

Bronze route inspired Nb₃Sn thin film coatings for SRF cavities

Wenura K. Withanage¹, Andre Juliao¹, Shreyas Balachandran¹, John Buttles², Choong-Un Kim³, Peter J. Lee¹, Lance D. Cooley¹

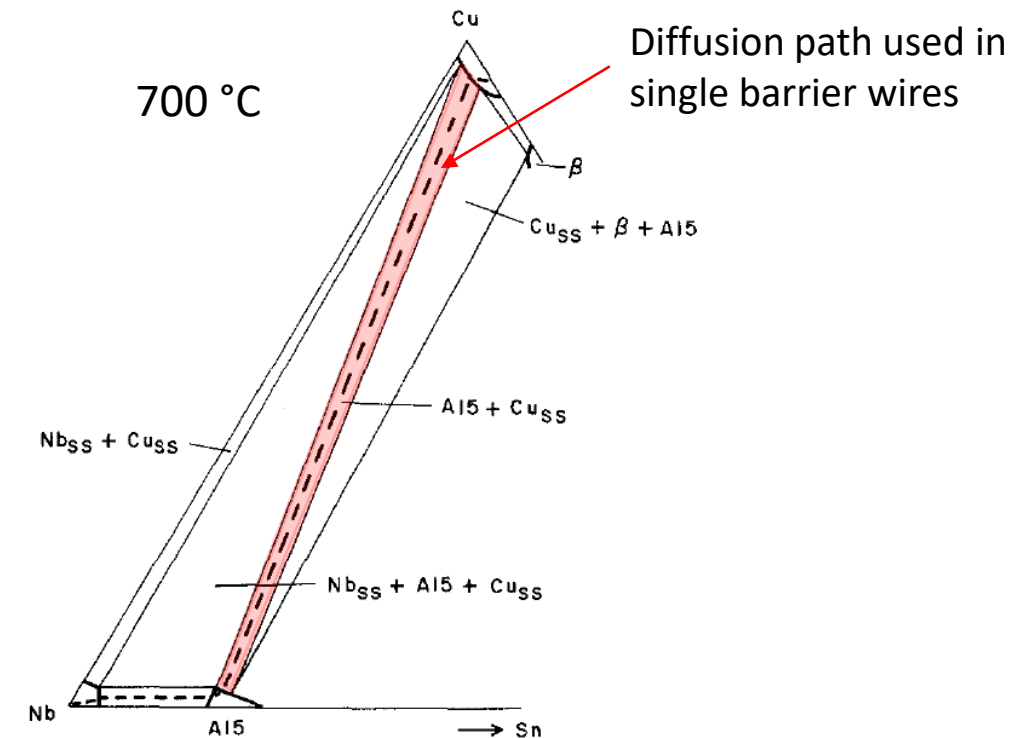
¹ Applied Superconductivity Center, National High Magnetic Field Laboratory,
Florida State University

² Bailey Tool & Manufacturing, Lancaster, TX

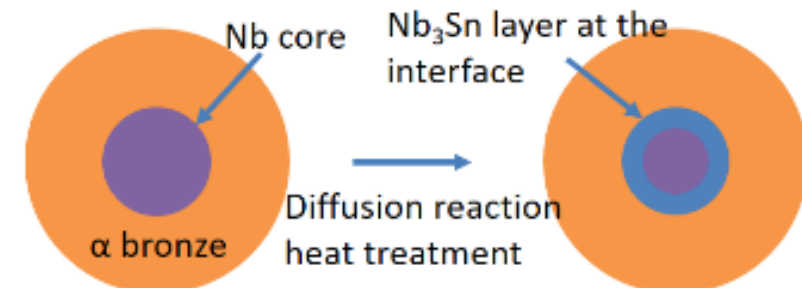
³ Materials Science & Engineering Department, University of Texas – Arlington, Arlington, TX

Bronze route Nb_3Sn growth

- Mature technology in wire fabrication
- Solid state diffusion reaction between Nb and Cu-Sn solid solution
- Low temperature ($\sim 650 - 800^\circ\text{C}$) Nb_3Sn phase purity

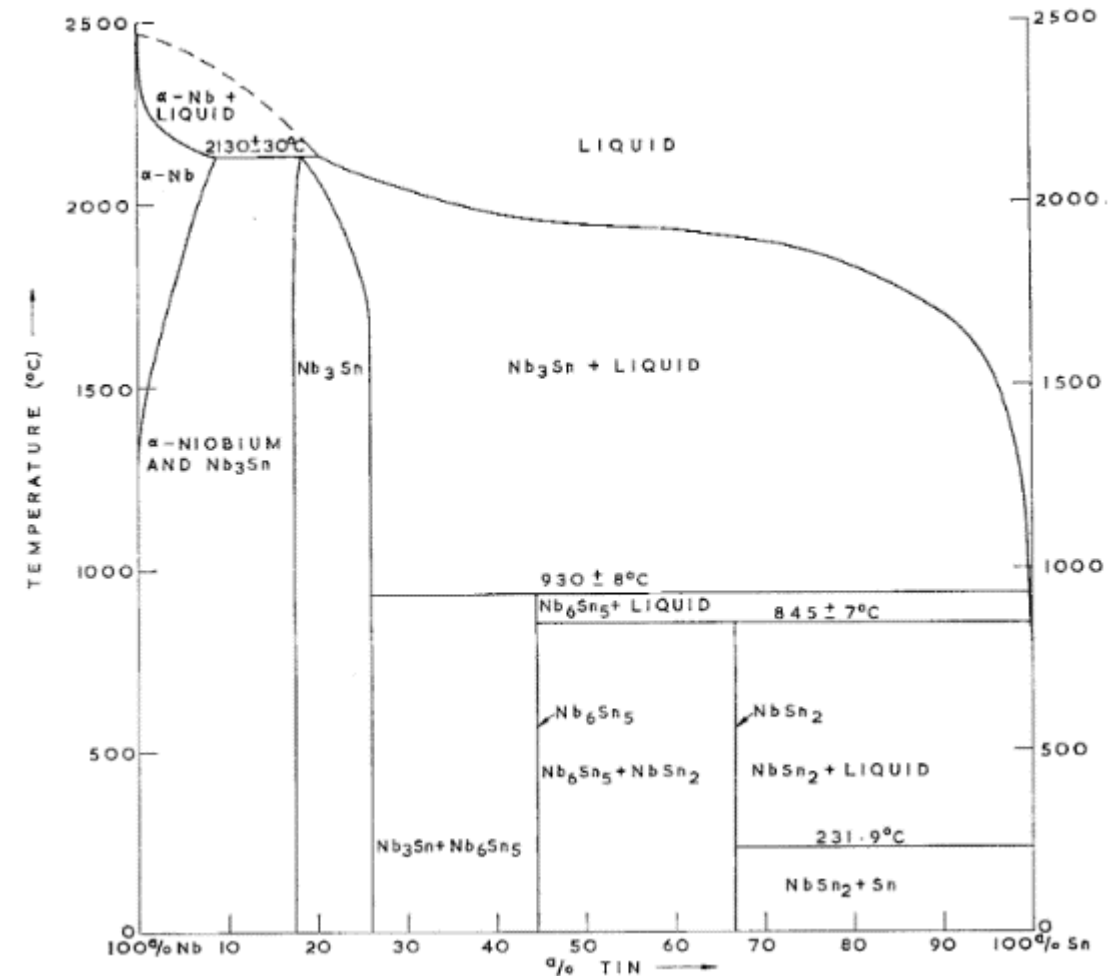


D. Dew-Hughes & T. S. Luhman Journal of Materials Science volume 13, pages 1868-1876 (1978)



Why investigate bronze routes?

