealing removes gross hydrogen

FIGURE 1. SIMS mass spectra showing difference in H between (a) non-heat treated and (b) heat treated sample.

Ciovati et al, PRSTAB 13, 022002 (2010)

Physical properties with heat treatment – FG 4K

Apparent YM Issue



Thomas Jefferson National Accelerator Facility
Institute for SRF Science and Technology

SRF 2003

3 September '03

Operated by the Southeastern Universities Research Association for the U.S. Department of Energy

Talk G.R. Myneni - WEO11



USPAS SRF Course Jan. 2015



Physical properties with heat treatment – FG 4K

TD data summary

Summary of the TD niobium mechanical properties

Niobium	Yield Strength (KSI)		Tensile Strength (KSI)		% Elongation		RRR	Hv
	SSR	FSR	SSR	FSR	SSR	FSR		
ASR	7.4	7.9	21	24	44	48	260	52
600 C	7.0	7.5	21	22	48	49	300	47
800 C	5.7	--	19	--	47	--	350	43
1250 C	4.5	6.3	15	19	32	33	375	36

SSR ~ 5.5E-5

FSR ~ 2.0e-4 up to Yield point and 1.0e-3 until break



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Talk G.R. Myneni - WEO11



No wet chemistry after heat treatment - JLab

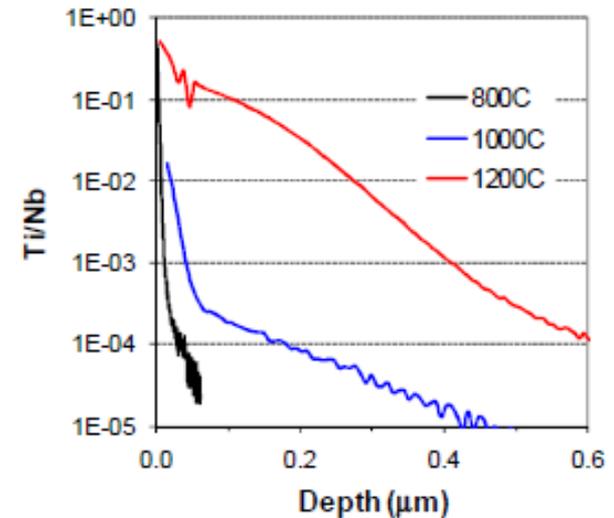
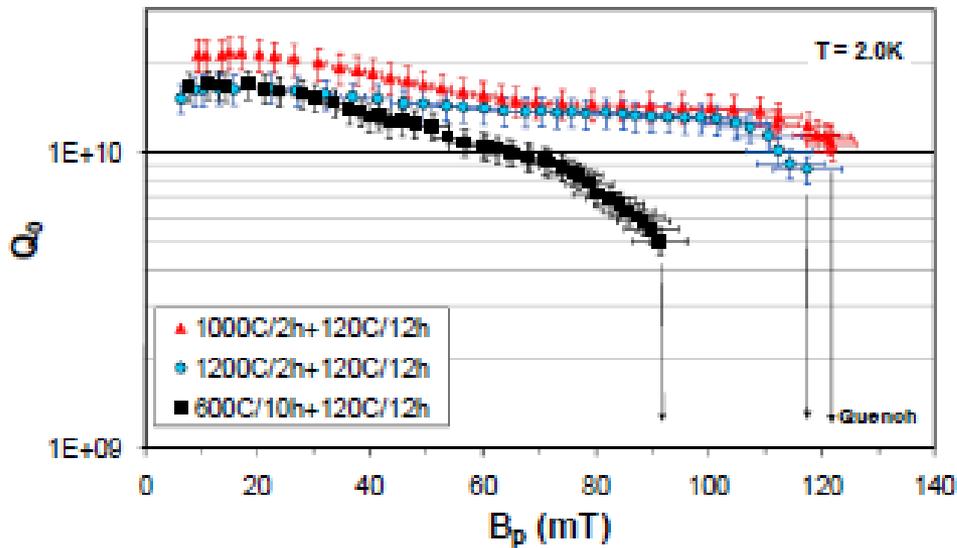
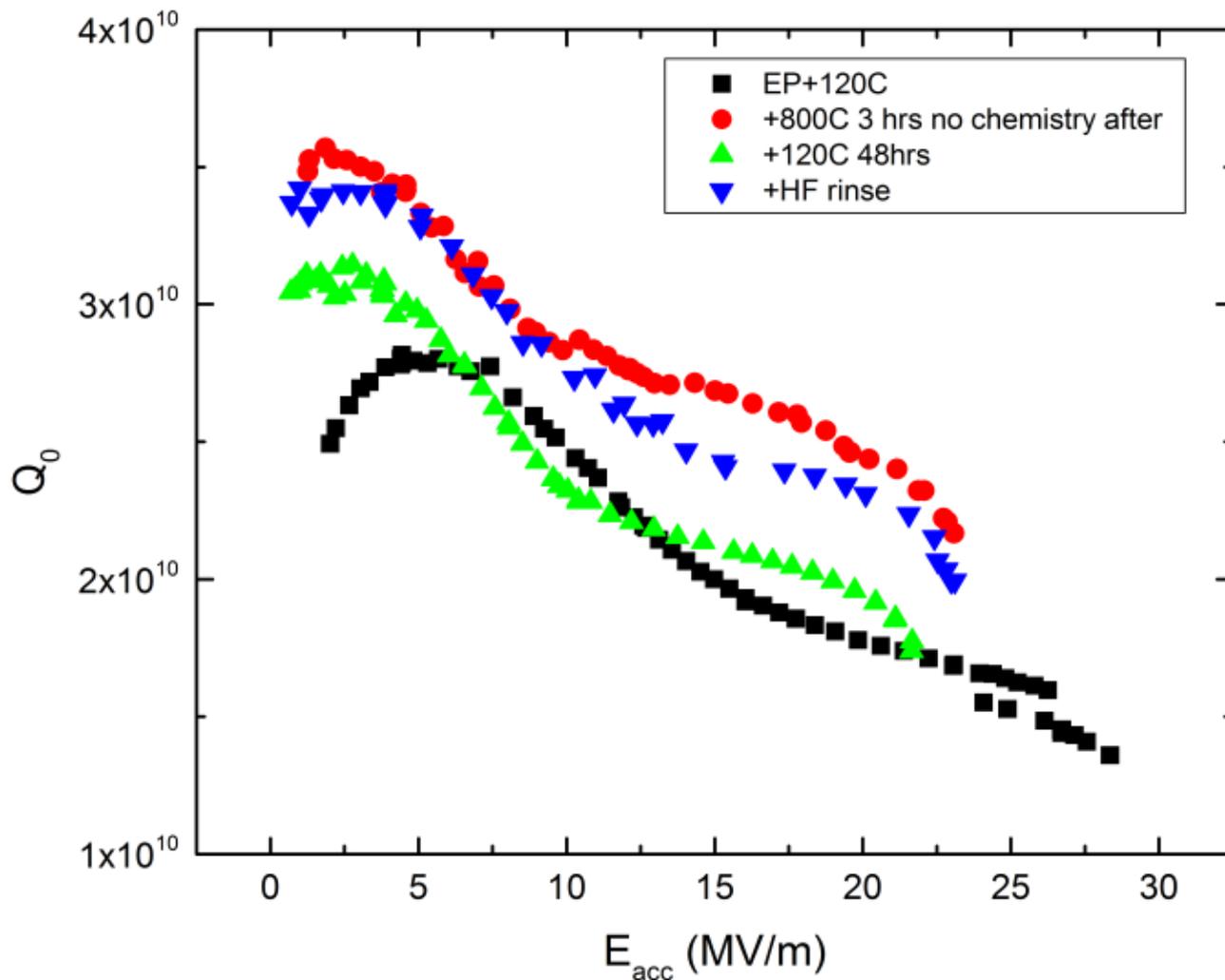


Figure 8: Depth profile of Ti in Nb samples heat treated at different temperature.

- Titanium contamination from furnace on surface no matter what heat treatment
- Large grain cavities are not limited by contamination except for Q slope

G. Ciovati et al. SRF2013 Chicago TUPO051

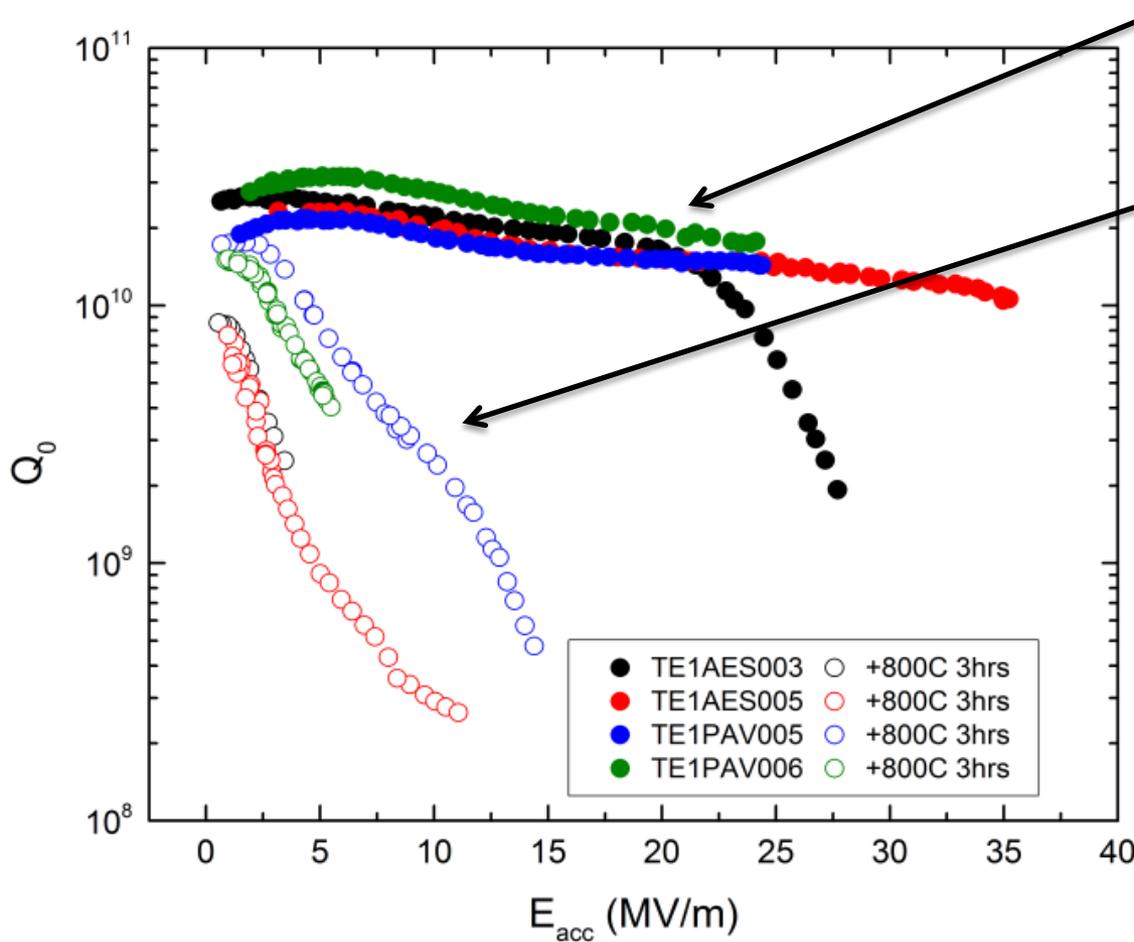
Heat treatment on Large grain material - FANL



