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Introduction

- Radioactivity is prevalent in the Earth's natural environment due to the presence of cosmogenic and primordial radionuclides. It is commonly found in various ecological, geological and environmental formations such as , rocks, soils, sand, plants, water, and air.
- The release of alpha- emitting natural radionuclides like ^{238}U , ^{232}Th and ^{222}Rn during production, manufacturing and discharge processes has steadily increased. Monitoring alpha radioactivity in the air, solid and liquid waste from these industries is crucial.
 - Thermal Power Plants
 - Fertilizer Plants
 - Granite Industries

Solid State Nuclear Track Detection

- Solid-state nuclear track detectors (SSNTDs) are materials, like certain types of crystals or polymers, that record the tracks of charged particles resulting from nuclear interactions. These detectors are often used in radiation dosimetry and nuclear physics experiments to study ionizing radiation.
- Simplicity, High Sensitivity & Cost effectiveness are inherent features.
- The CR-39 detector stands as the backbone of the SSNTD technique.
- Invaluable for quantifying emissions of extremely low rates of charged particles such as alpha particles.
- Exceptionally user friendly & cost effective.
- Adaptable detector nature No radioactive decay No electronic break down
- Collective nature of stored information Applicable to various geometrical measures including 2π , 4π , For ward & Backward geometry.

Tetraethyl Ammonium Bromide:

As New Chemical Etchant

Adding TEAB to NaOH etchant accelerates track formation with rapid interaction and energy transfer on the detector surface, efficiently revealing track path. The use of TEAB as a phase transfer catalyst enhances the effectiveness and accessibility of complementary raw materials in this modern chemical etchant.

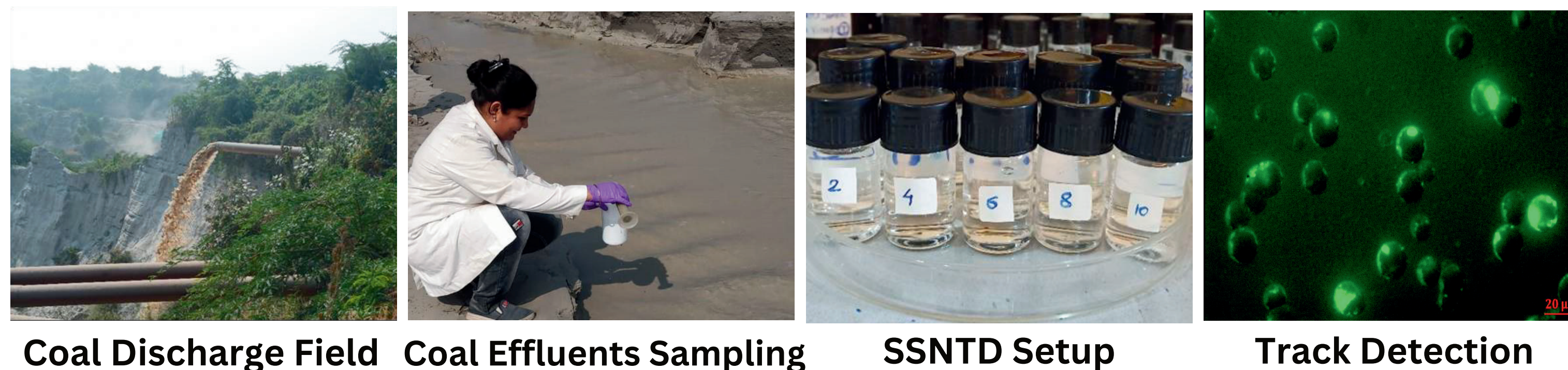
Steps involved in Solid State Nuclear Track Detection