- Move reaction window from $\sim 1100^\circ\text{C}$ to $\sim 700^\circ\text{C}$
 - Avoid the Nb_6Sn_5 and NbSn_2 phases by exploiting ternary Cu-Sn-Nb system
- Make the reaction compatible with cavity bodies made from copper; avoid bulk niobium (and)
 - Possible cost and formability advantages
 - High thermal conductivity of Cu
- Avoid Sn vapor, chlorides of Sn and Nb polishing chemistry (corrosive, toxicity).

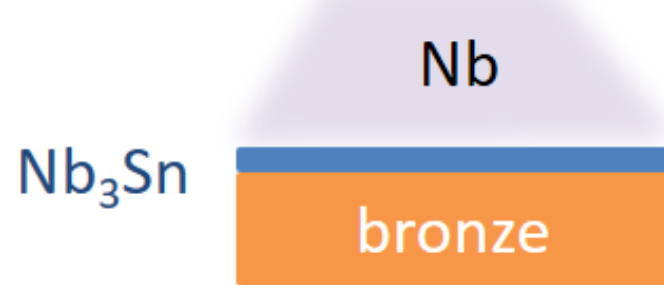


J. P. Charlesworth *et al* J. Mat. Sci.5, 580 (1970)

Nb on 15wt.% Sn bronze

Bronze is cold-rolled and annealed at 525°C to homogenize, then mechanically polished

Hot bronze method

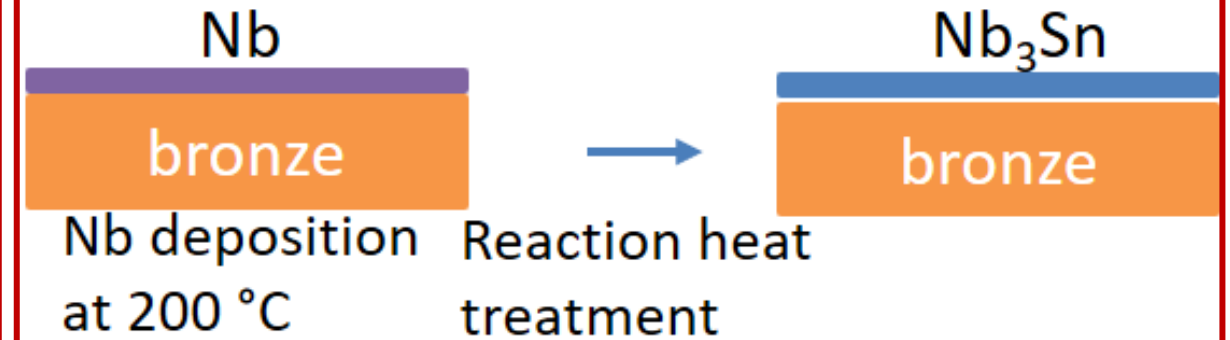


hot bronze: 650 – 715 °C

Route 1: Nb₃Sn forms concurrent with Nb deposition on hot bronze

- This route is different than wires
- Bronze surface become Sn rich
- **Very fast! 700 nm in only 22 minutes**

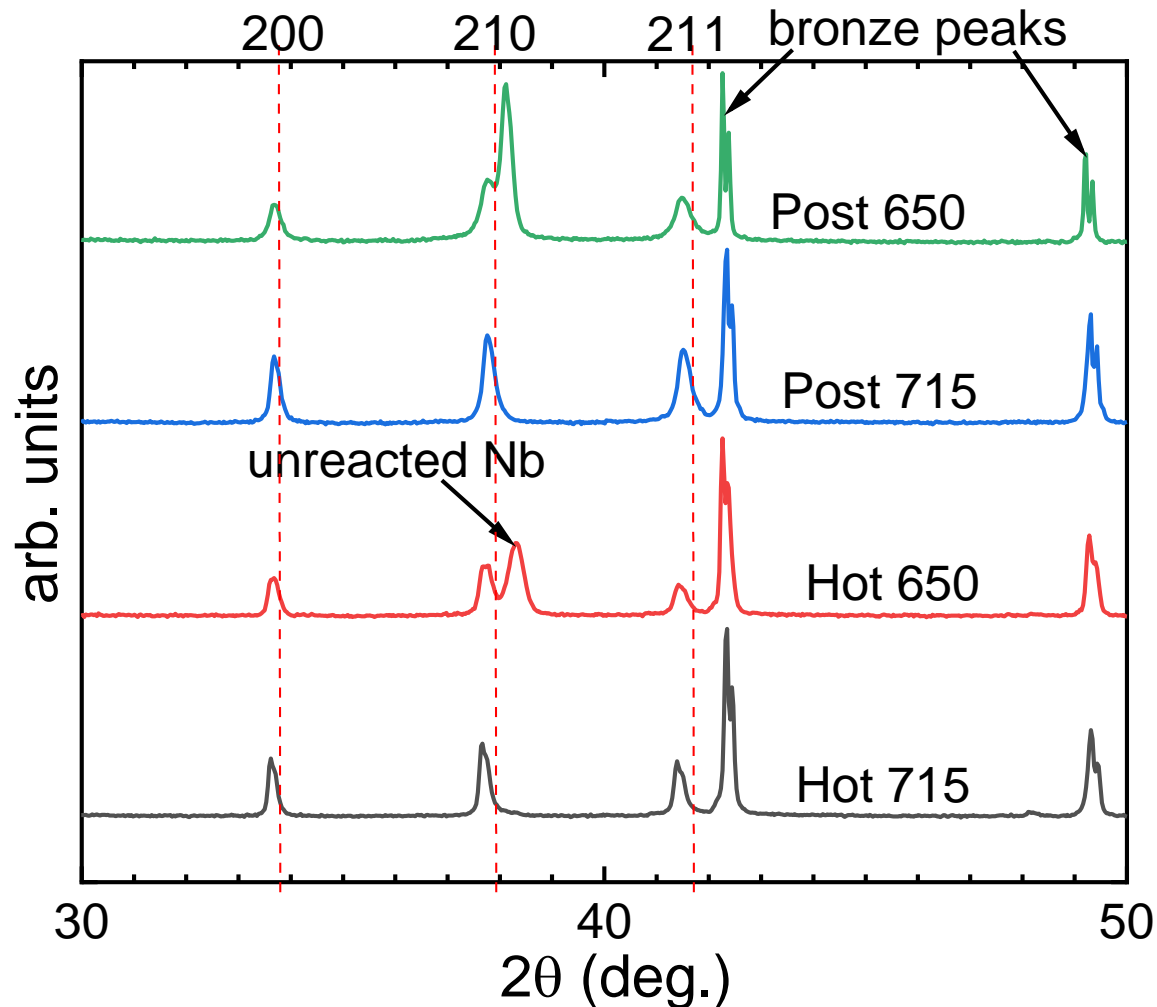
Post reaction method



Route 2: Nb₃Sn forms during a diffusion reaction after deposition

- This route is like wires
- *In-situ* reaction without breaking vacuum
- **700 nm in 360 minutes**