Small $\sim 30\%$ improvements to Q_0 removing final chemistry on fine grain cavities.

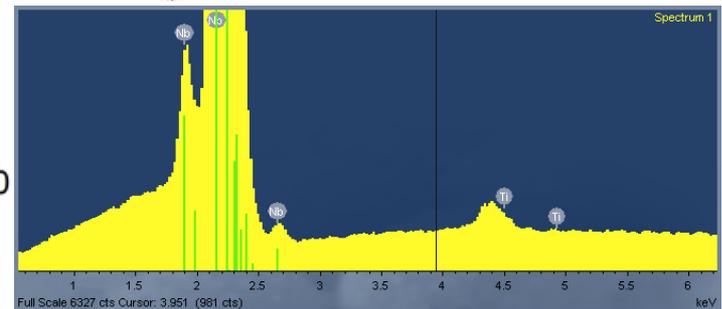
A. Grassellino - <http://arxiv.org/ftp/arxiv/papers/1305/1305.2182.pdf>

Heat treatment on Fine grain material - FNAL



Before furnace treatment

After furnace treatment
 - Titanium contamination from furnace on grain boundaries (many boundaries on FG cavity)

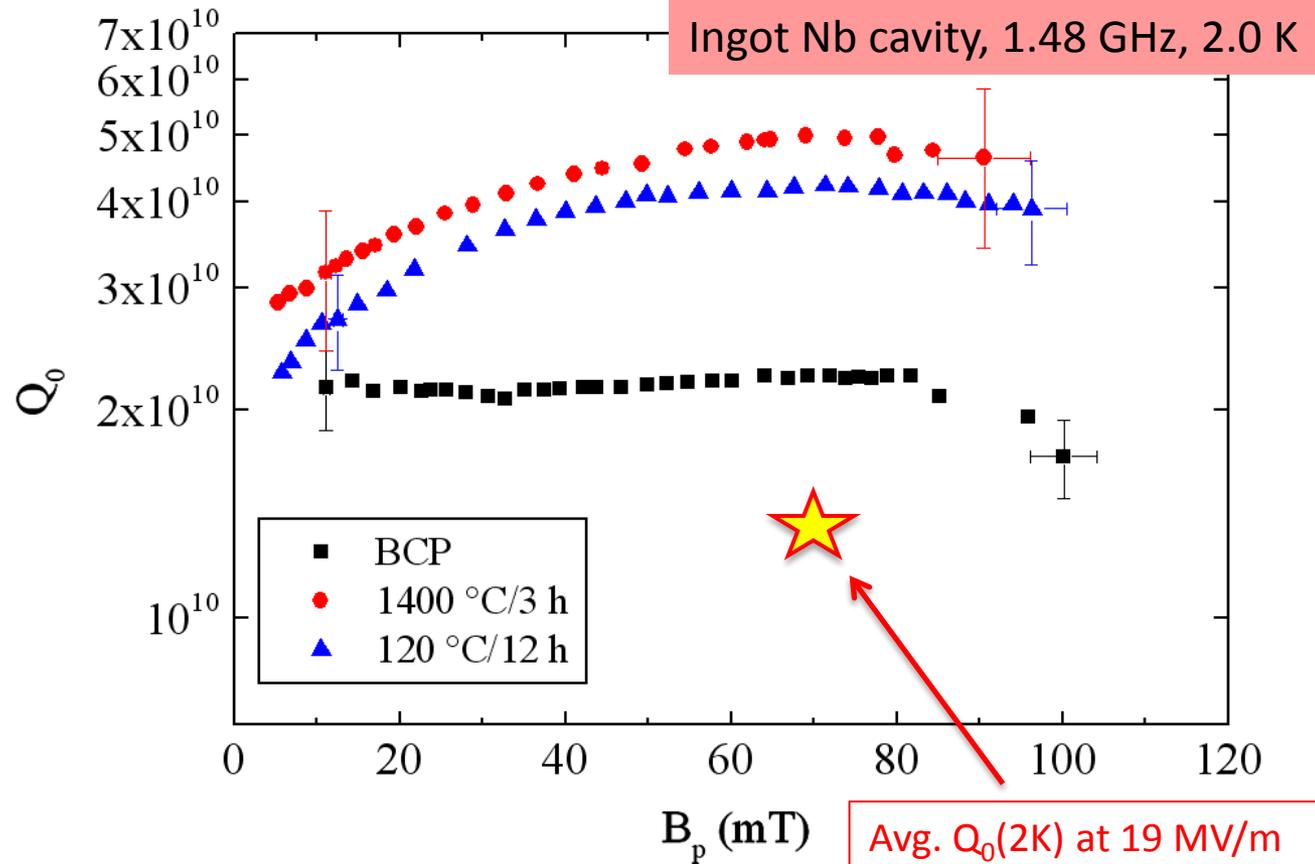
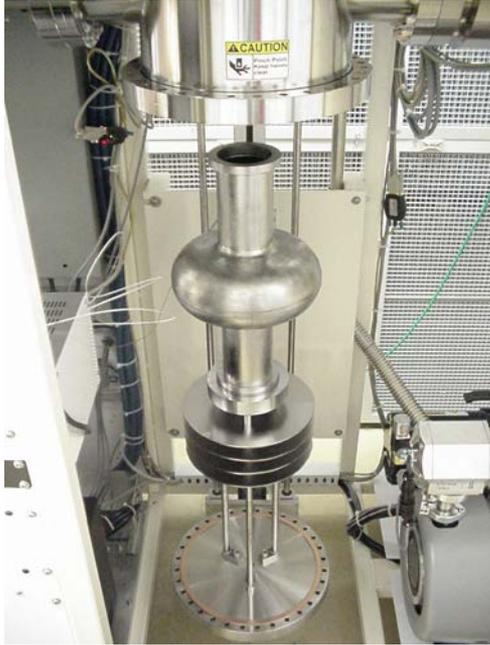


A. Grassellino - <http://arxiv.org/ftp/arxiv/papers/1305/1305.2182.pdf>

High- Q_0 by Ti doping during furnace treatment

- A new induction furnace was designed and installed at JLab to continue the high-temperature annealing study above 800°C in a “clean” environment and without subsequent chemistry.
- In 2012, heat treatment at 1400°C/3h of an ingot Nb cavity with NbTi flanges at JLab resulted in doping of the surface with Ti (~1 at./%, ~1 mm deep) producing an unprecedented high $Q_0 \cong 4.5 \times 10^{10}$ at 2 K, 90 mT

