

# Cristian Pira

# Nb<sub>3</sub>Sn films via liquid tin diffusion for SRF application

Work supported by the INFN CSNV experiment TEFEN



This project has received funding from the European Union's Horizon 2020  
Research and Innovation programme under Grant Agreement No 730871.



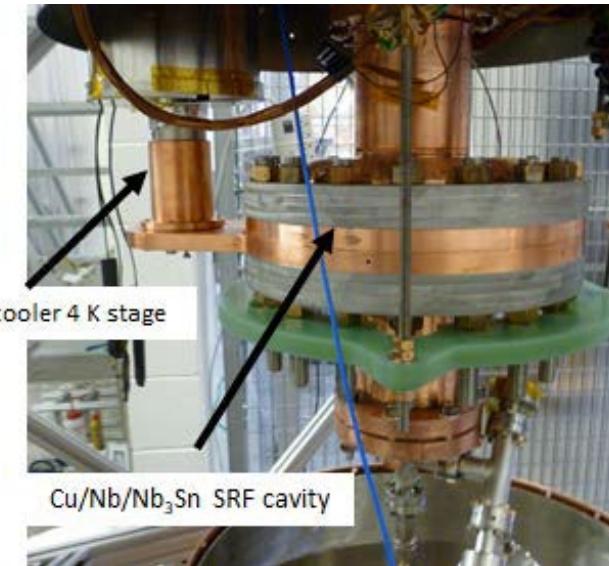
# $\text{Nb}_3\text{Sn}$ on Cu - Motivation