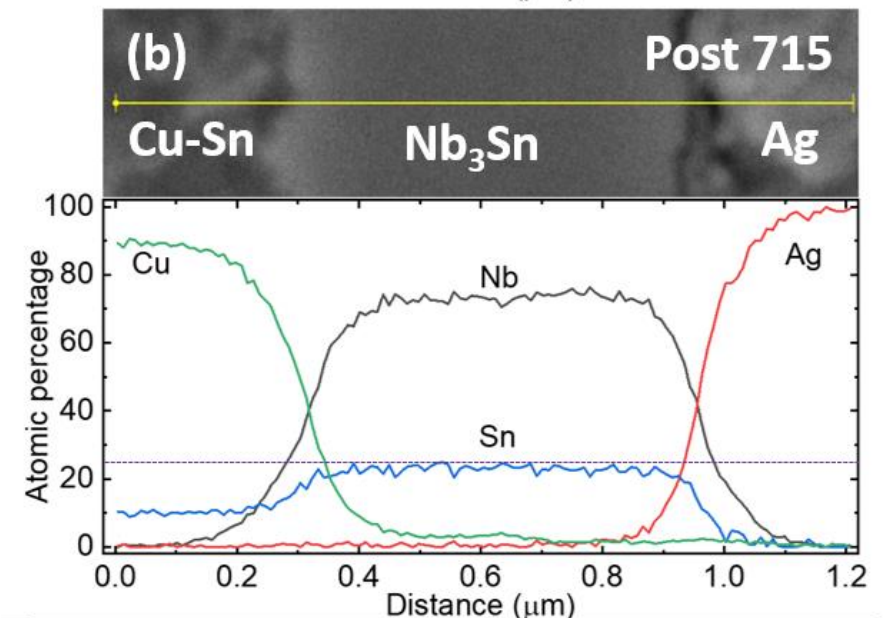
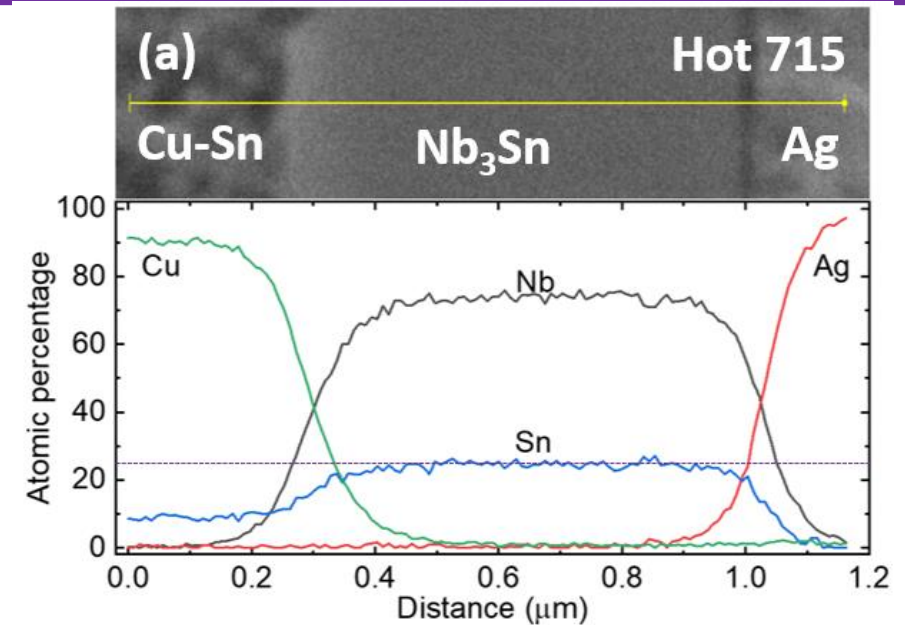
XRD shows an additional strain component for the hot bronze approach



- Full reaction to Nb₃Sn phase at 715°C
 - Unreacted Nb at lower reaction temperature
- Strain in hot bronze samples $\sim -1.2\%$
- Strain in post-reaction samples $\sim -0.9\%$
- CTE mismatch (715 °C \rightarrow 25°C) creates -0.7% strain (for 10%wt. Sn bronze)
- Hot bronze samples contain additional strain component
 - Strain determined from lattice parameter by peak fitting and Poisson's ratio

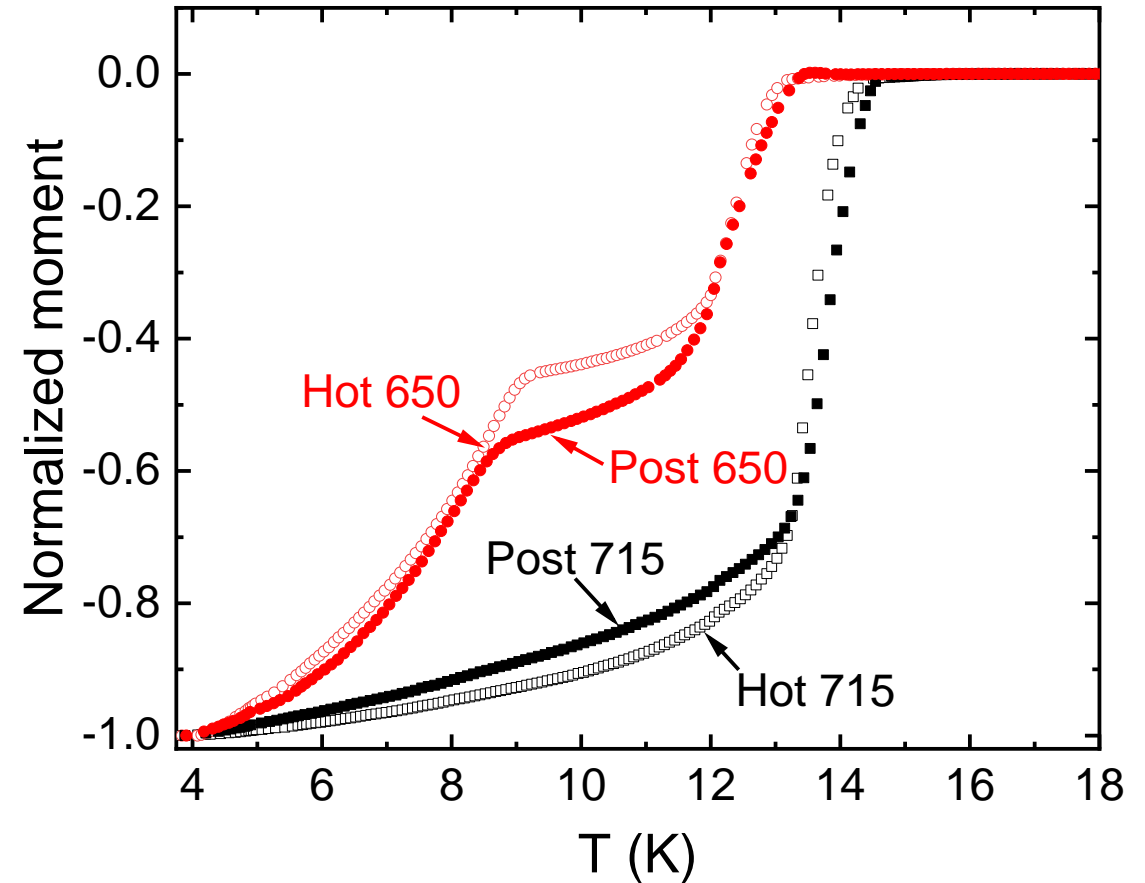
Hot bronze route gives Sn-rich Nb_3Sn ?!