High- Q_0 by Ti doping



P. Dhakal, Rev. Sci. Inst. **83**, 065105 (2012)

P. Dhakal et al., Phys. Rev. ST Accel. Beams **16**, 042001 (2013)

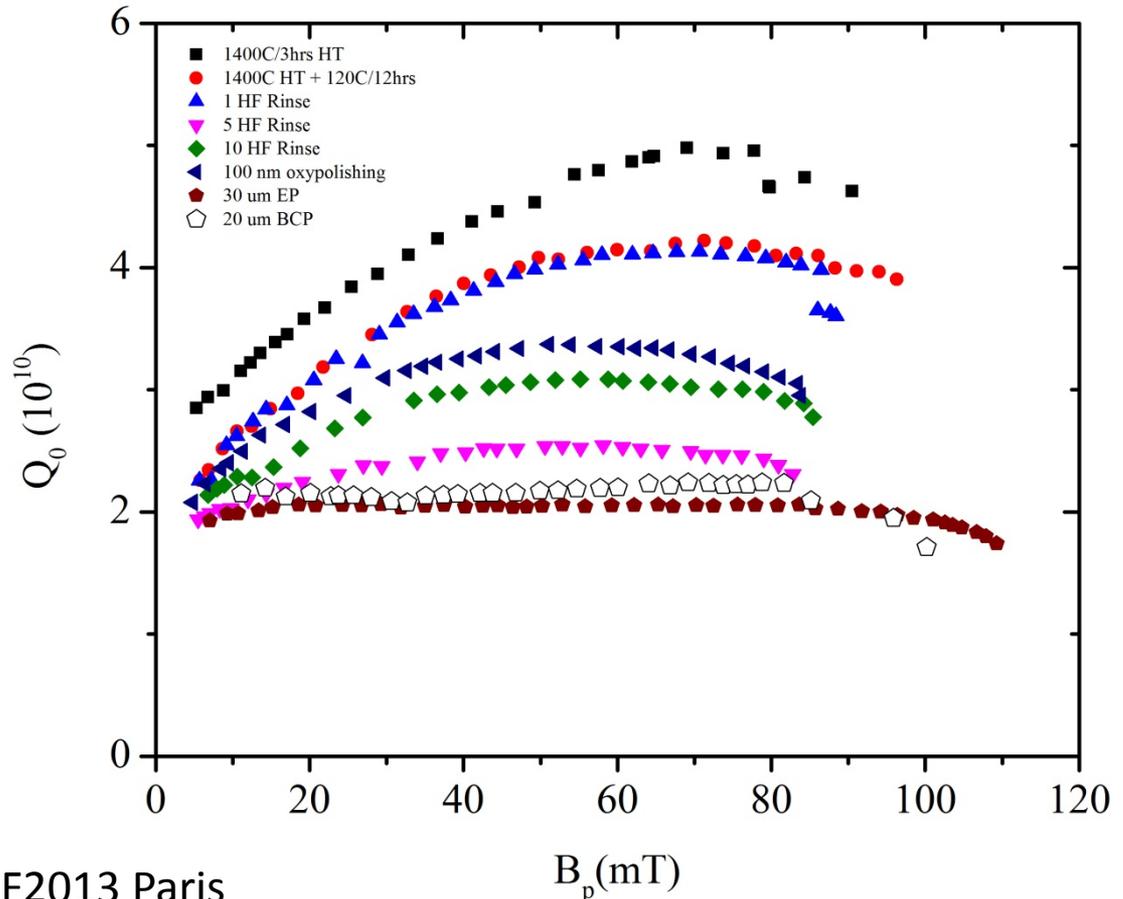
P. Dhakal et al., IPAC'14, p. 2651

Ciovati

Ti-doping and nano-removal

Multiple nano-removal, oxypolishing and EP was done

- No performance degradation while keeping in cabinet for a year
- Extended Q-rise present even after the removal of ~ 120 nm inner surface
- EP after $30 \mu\text{m}$ reproduce the baseline performance
- Sims measurements

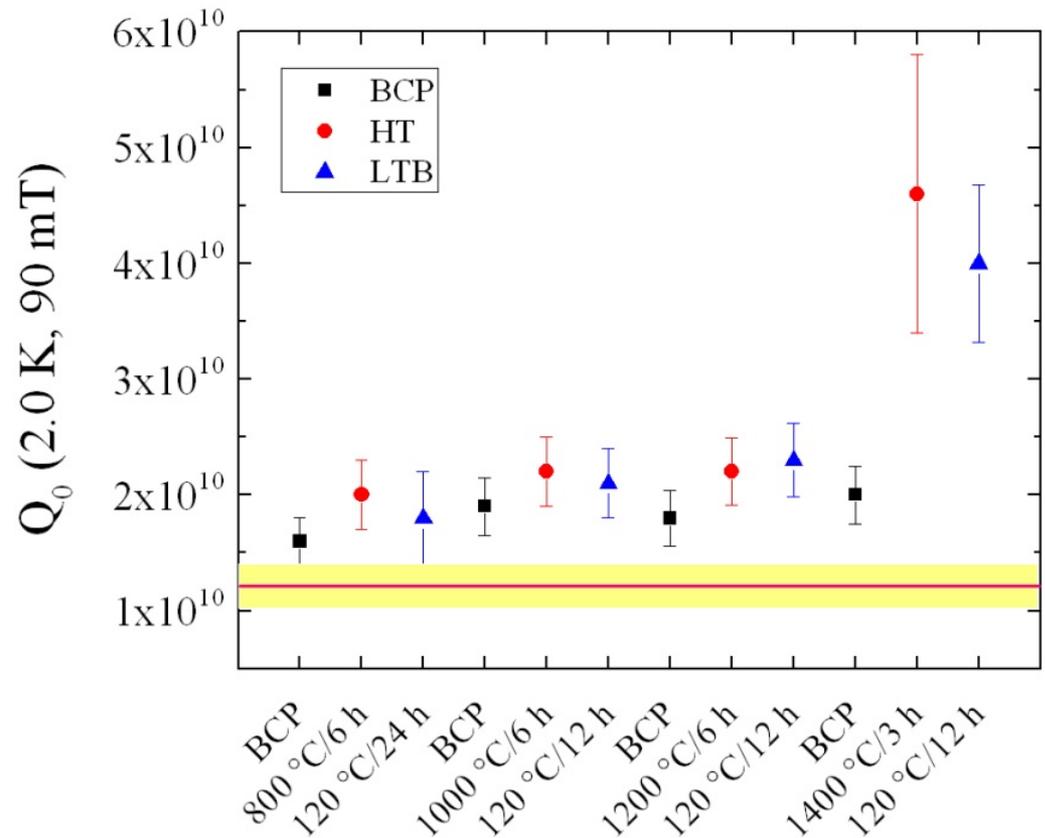


TUIOC04_talk pashupati SRF2013 Paris

HT extended up to 1400°C with new furnace

- Ingot Nb cavity from **CBMM (RRR~200, Ta~1375 wt.ppm)**, treatment sequence after fabrication: CBP, BCP, HT, HPR

Samples' analysis after 1400°C show:
Reduced H content and ~1 at.% Ti content
Higher energy gap and reduced broadening parameter



Phys. Rev. ST Accel. Beams **16**, 042001 (2013)



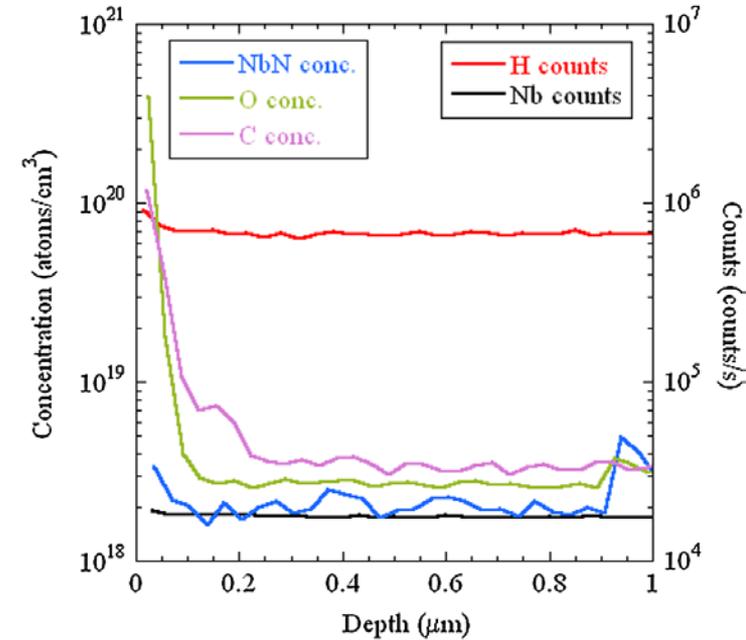
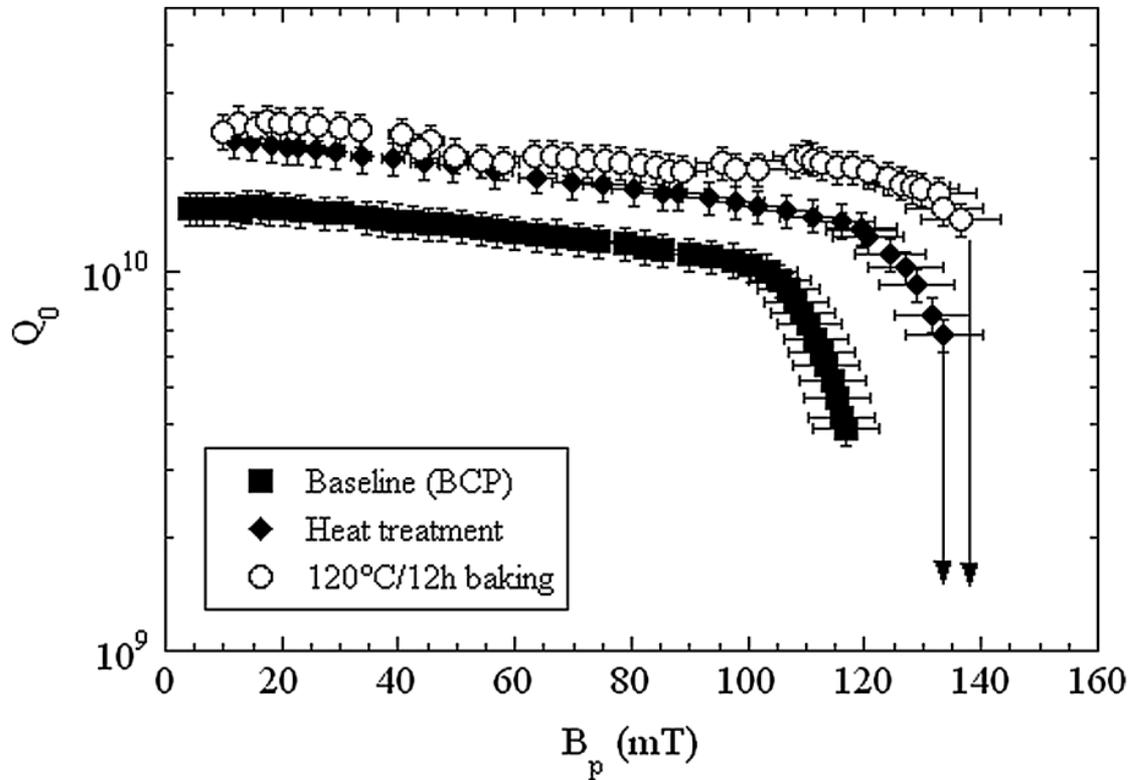
September 23-27, 2013