Registration of Latent tracks	Enlarging Tracks (Etching)	Counting of Tracks
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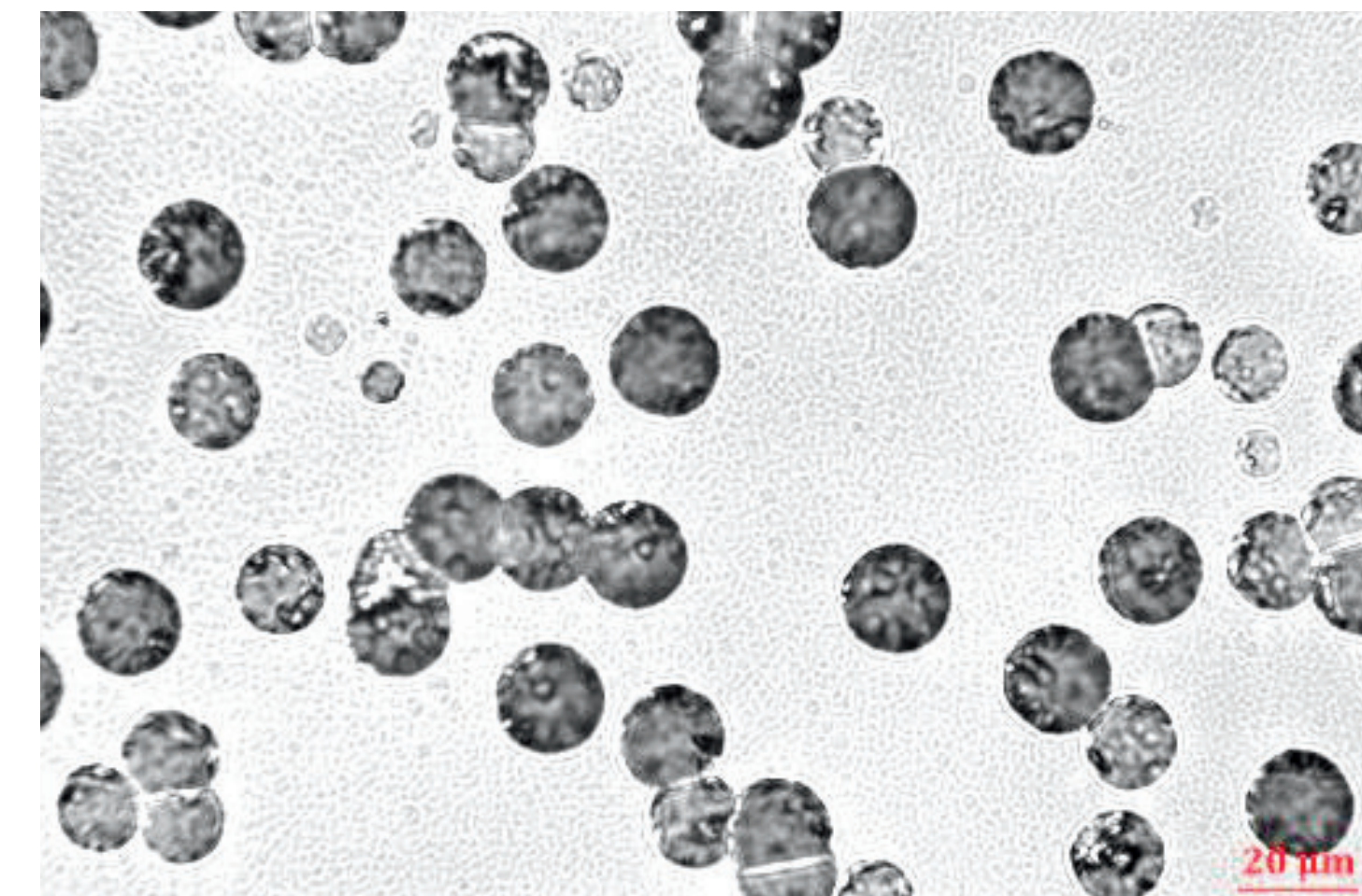
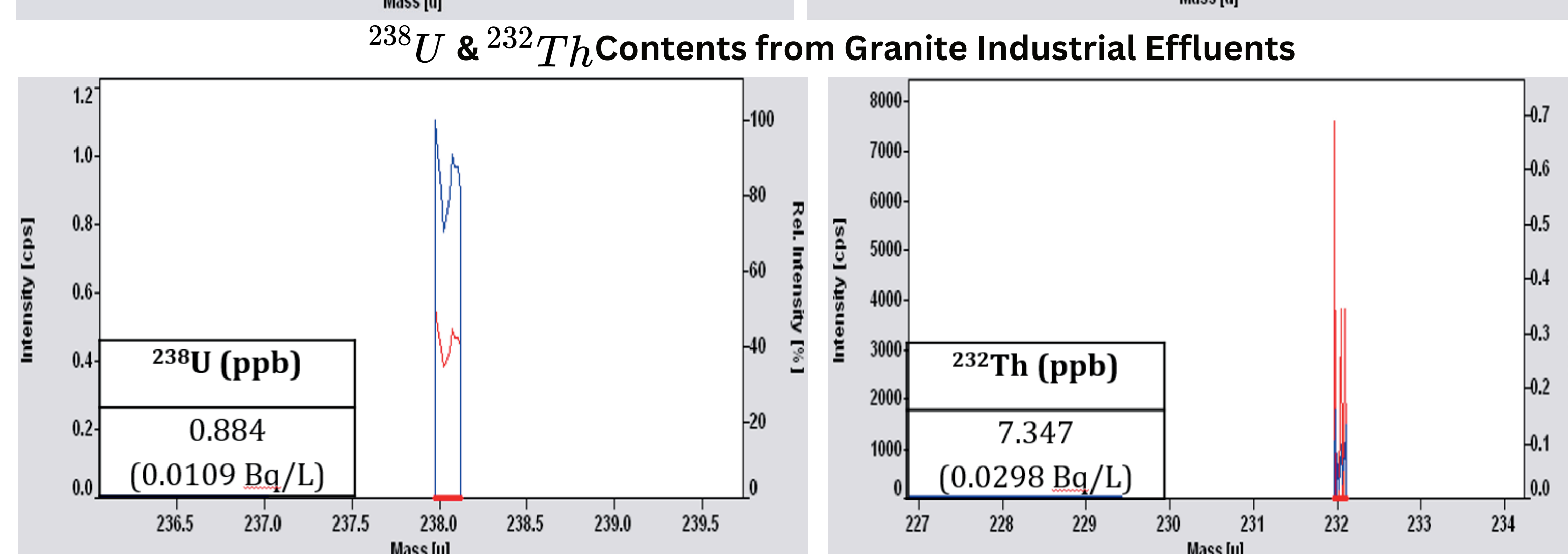
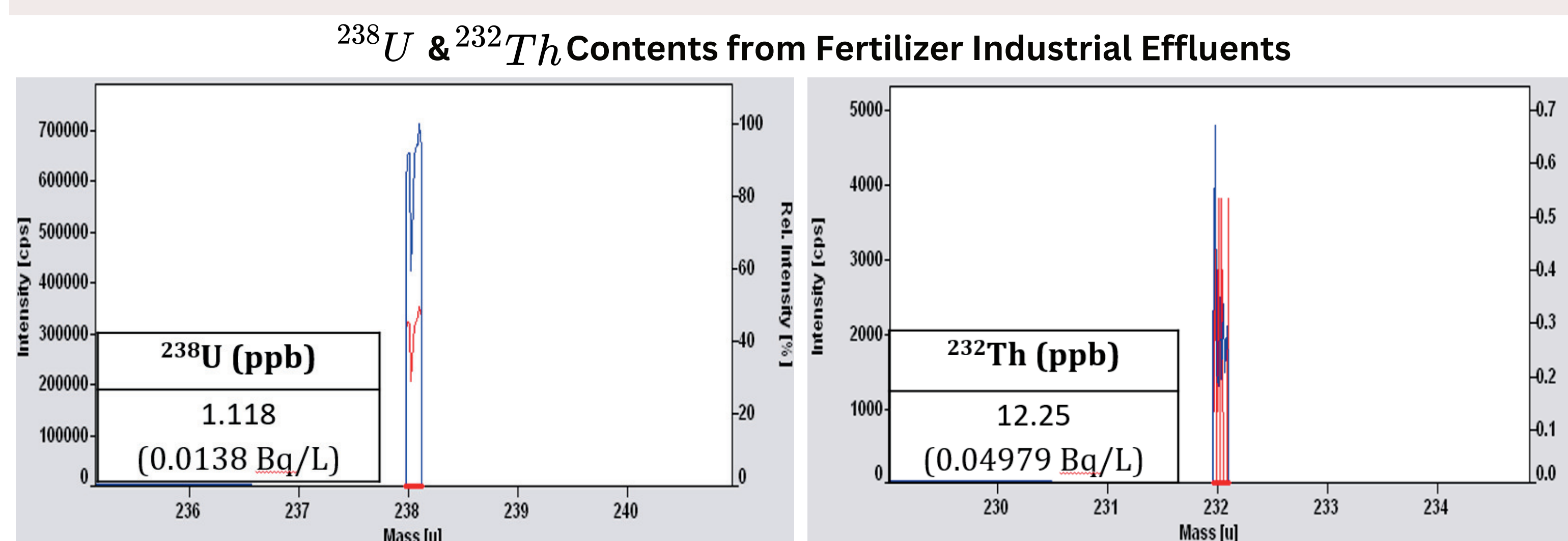
Experimental