- Hot bronze Sn - ~26%
- Post-reaction - ~ 24%
- Nb:Sn ratio from point scans in center of layer
- Tin-rich Nb_3Sn is not common for bronze reactions
- Relatively constant Sn content across thickness



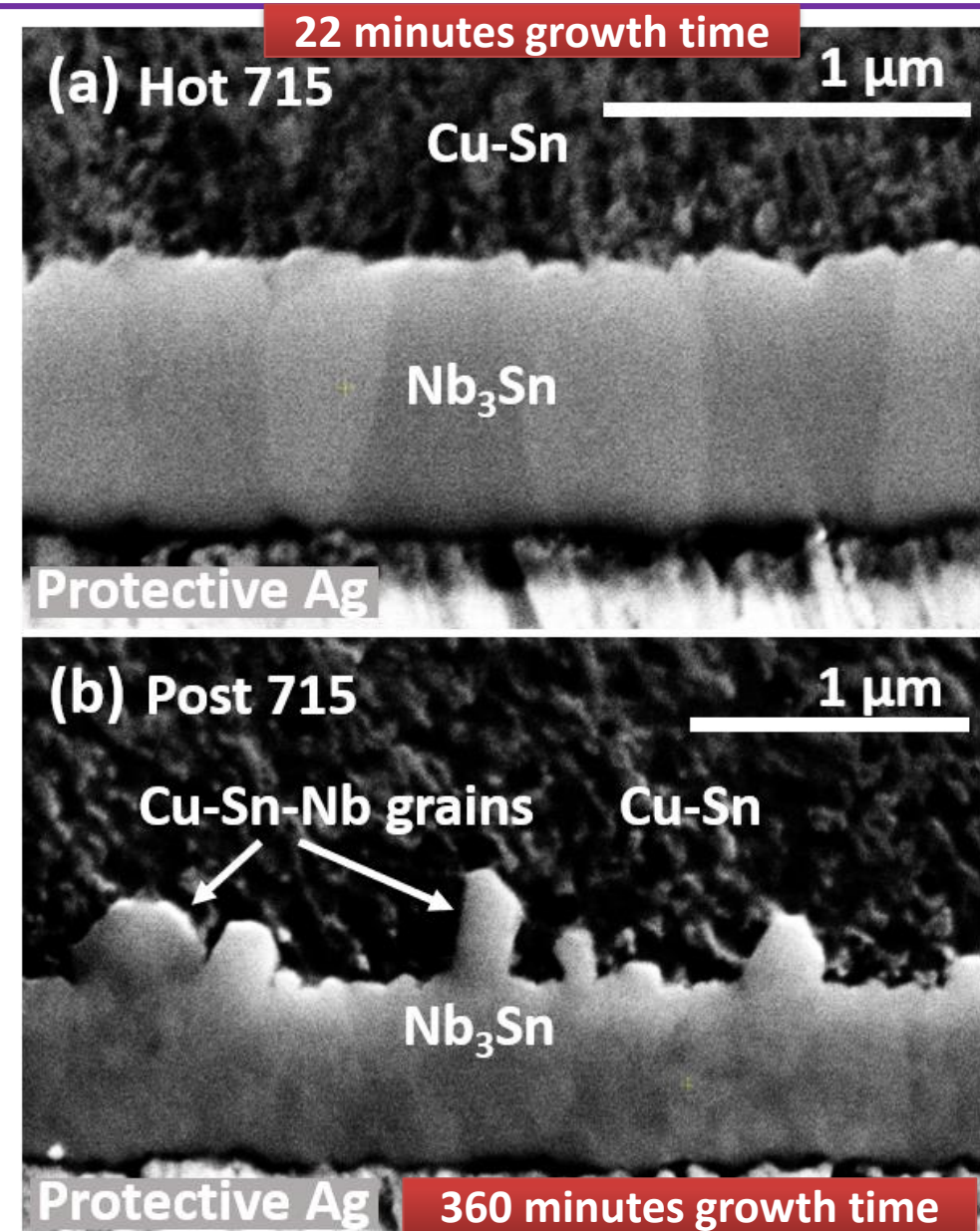
Similar T_c in both cases!

- Strain affects T_c of Nb_3Sn
 - At zero strain & 25% Sn, T_c is ~ 18 K
- Similar T_c in both approaches, $T_c \sim 14$ –
16 K due to strain from CTE mismatch
and intrinsic strain
- At lower temperature, unreacted Nb-
low Sn activity



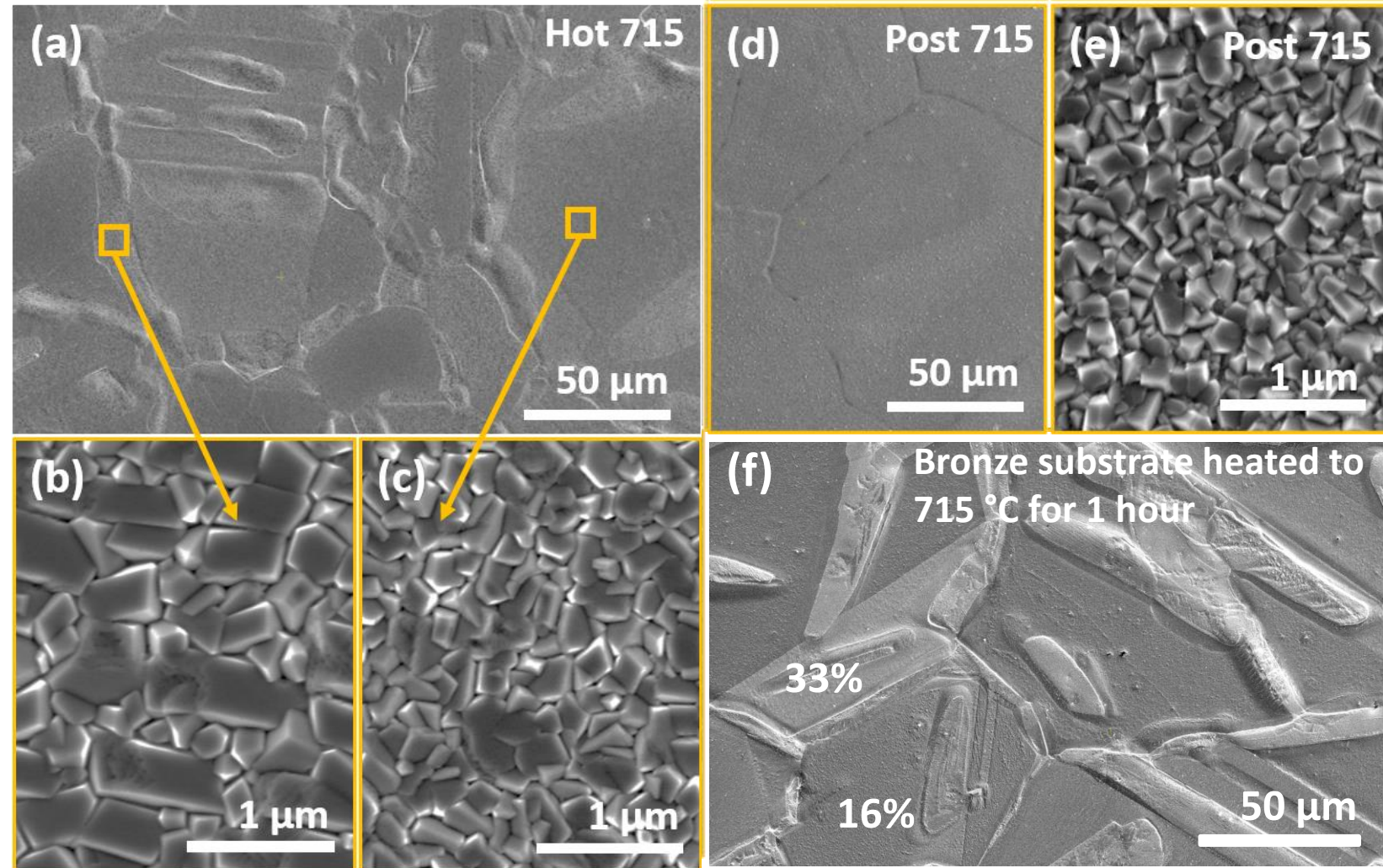
Columnar grains in hot bronze samples

- Columnar grain growth in hot bronze sample -> zone 2 in Thornton SZD
- Smaller grains in post-reaction samples
- No cracks