Nitrogen doping during furnace treatment

- 2009 JLab attempted to make a Hydrogen blocking niobium nitride layer on the surface of a cavity (purposed in the 1970's), with no post heat treatment chemistry. Limited to $1e^{-4}$ torr because of interlock so higher pressures never used. ~30% gain in Q_0 (not doping)
- 2013 an attempt was made to create niobium nitride ($Tc=NbN$) on the surface of the SRF cavity with nitrogen @ ~20mtorr and 800C. The experiment failed at FNAL. $Q = 1e7$. But after random removal choice cavity showed new Q-rise not seem before (except with Ti doping the year before)

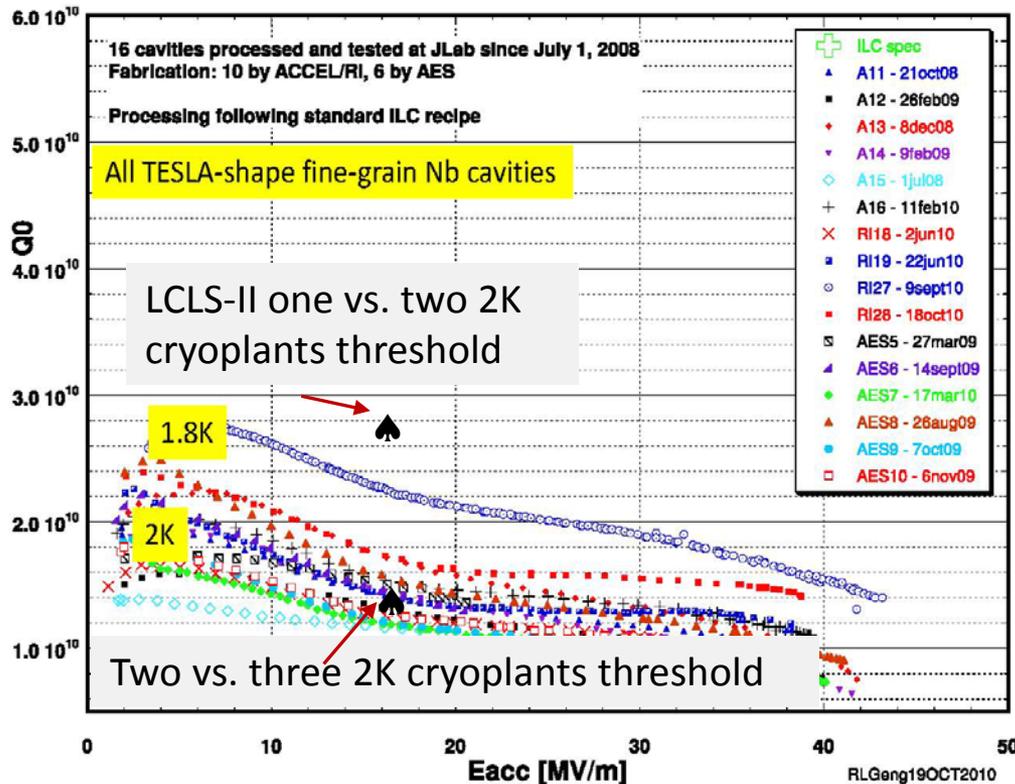
Niobium nitride study JLab



Clean furnace, so Q_0 gain was the same as no doping, study canceled because pressures could not go high enough
Because of safety interlocks on furnace!

G. Coivati et al., PRST - ACCELERATORS AND BEAMS 13, 022002 (2010)

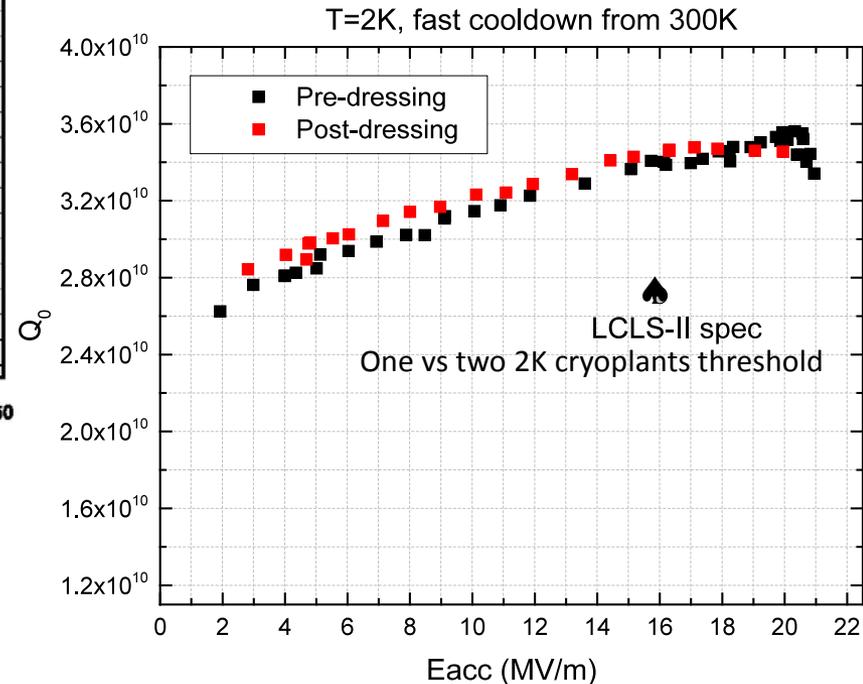
XFEL/ILC recipe vs. N doping



“The best cavities of 2010” (120C bake)
 → Could be marginal for 2 cryoplants
 (likely to require 3 with slow cooling)