High performance of  $\text{Nb}_3\text{Sn}$  @ 4.2 K → cooling by cryocooler

High thermal conductivity substrate is preferred



Courtesy of G. Ciovati, JLab



# $\text{Nb}_3\text{Sn}$ @ LNL - Motivation

**Goal of I.FAST collaboration** → Produce a  $\text{Nb}_3\text{Sn}$  coated cavity on Cu

**Request** → A  $\text{Nb}_3\text{Sn}$  target for cylindrical configuration

**Idea** → Single-use  $\text{Nb}_3\text{Sn}$  target by Liquid Tin Diffusion (LTD)

# LTD Motivation

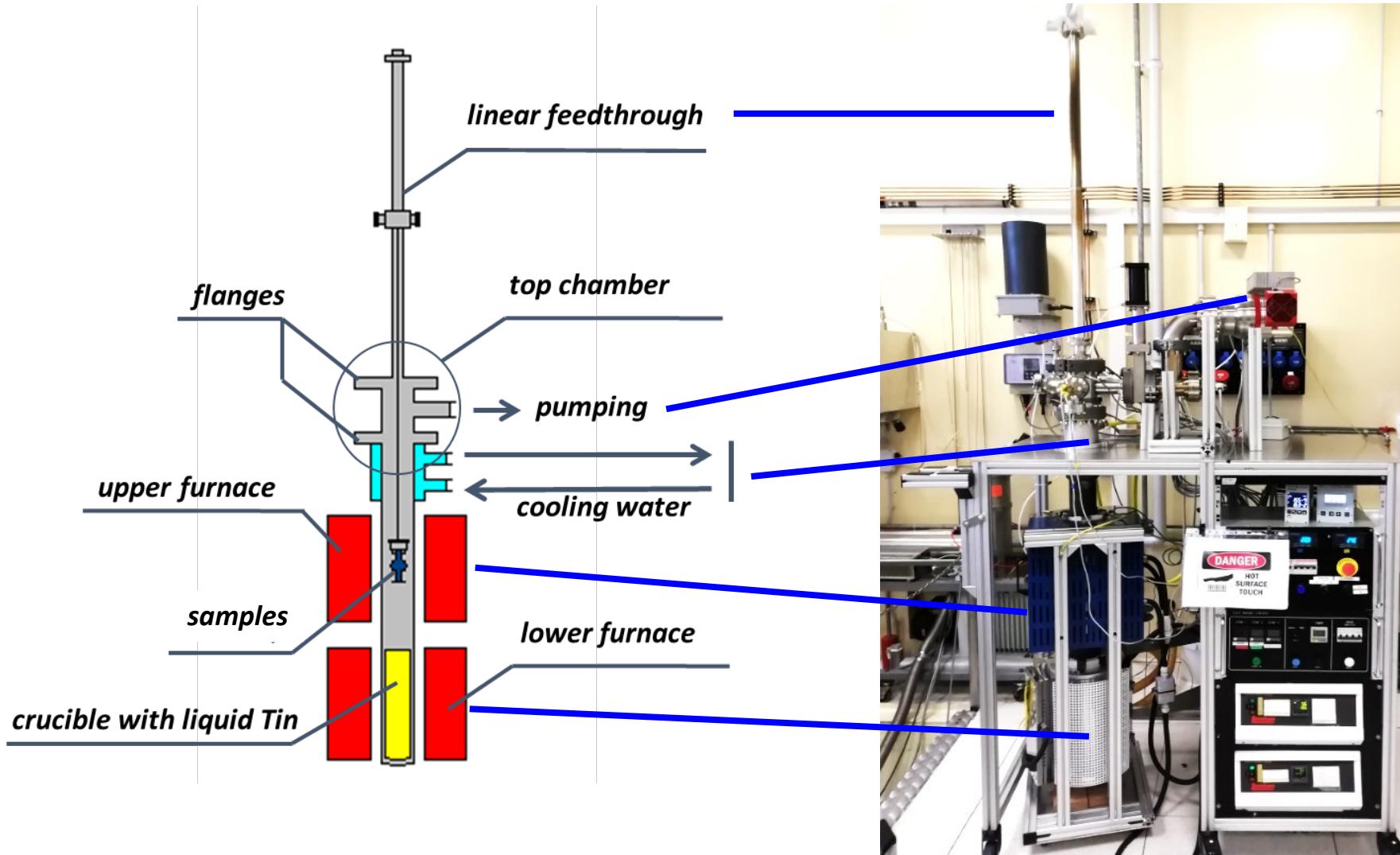
Already explored for SRF @LNL (from 2005 to 2010)