Larger Nb₃Sn grain in hot bronze samples

- Larger Nb₃Sn grains in hot bronze case
- Surface avg. roughness is ~10 - 15 nm in 100 μm²
- Effects from underlying bronze substrate Sn contents seen in hot bronze samples
- No cracks

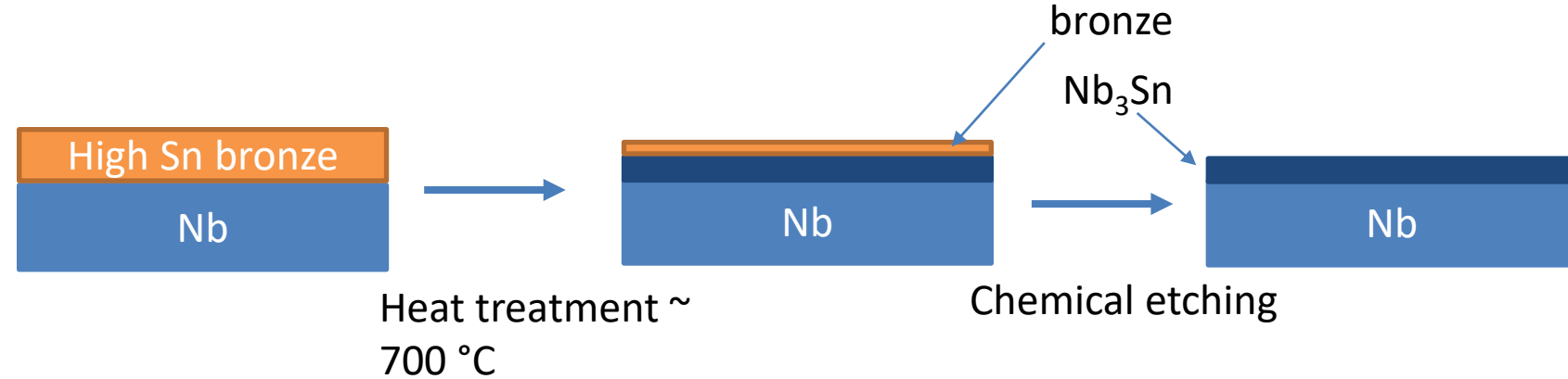


Discussion – Nb on hot bronze

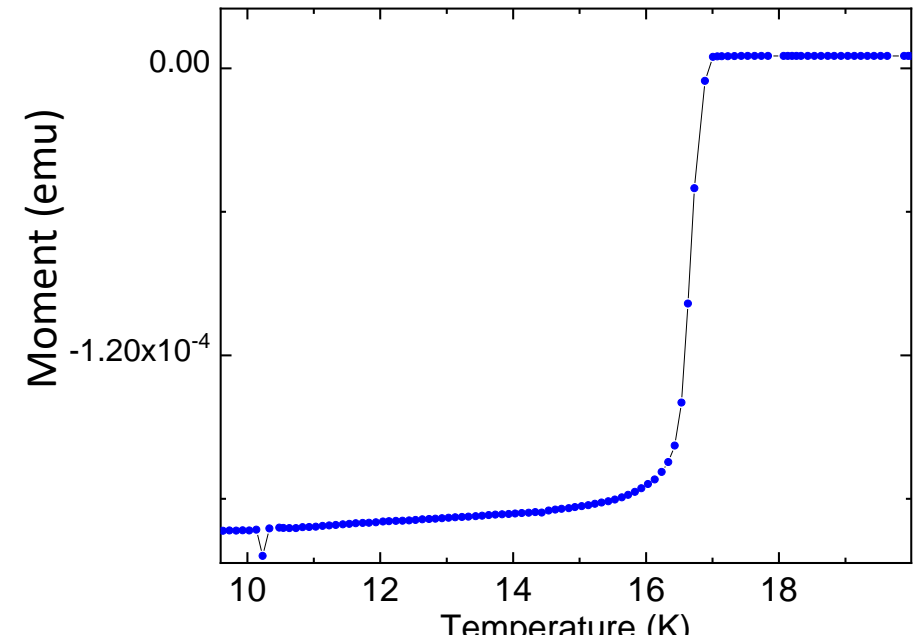
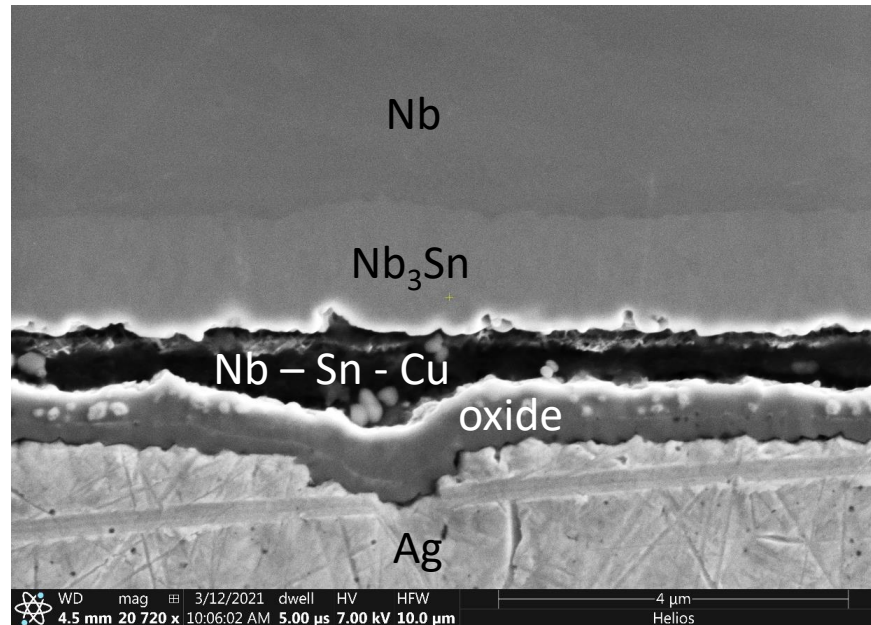
- Tin-rich Nb₃Sn: 26% suggested by EDS
- But *same* T_c as for post-reaction, which EDS suggested was only 24% Sn
- Why?
 - Excess strain seen from XRD
 - Zone 2 in Thornton SZ diagram
 - **Is the excess strain due to a new growth mode? We think so:
high mobility Volmer-Weber growth**
 - If so, this is new, and it is not available in solid-state reactions

Bronze on Nb

- Evaporation of bronze reduces steps involved
- Sn gradient in the Nb_3Sn layer, 25% to 18%
- Susceptible to oxidation