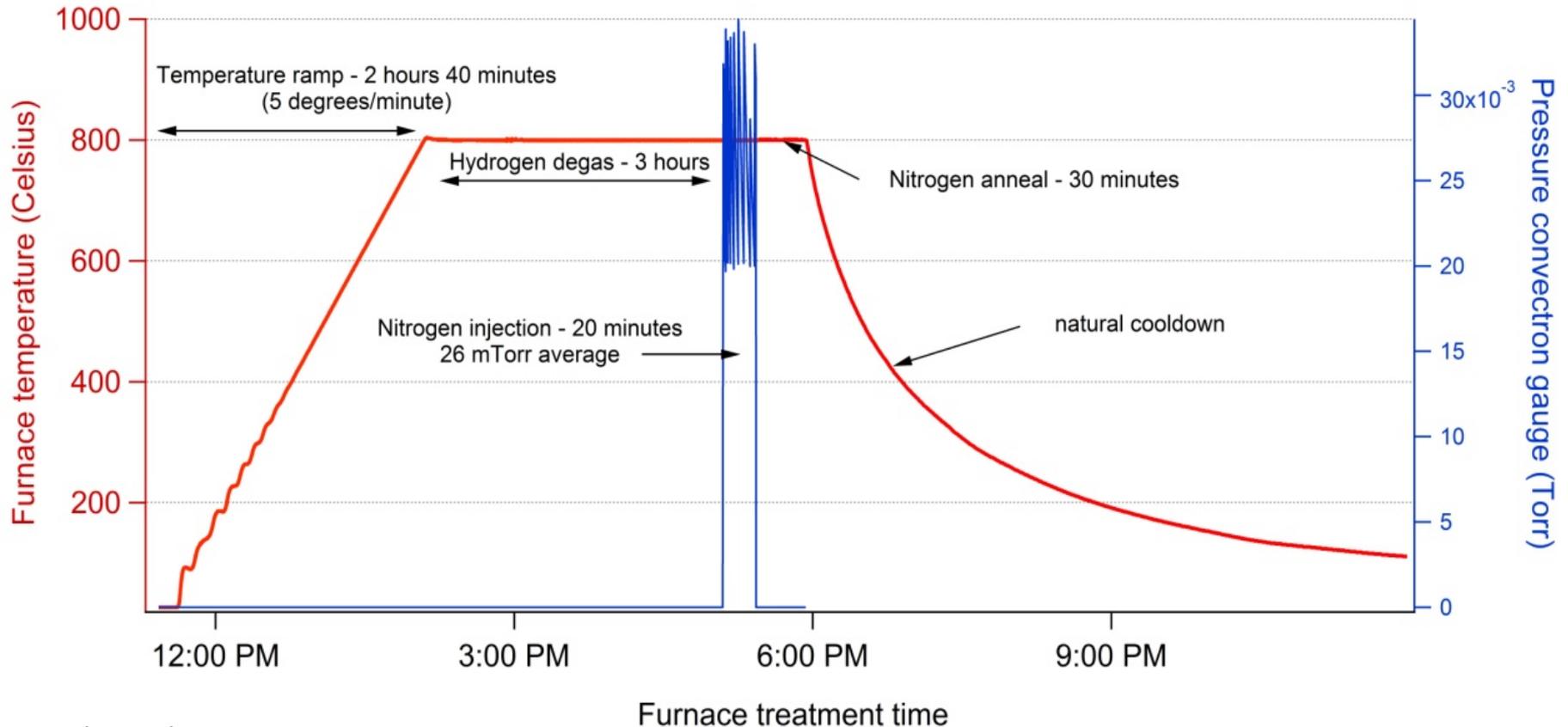
The SRF world is changing

“The best dressed cavity of 2014”
 (N doped)
 TB9AES011



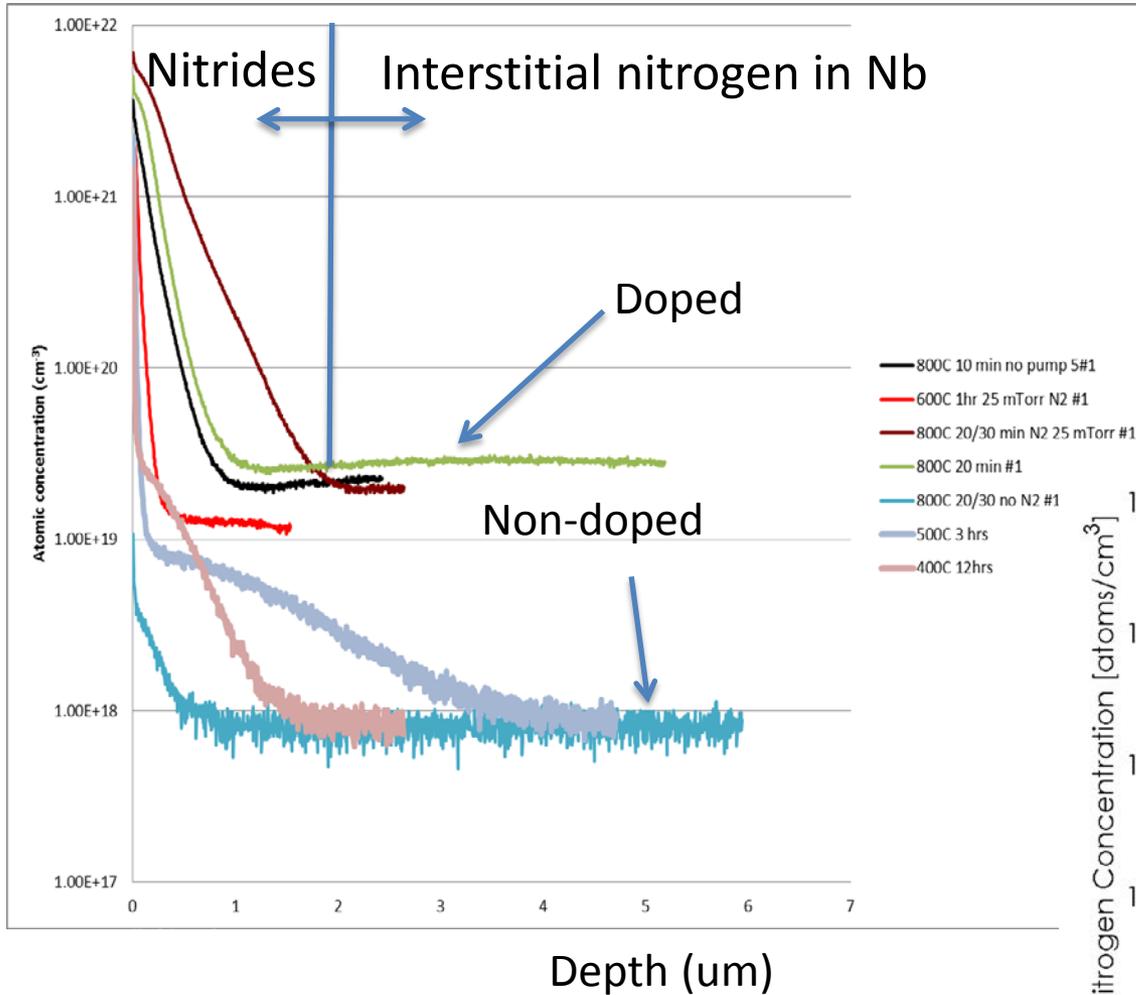
Nitrogen Doping Process

LCLS-II example furnace doping - JLAB

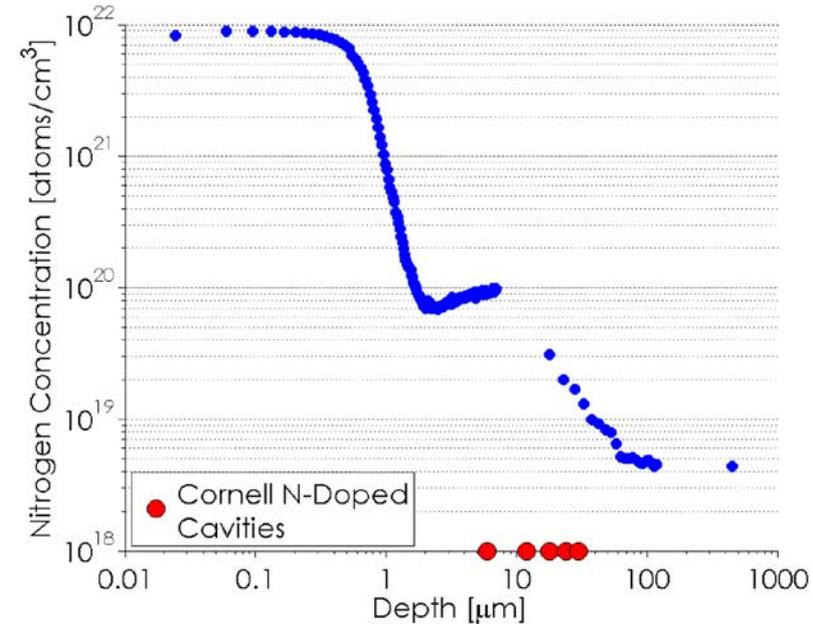


Palczewski

What does N treatment do? N depth profiles by SIMS



See A. Romanenko, talk at LINAC 2014, Geneva
 And D. Gonnella et al, LINAC 2014, Geneva



Amount of nitrogen absorbed

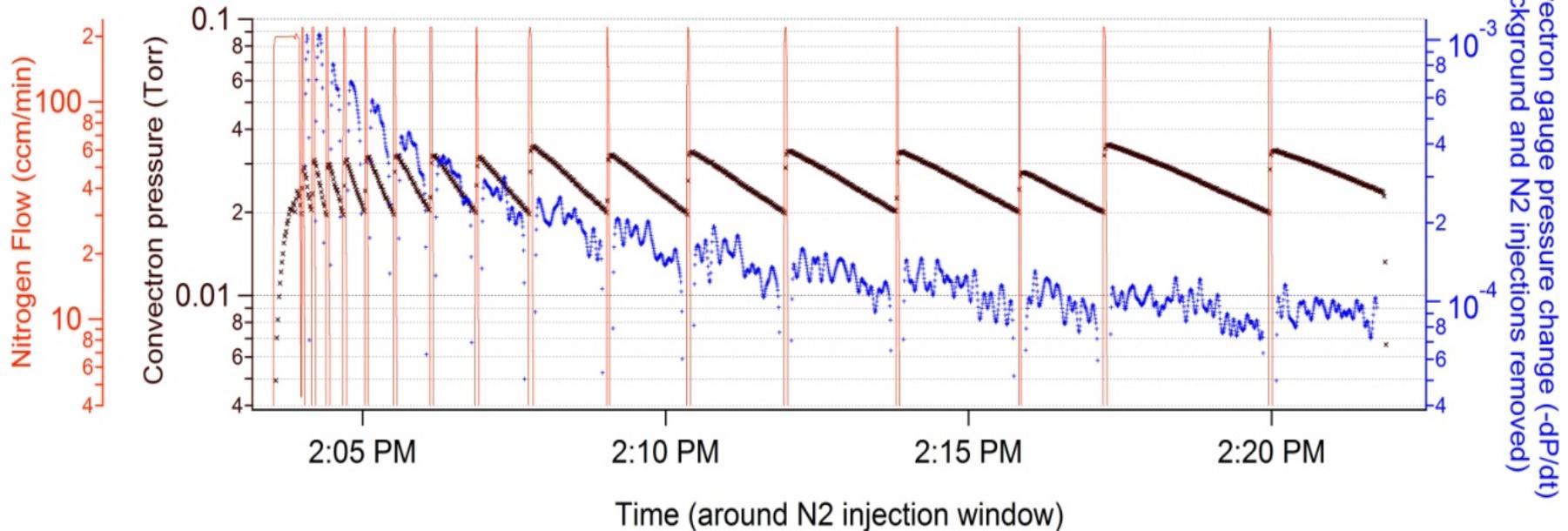
800C 180_N20_A30
AES031

Average pressure over 20 minutes (26mtorr)

Total gas absorbed using pressure drop (540 Torr-liters) - 2500L@800C

Total gas absorbed using mass flow controller total flow (155 Torr-liters) - 25C calibrated source

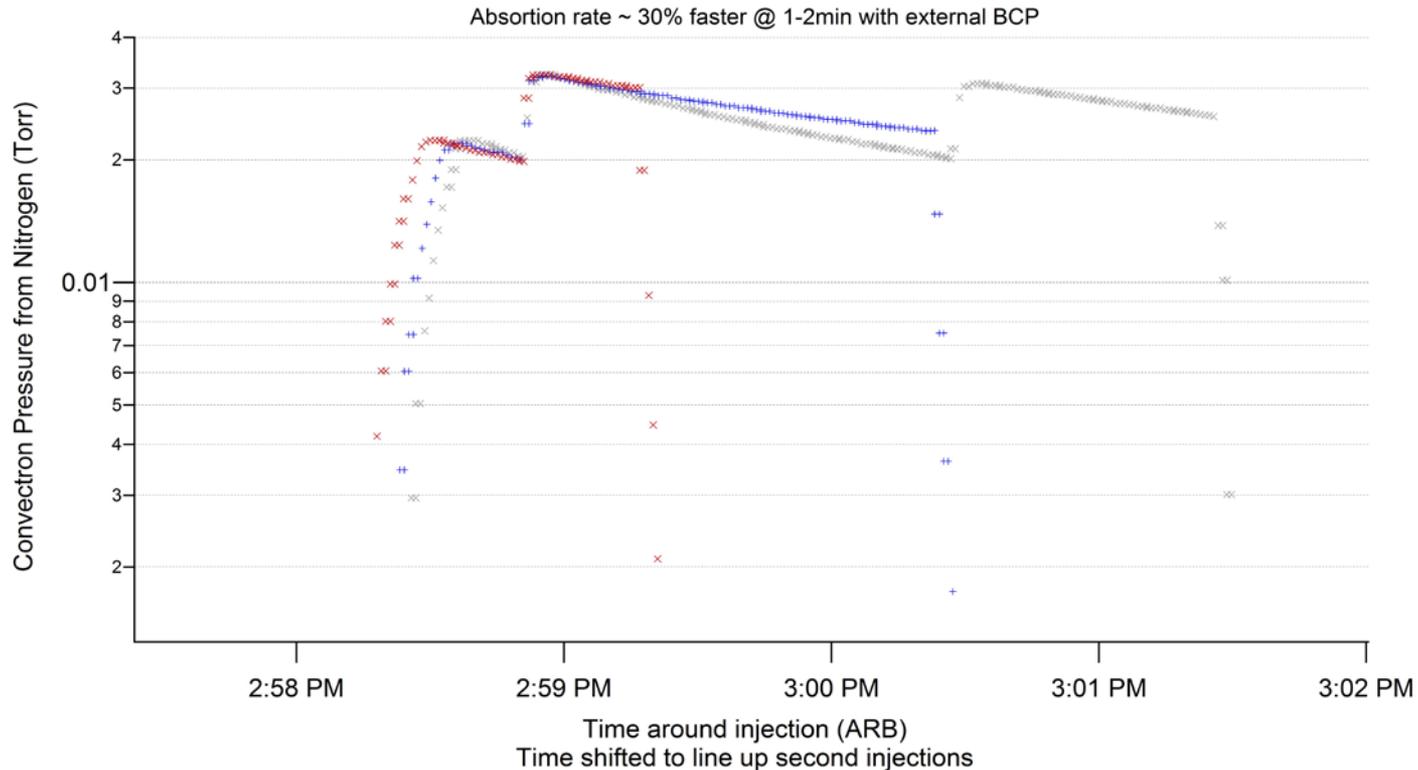
Total gas absorbed using mass flow controller total flow (570 Torr-liters) - converted to 800C from ideal gas law