Simple process

Possibility to adopt solutions developed for Tin Vapour Diffusion

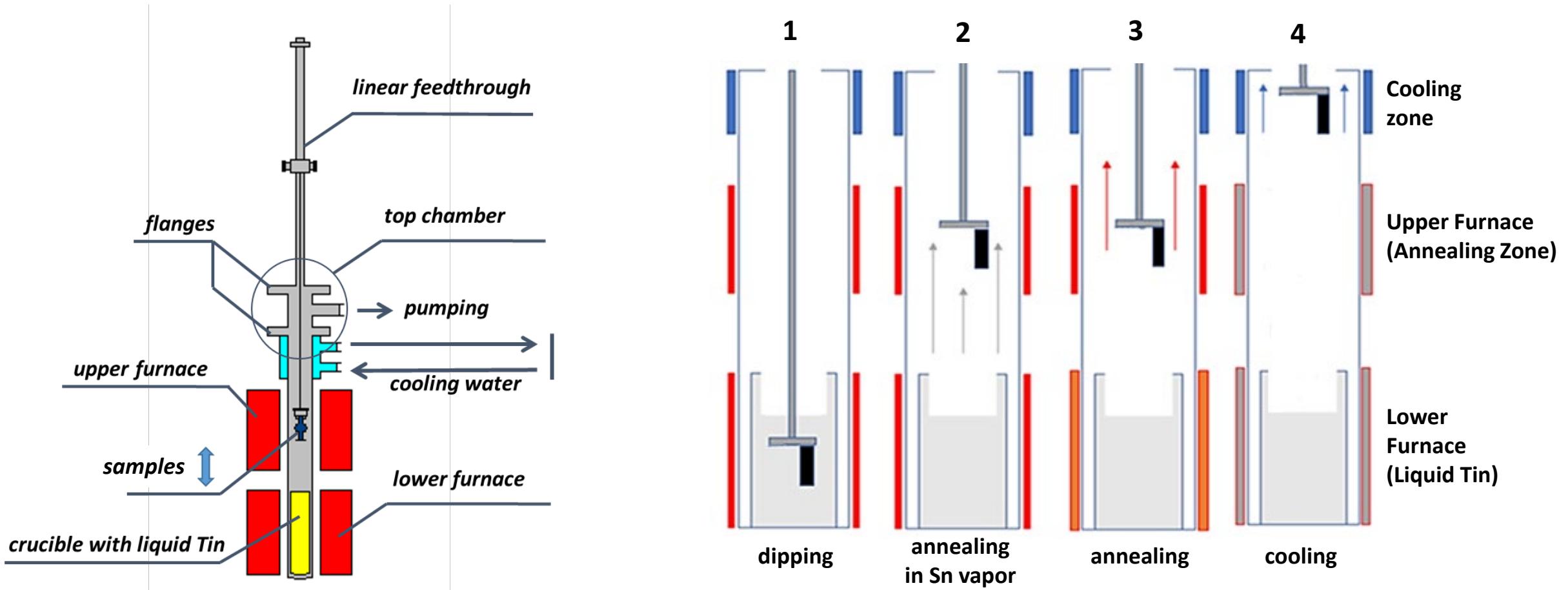
Possibility to **grow thick layer** of  $\text{Nb}_3\text{Sn}$  ( $> 10 \mu\text{m}$ )

# Liquid Tin Diffusion process set-up



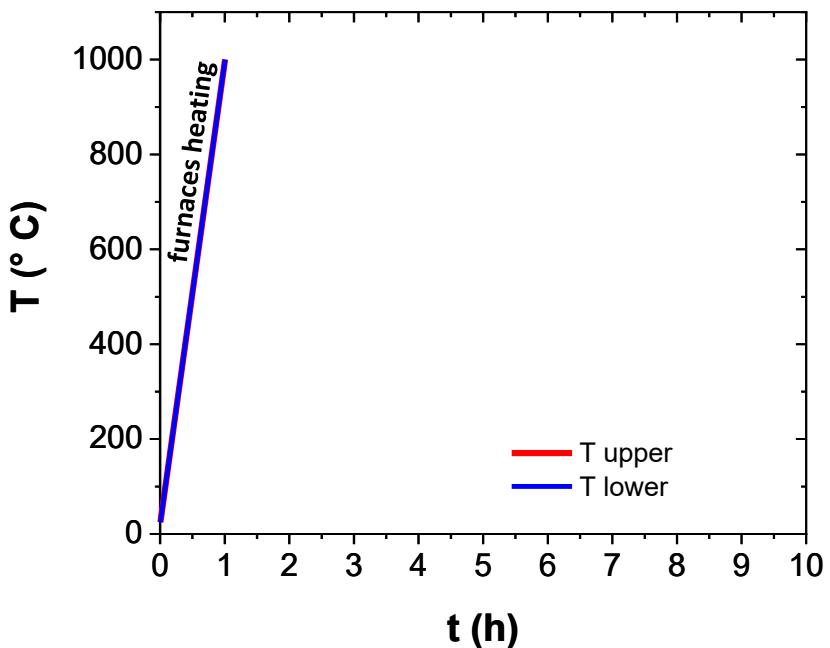
# Liquid Tin Diffusion process (LNL 2006)