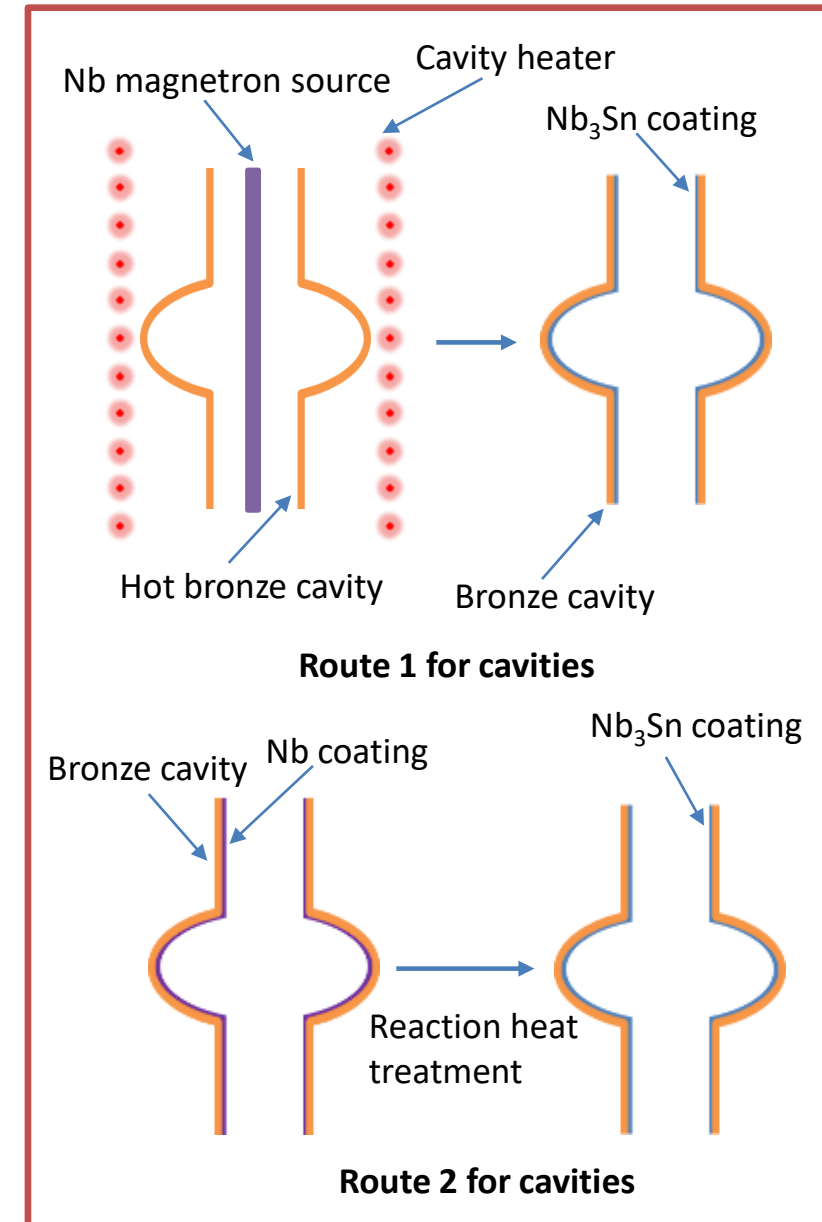
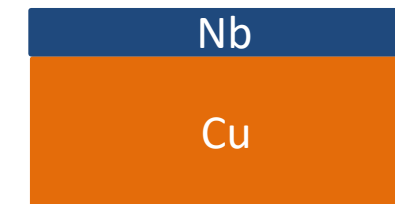
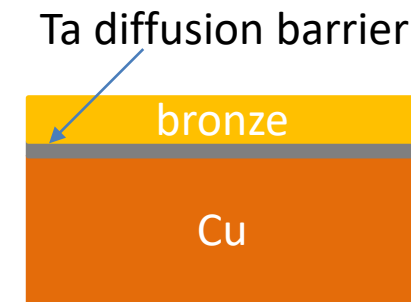


2 μm 40% Sn bronze layer on Nb + 6 hours at 700 °C



How to use Cu substrate and scaling to cavities

- 2 – 5 μm high Sn bronze layer is enough to produce $\sim 1\text{-}2\ \mu\text{m}$ Nb_3Sn layer
- Diffusion barriers \rightarrow stop Sn diffusion into Cu
- Bronze/Ta/Cu can be used in hot-bronze or post-reaction method
- Nb_3Sn grain size can be controlled with Sn content in bronze layer
- Nb/Cu can be used with bronze evaporation + heat treatment+etch approach
- Process can be scaled to cavity coating easily



Summary

- Deposition of Nb onto hot bronze seems to be very promising
 - Tin-rich coating
 - Rapid growth of layer
 - Zone 2 grain structure
- A wide range of substrate configurations is possible
 - Diffusion barriers and CTE adjustments
 - Compatibility with external copper bulk
- Schemes for scaling to cavities are in progress

Acknowledgment

- Alexander Wozny and Jonathan Wozny for their support with bronze substrate polishing.
- This work at ASC, NHMFL-FSU was funded by U.S. Department of Energy, Office of Science, Office of High Energy Physics under Award No. DE-SC 0018379.
- A portion of this work was performed at the National High Magnetic Field Laboratory, which is supported by National Science Foundation Cooperative Agreement No. DMR-1644779 and the State of Florida.