Re-doped external BCP - % nitrides

Short injection - Long anneal single cell absorption compare

- × RDT-13 800C_A180_N1@27mtorr_A60 - 4.85 to 6.25 Torr liters @ standard atm (25C)
- + RDT-14 800C_A180_N2@26.5mtorr_A60 - 6.9 to 8.3 Torr liters @ standard atm (25C)
- × RDT-15 800C_A180_N3@26mtorr_A60 (external BCP) - 13.2 to 14.5 Torr liters @ standard atm (25C)



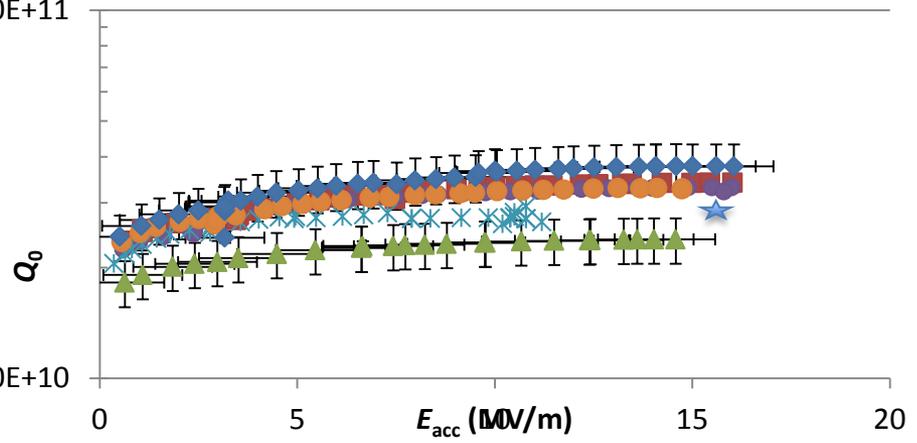
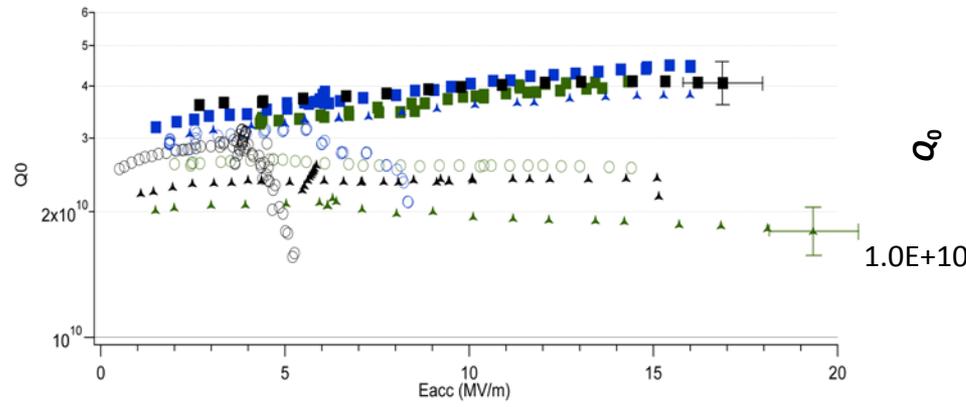
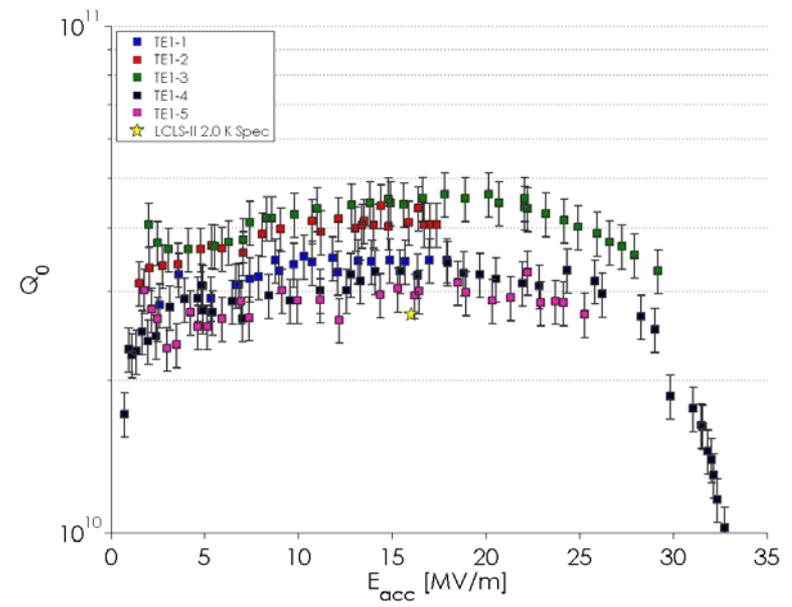
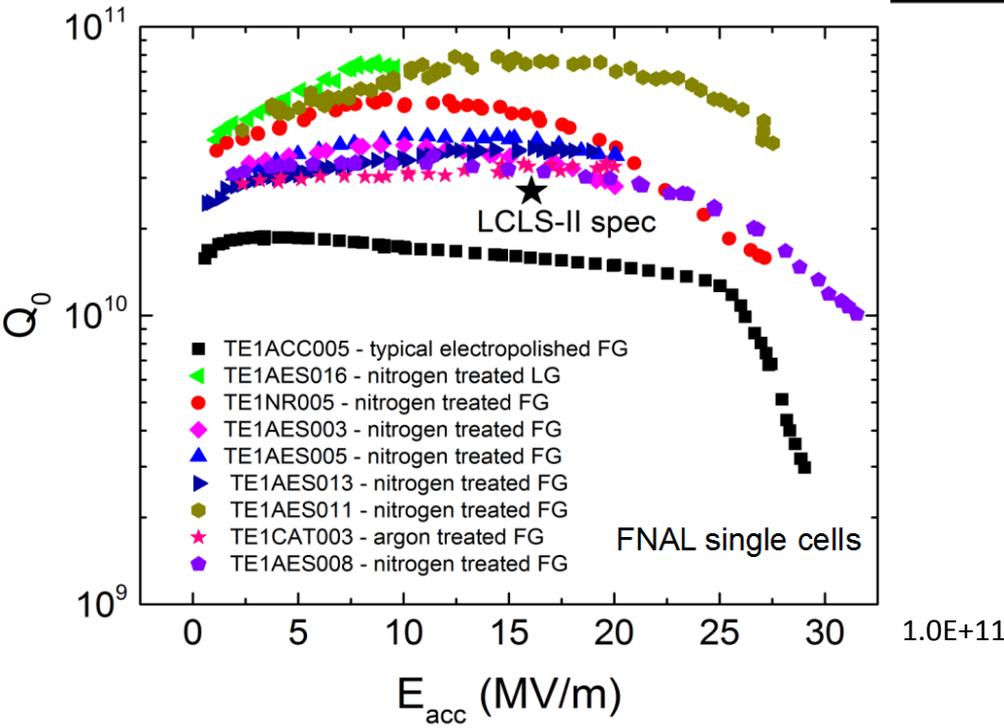
Minimum 60 to 70% of nitrogen goes into nitrides on surface, the rest goes into the cavity

Current doping/EP recipes tried that “worked” – rising Q0

LAB	pressure	Time N	Anneal time	EP microns
JLAB	~26mtorr	1	40,60	5,10,15
JLAB	~40mtorr	2	10,20,30	5,10,15
JLAB	~26mtorr	20	10,30,60	10,15,17,20
JLAB	~26mtorr	2	6	5
Cornell	~40mtorr	20	30	5,12,18,24,30
FNAL	~20mtorr	10	0	10
FNAL	~20mtorr	2	6,20	5-30
FNAL	~20mtorr	20	30	10,20,30
FNAL	~10mtorr		?	?
FNAL	~20mtorr	60	0	10,40,80