S. M. Deambrosis et al., "Al<sub>5</sub> superconductors: An alternative to niobium for RF cavities," *Physica C*, 2006

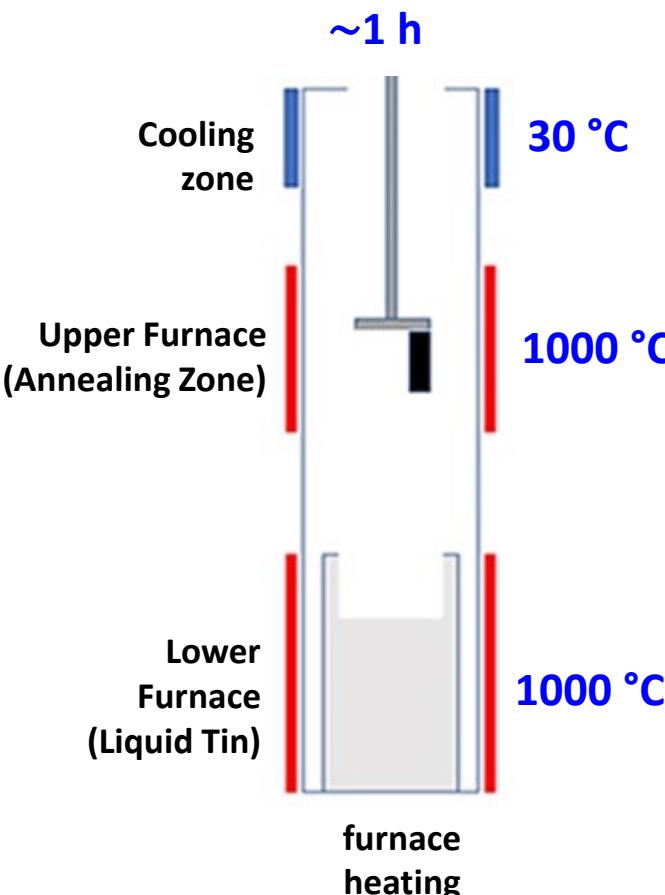


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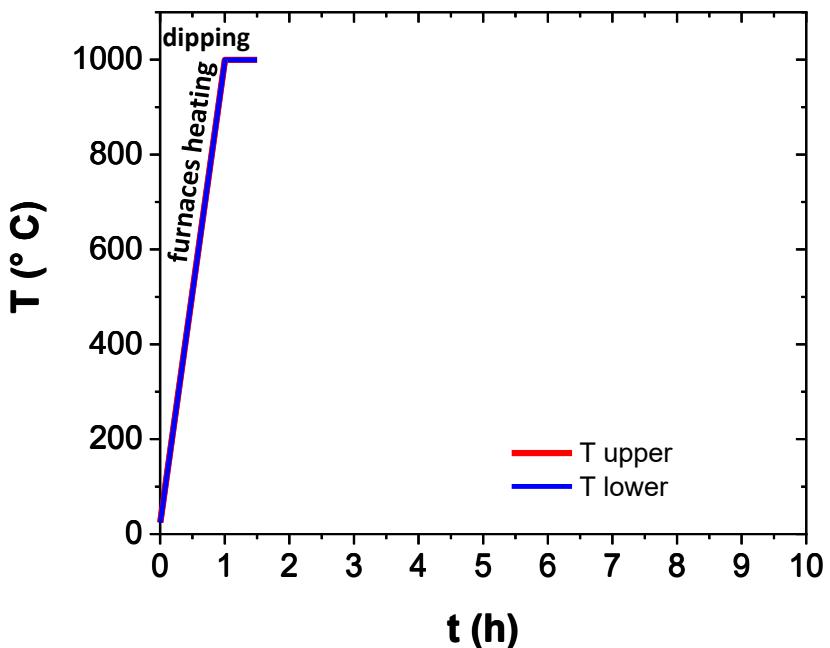