Backup Slides

Typical stress development in high-mobility VM growth

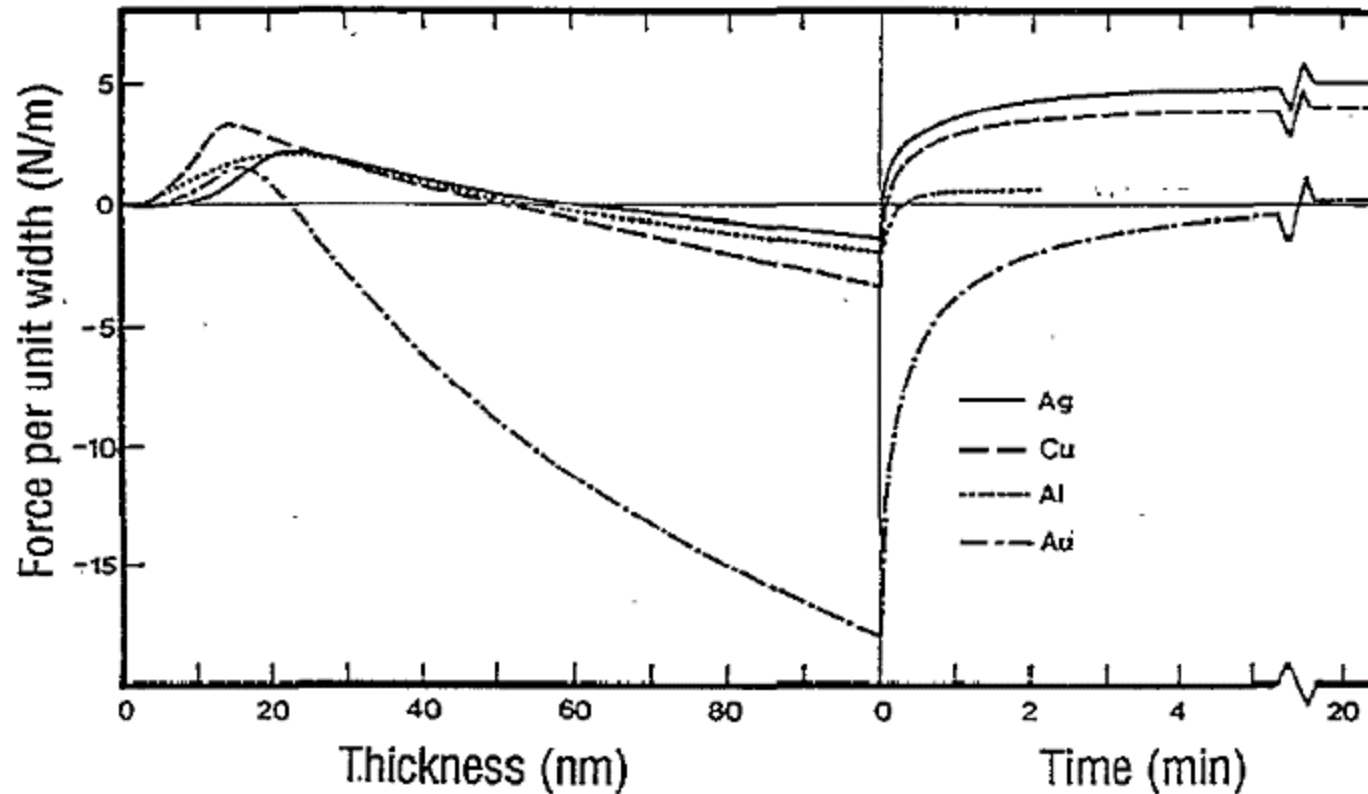
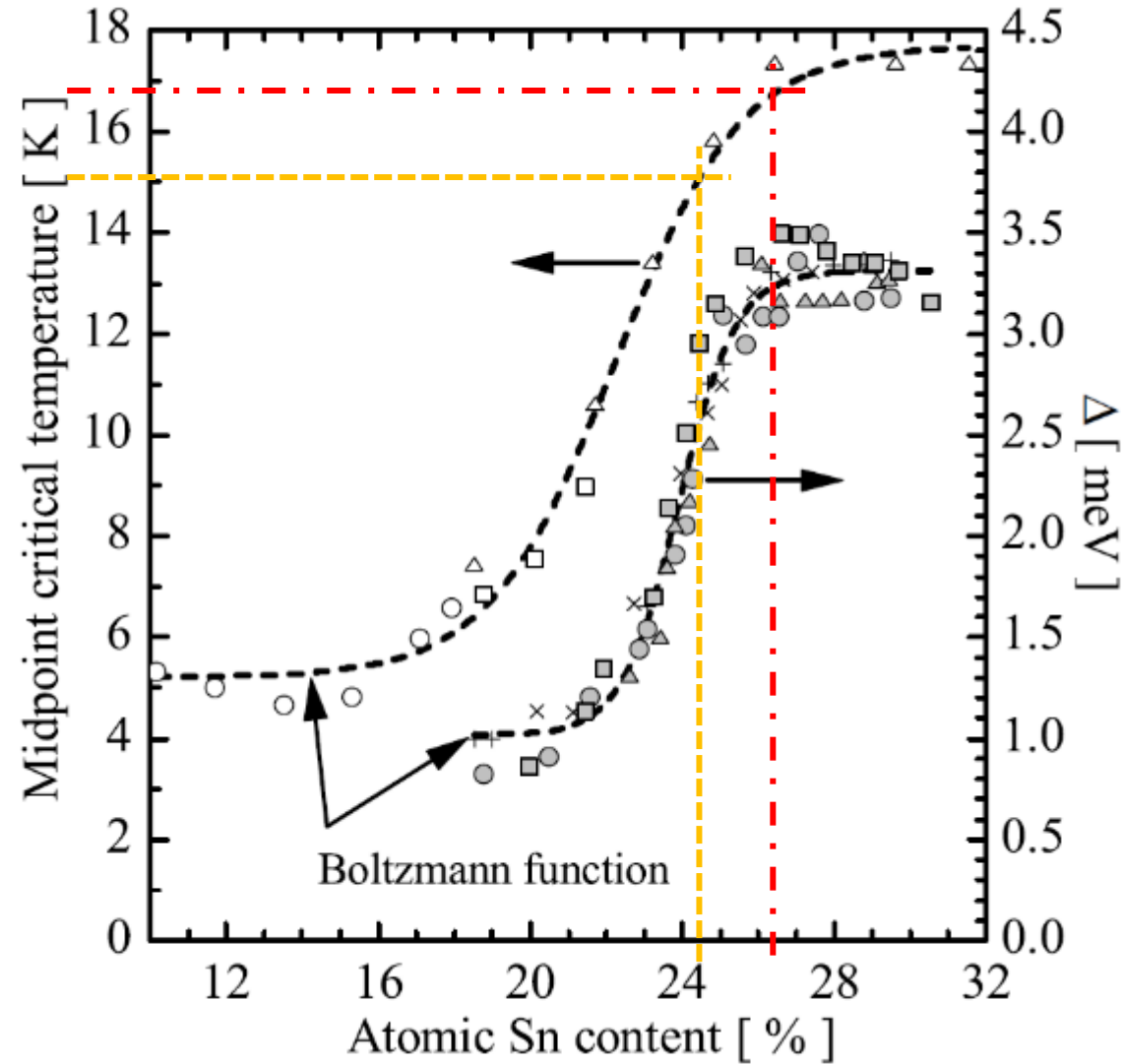


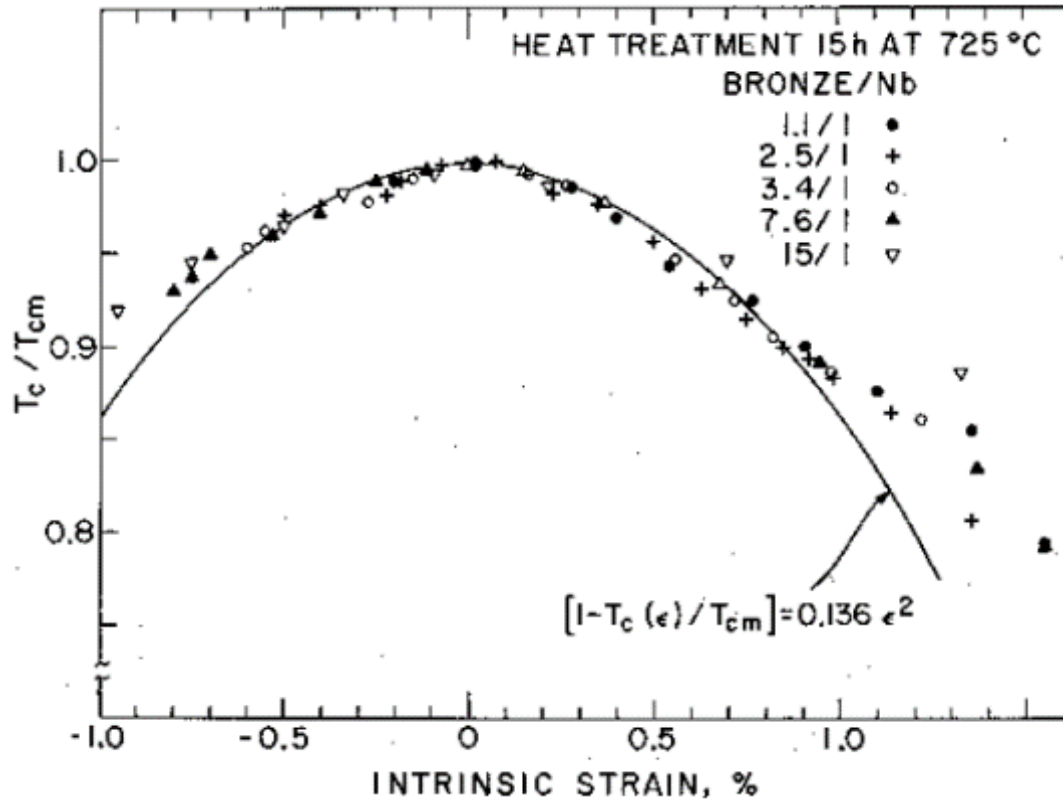
Figure 9. High-mobility Volmer-Weber growth: film forces per unit width against mean thickness (left-hand side) and time (right-hand side) of Ag, Cu, Au (from [16]) and Al films (from [9]) by courtesy of R Abermann UHV deposited at room temperature onto MgF_2 coated glass substrates.