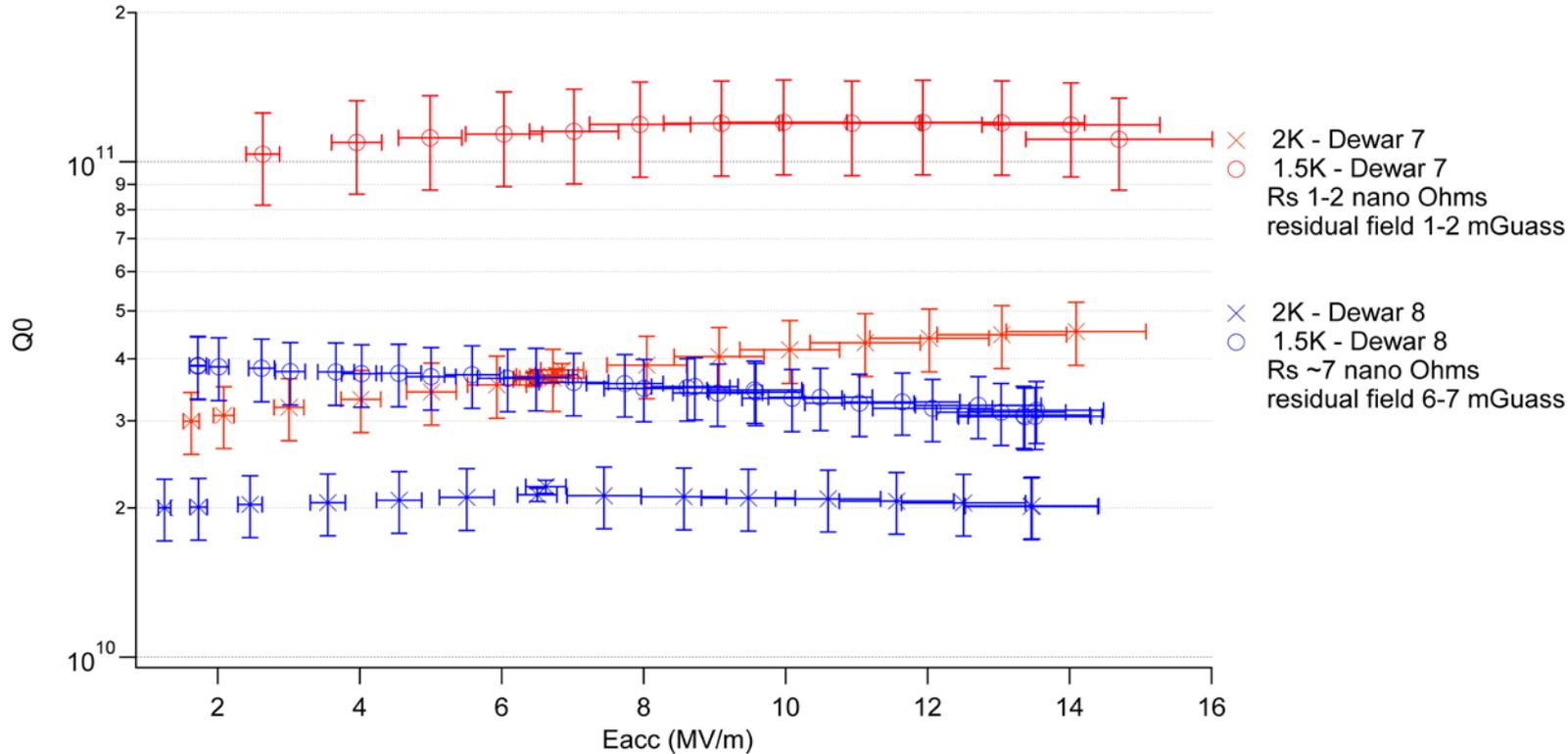
Incomplete list – ones I have verified with my notes, I know there are others especially at FNAL (Sorry)

Multiple cavity tests – N doping



Problem with N-doping- environment

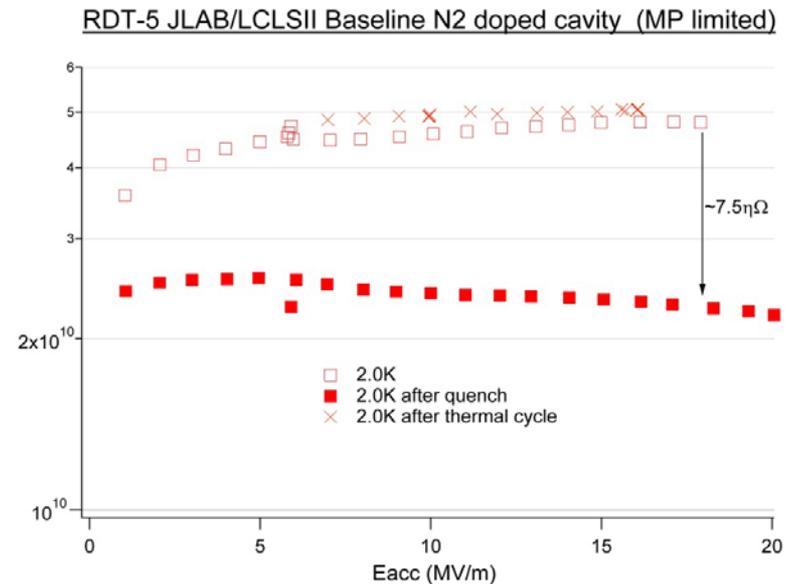
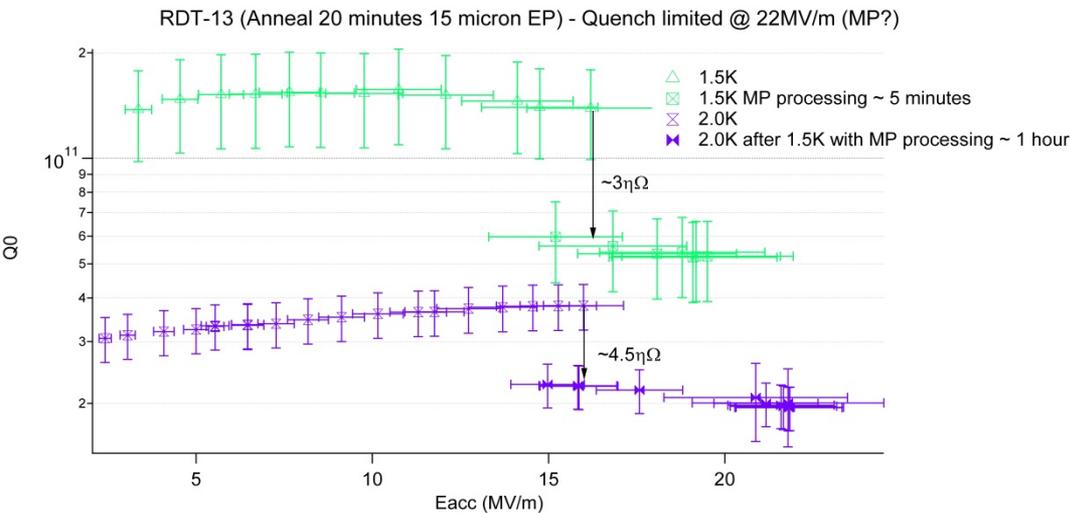
RDT-15 800C 3h_N20_A_50 + 15micron EP - quench @ 15MV/m



Cavities are highly supposable to environmental factors, where the remnant magnetic field can dramatically change Q performance, for standard cooldowns

Problem with N-doping- temporary quench degradation

Amount of flux captured during a quench and the resulting drop in Q_0 is quite variable.



Example of multipacting-induced quenching Q_0 degradation and recovery with thermal cycle

Residual Resistance vs Trapped Flux

- N-doped cavities appear to be more sensitive to trapped flux.
- Higher R_{res} for same flux
- Due to higher R_{NC} from lower mfp?