Temperature furnaces profile

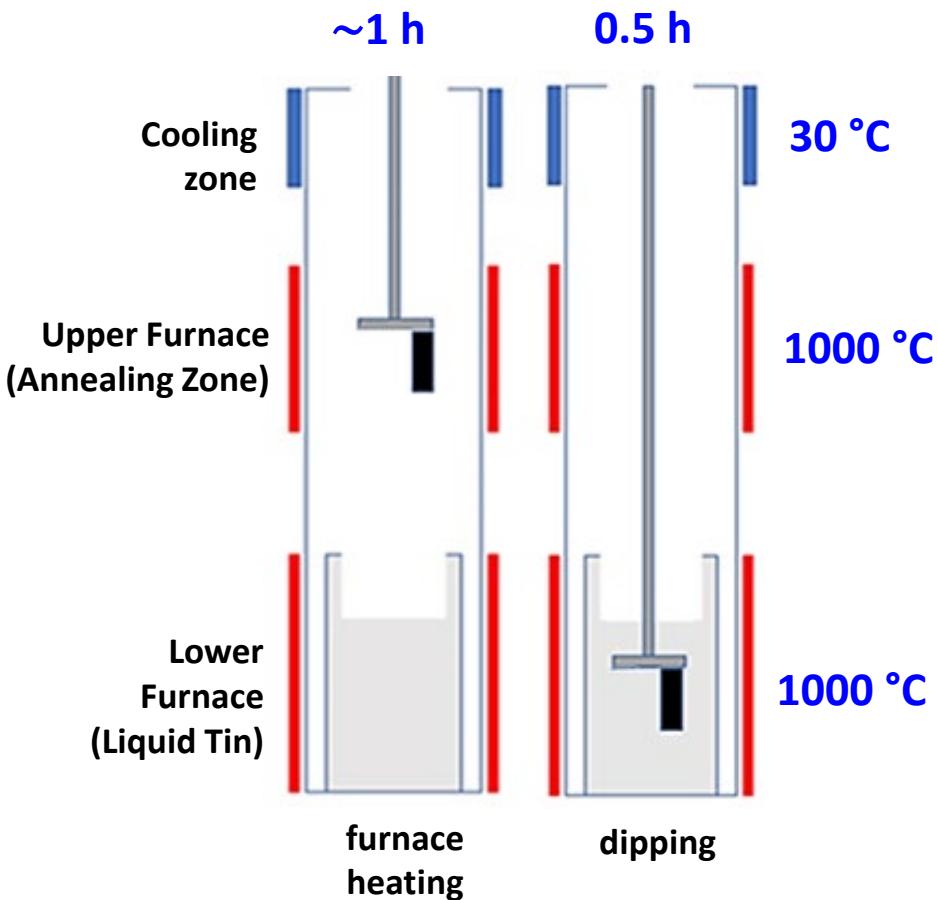


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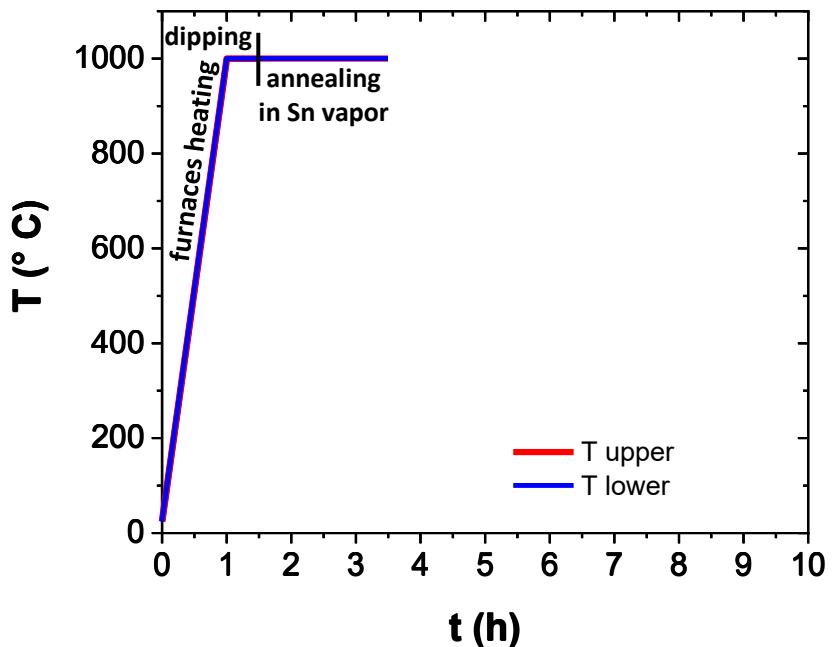