- Standardizing CR-39
- Employing $[\text{Th}(\text{NO}_3)_2 \cdot 5\text{H}_2\text{O}]$ as the RS
 - Subjecting CR-39 (0.5 mm, Size: $1\text{X}1\text{cm}^2$) to Effluents (10mL) exposure.
 - Preparing Etchant with 5% w/w Tetra Ethyl Ammonium Bromide with 6M NaOH
 - Etching CR-39.
 - Measuring Track Density using formula: $T_d = \frac{\text{Total counts}}{\text{Total area}}$
 - Track Diameter: Magvision Software
 - Track Registration Efficiency: $K_{wet} = \frac{T_d}{n\lambda t}$
 - Analyzing Track Profile using Techniques Spinning Disc Confocal Microscope & Atomic Force Microscope, ESEM, LM
 - Conducting ICP-MS & EDXRF Analysis for comprehensive assessment.

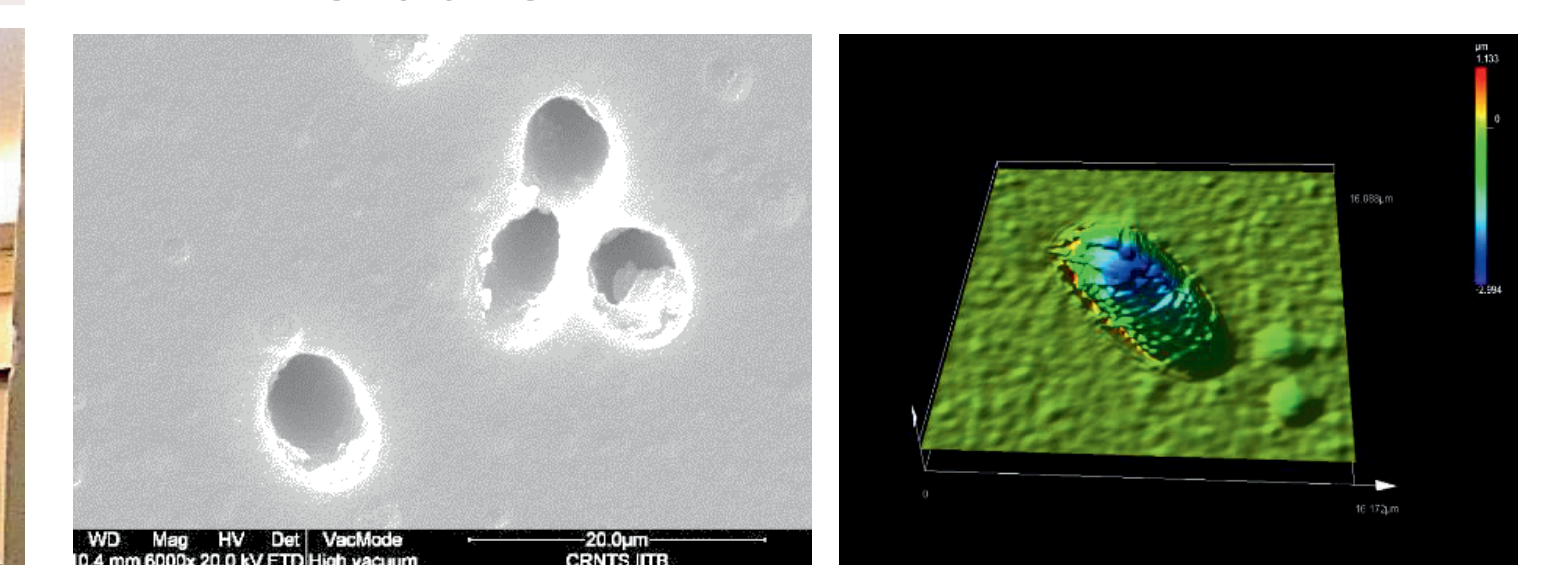
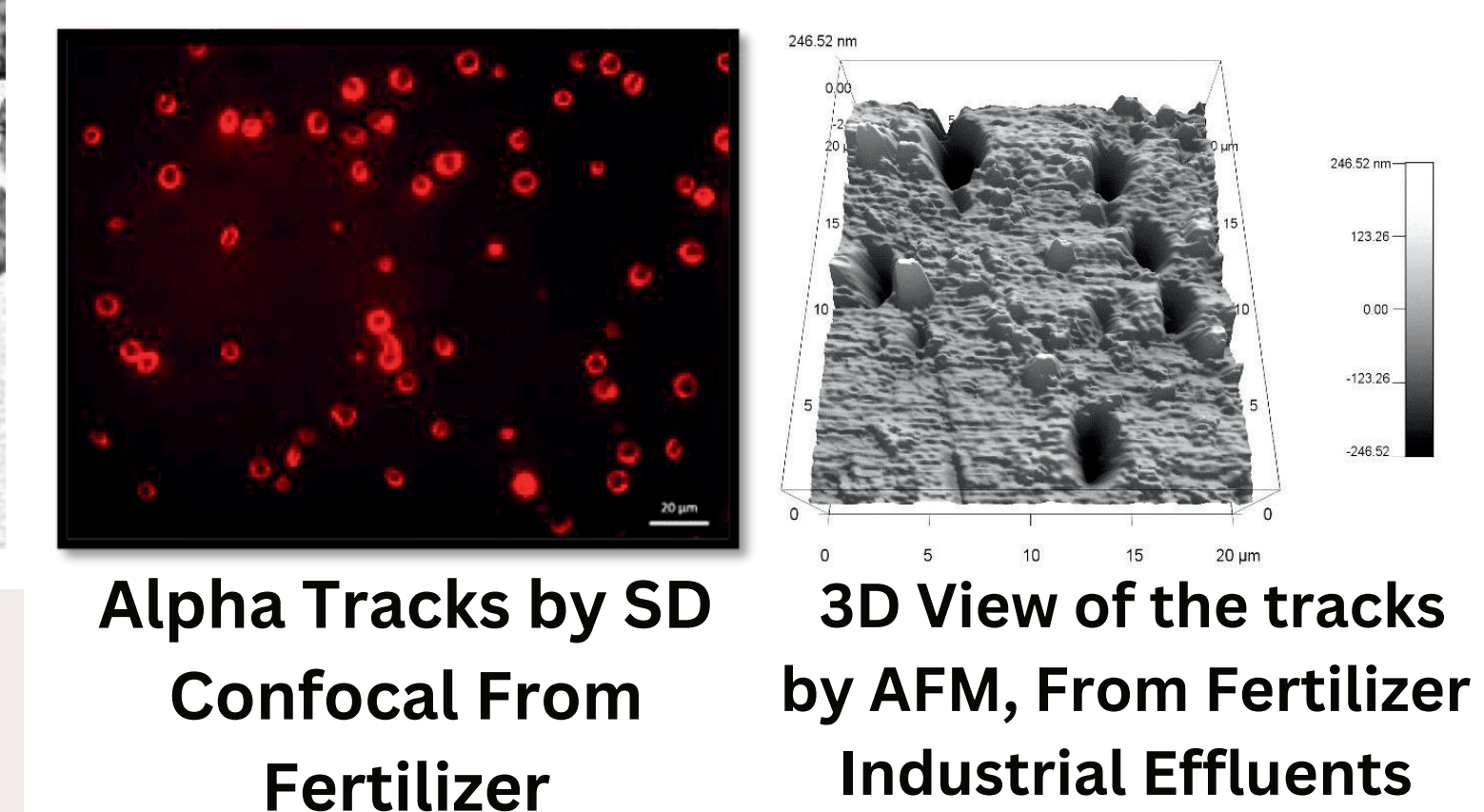
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| 1. Concentration:(600 ug/L) | 4. Temperature: (60° C) | 7. Thickness: (0.6mm) |
| 2. Etching Time : (6H) | 5. Normality: (6N) | 8. Detector: (CR-39) |
| 3. Exposure time: (24H) | 6. Amount of TEAB Added to NaOH : (5%) | |



Radionuclide Contents of Effluent Samples By ICP-MS



Results



ESEM Track profile from PF Effluent on CR-39 Using 5% TEAB + 6 NNaOH

Laser Microscopy: Track profile on CR-39 Using 5% TEAB + 6 N NaOH

- The CR-39 detector exhibits distinctive dark black circular tracks, varying in proportions and depth. These tracks signify low-level alpha radioactivity.

Conclusions

- The use of TEAB proves highly effective as an etchant, facilitating the detection & revelation of alpha tracks on the CR-39 surface from solution medium.
- Etching times of 1.5 & 6H for alpha track detection ^{238}U at 1.118 ppb (0.0138 Bq/L or 0.828 dpm/L) and Bq/L or 2.9874 dpm/L). ^{232}Th at 12.25 ppb (0.04979)

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References

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