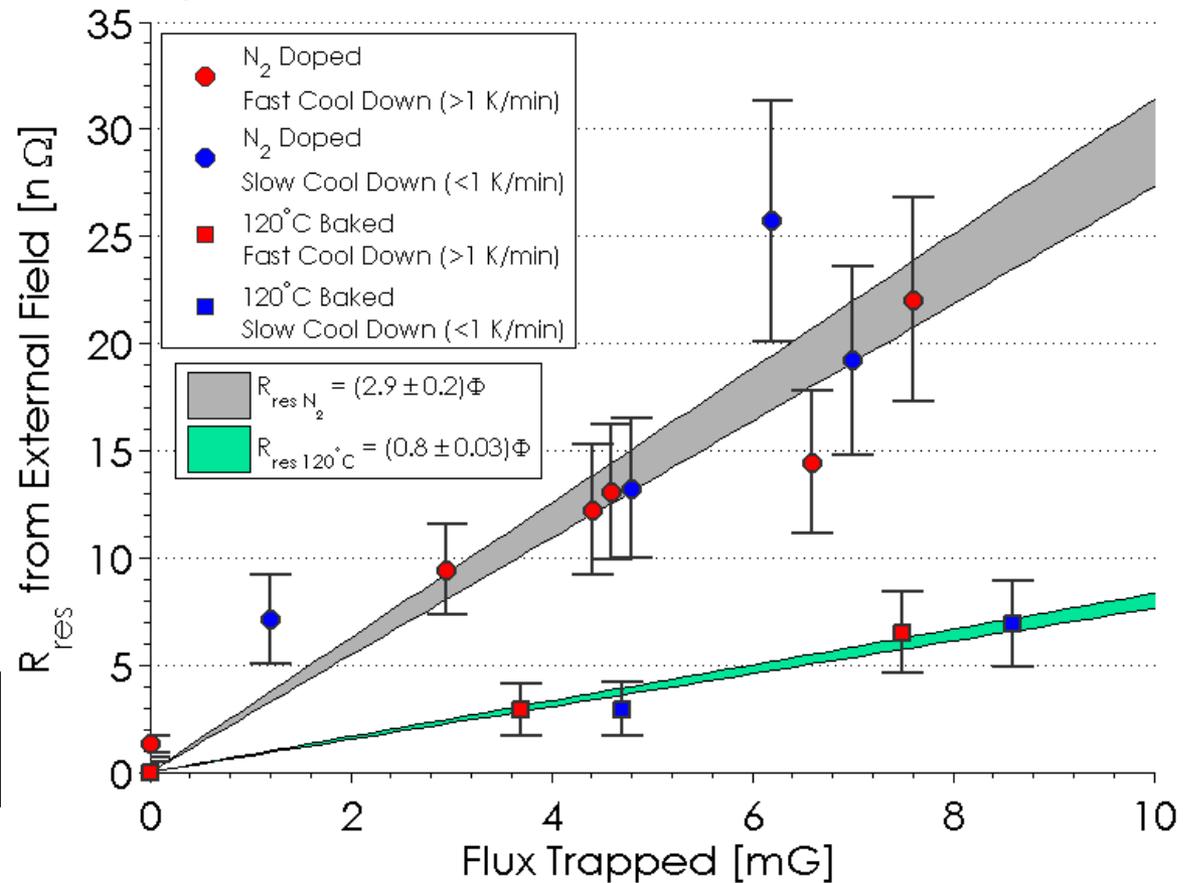
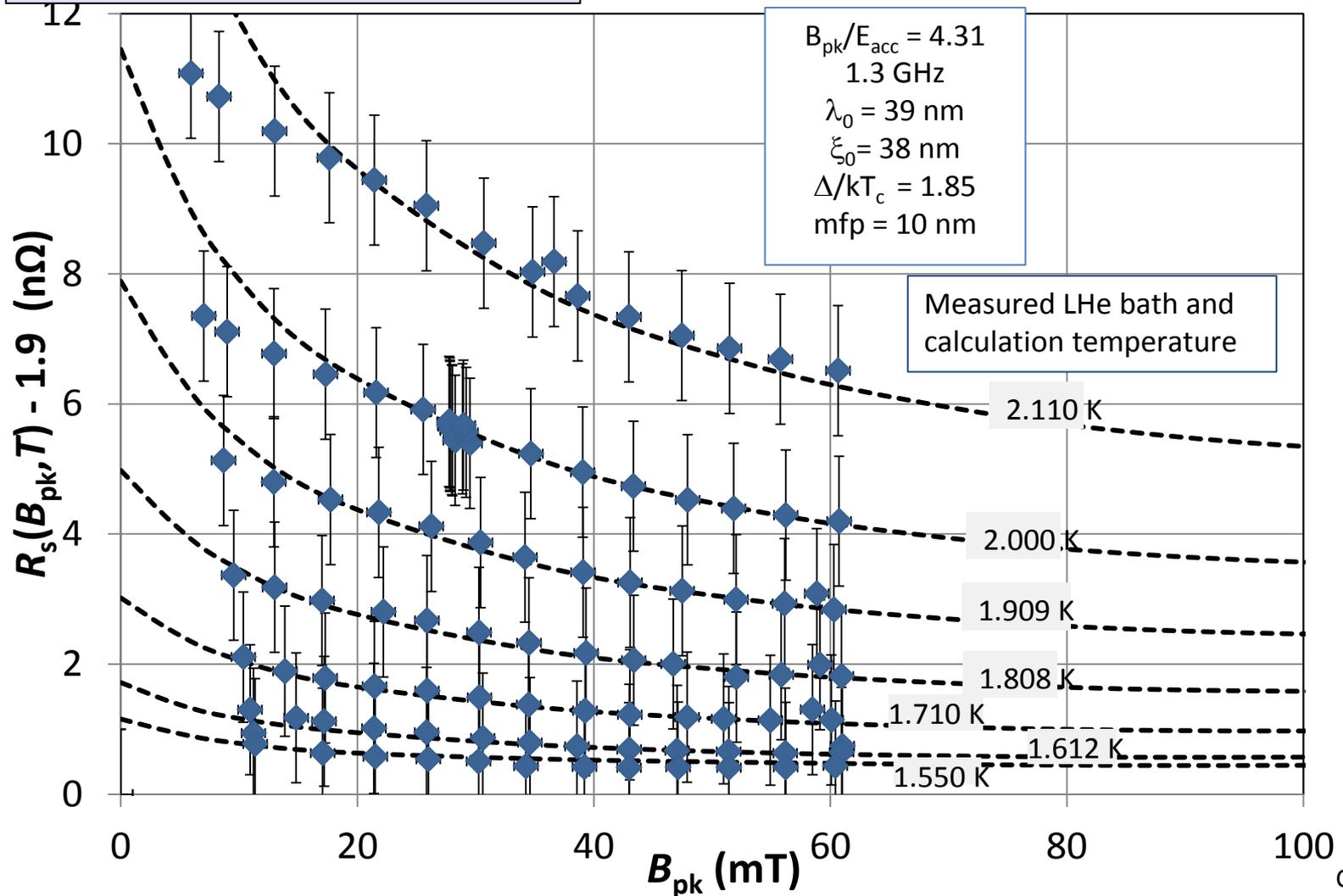


Image Courtesy of Dan Gonnella
Cornell University
See IPAC13 WEPRI063

The Best Doped Cavities Match the new R_s Theory

RDT-15 180/20N/50 + 15 micron EP

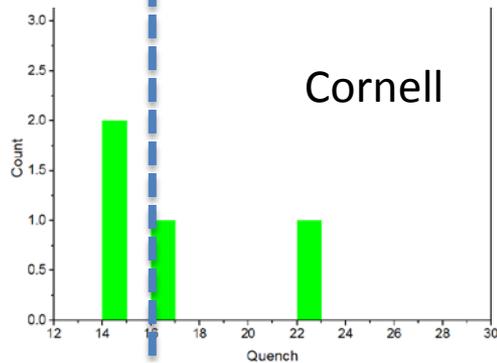
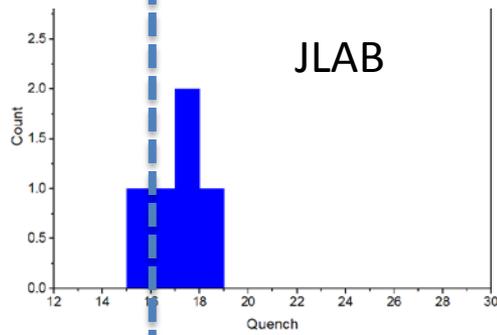
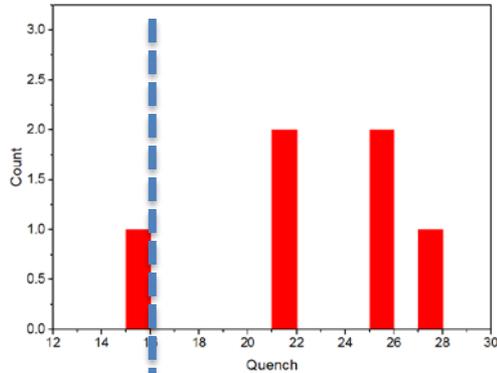
Not yet sure how to interpret this agreement



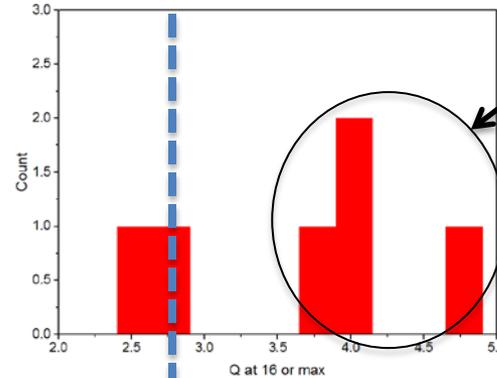
C. Reece

9 cell studies LCLS-II baseline Q0 and quench field – Nitrogen doping

16 MV/m

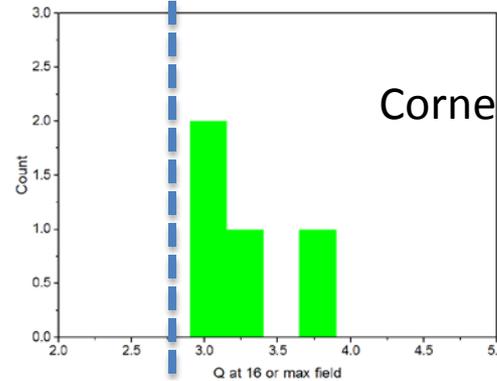
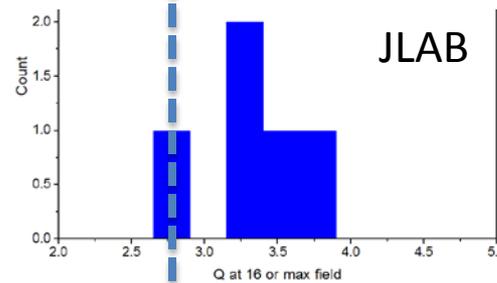


2.7e10



NbTi flanges

FNAL N2 A6 EP 5



Quench field definitely dependant on doping, where lower doping is better! From single cell higher doping appears to produce better Q0

N20 A30 EP~15-25



Nine cell frozen recipes results

	Gas bake details	Average Q	Average quench field	First pass yield	Second pass yield
FNAL "recipe 1" N=6	<ul style="list-style-type: none"> 800C 3 hours in HV 2 min at 800C with N ~ 20 mTorr 6 min at 800C in HV 	3.7e10	~23 MV/m (2 nd pass) ~21 MV/m (1 st pass)	67% @18 MV/m	83% @18 MV/m
Jlab/Cornel I "recipe 2" N=10	<ul style="list-style-type: none"> 800C 3 hours in HV 20 min at 800C with N ~ 40 mTorr 30 min at 800C in HV 	3.5e10 (Jlab) 3e10 (Cornell)	~16.6 MV/m (Jlab) 17 MV/m (Cornell)	60% @16 MV/m 20% @18 MV/m	

Anna Grassellino, LCLS-II DOE Status Review, June 30th, 2014

Question?