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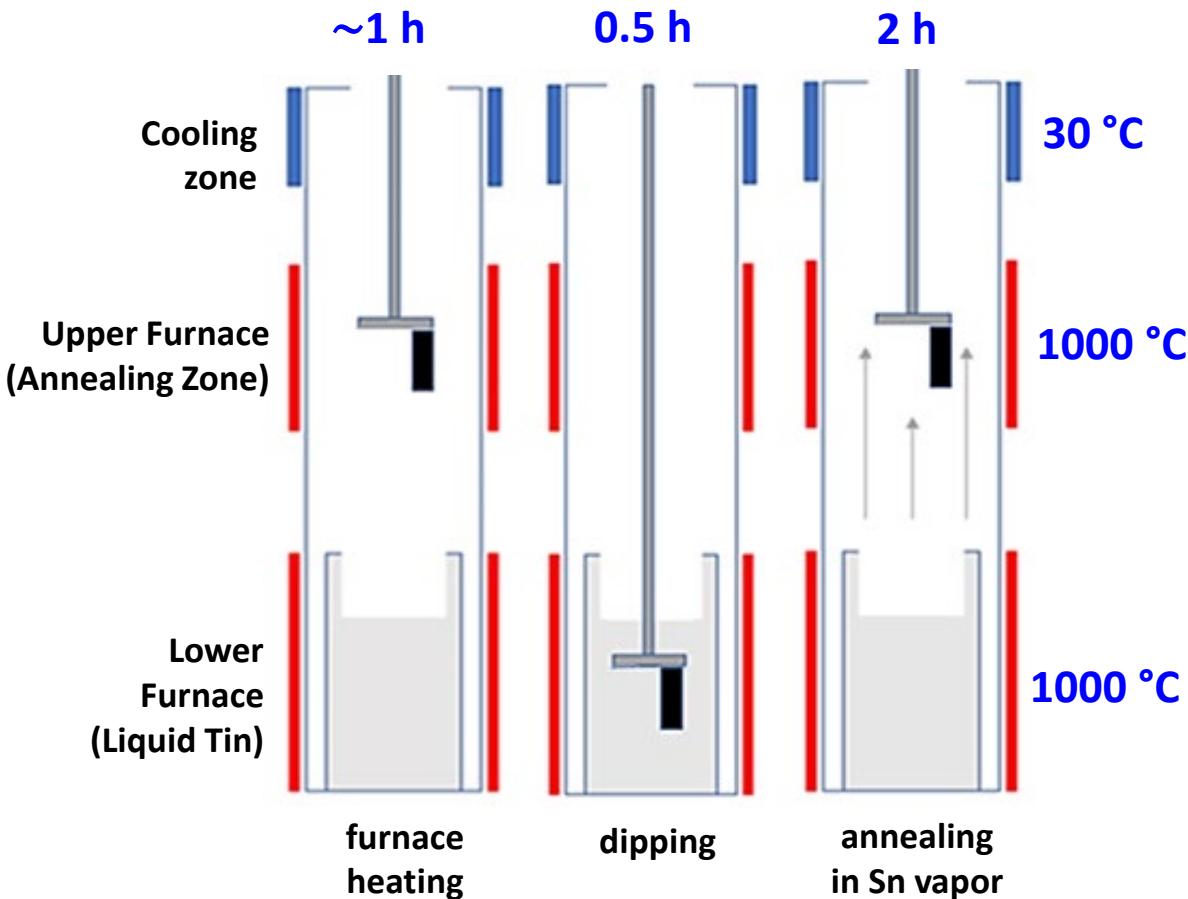