Koch R 1994 *J. Phys.: Condens. Matter* 6 9519

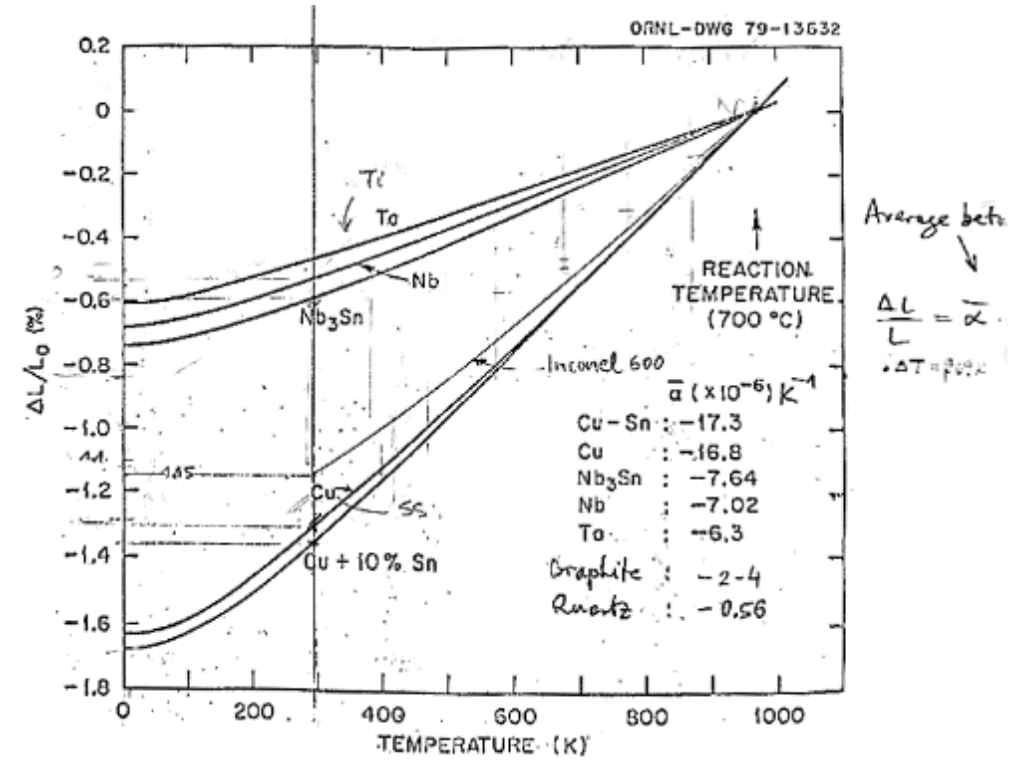
T_c vs Sn content



CTE mismatch

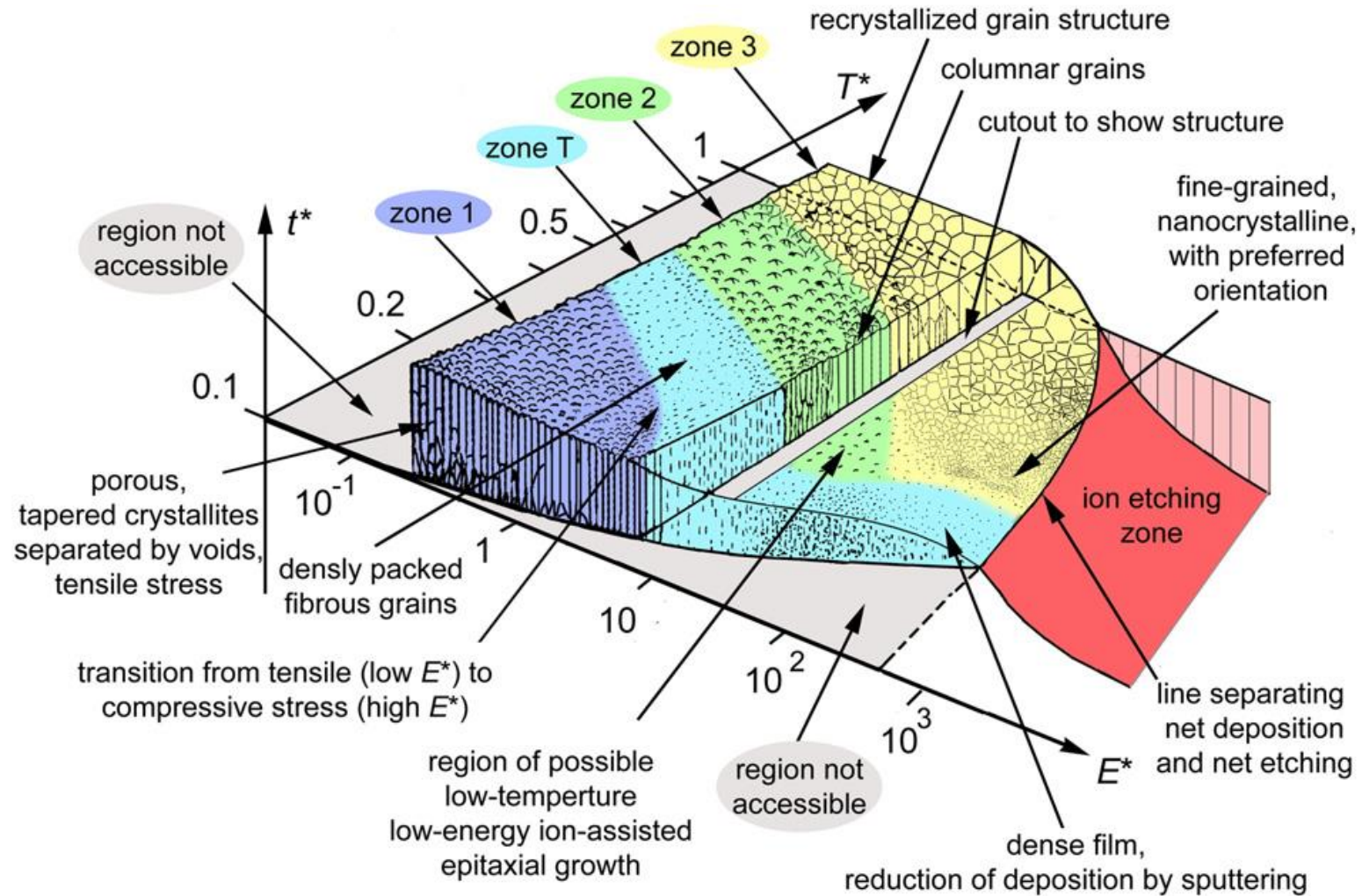


D. O. Welch, *Adv. Cryo. Eng.*, 26, pp. 48-65, 1980



CTE mismatch between Nb₃Sn and common substrates

SZD



Proposed deposition chamber