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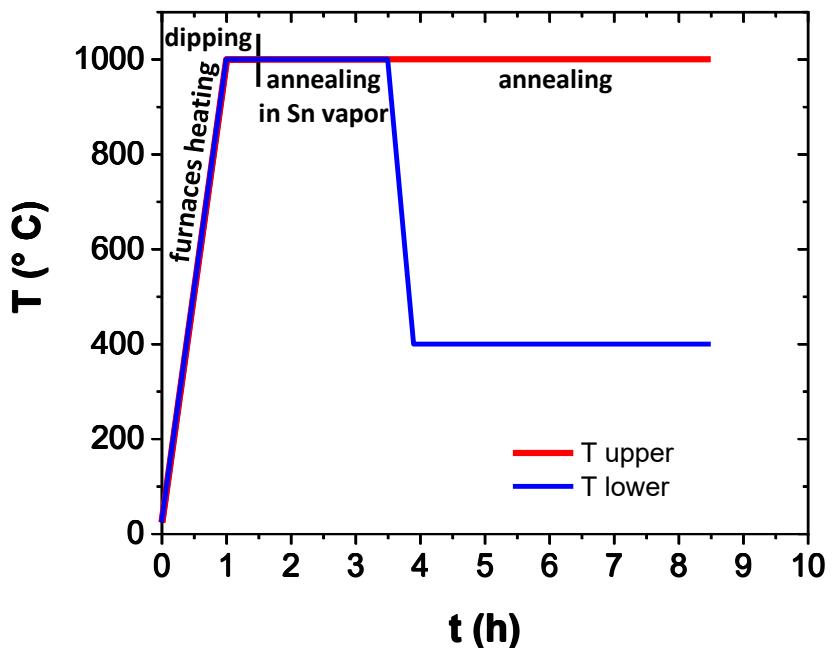


Temperature furnaces profile

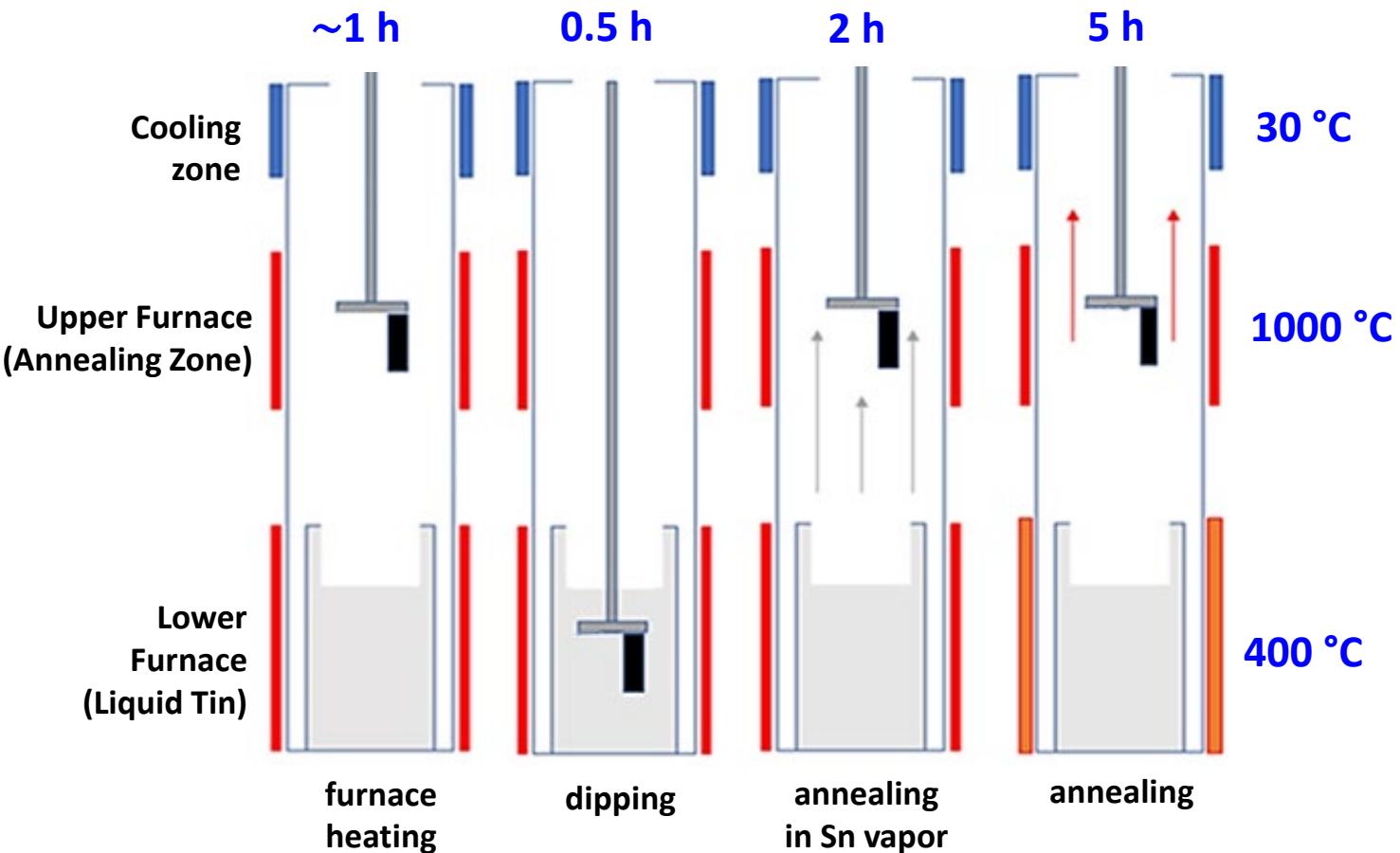


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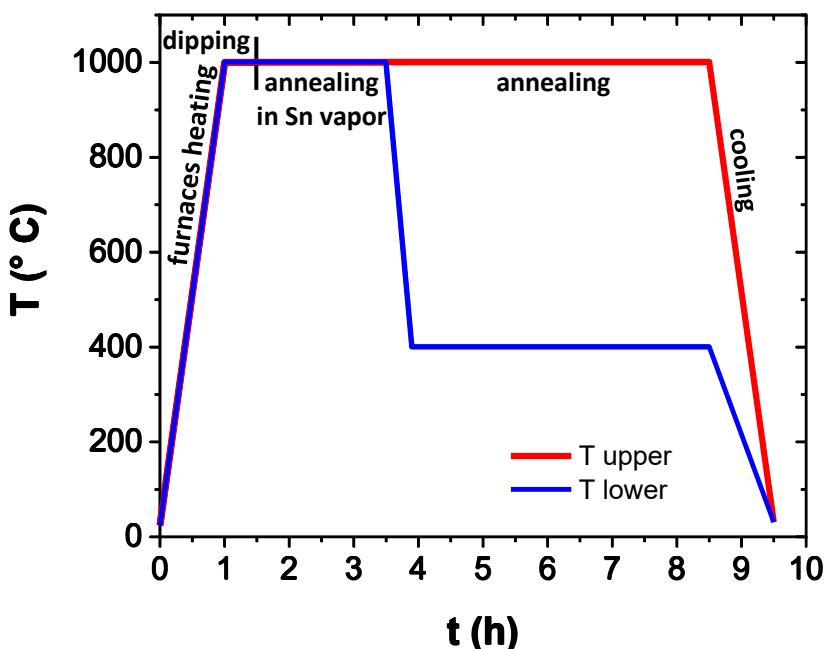


Temperature furnaces profile

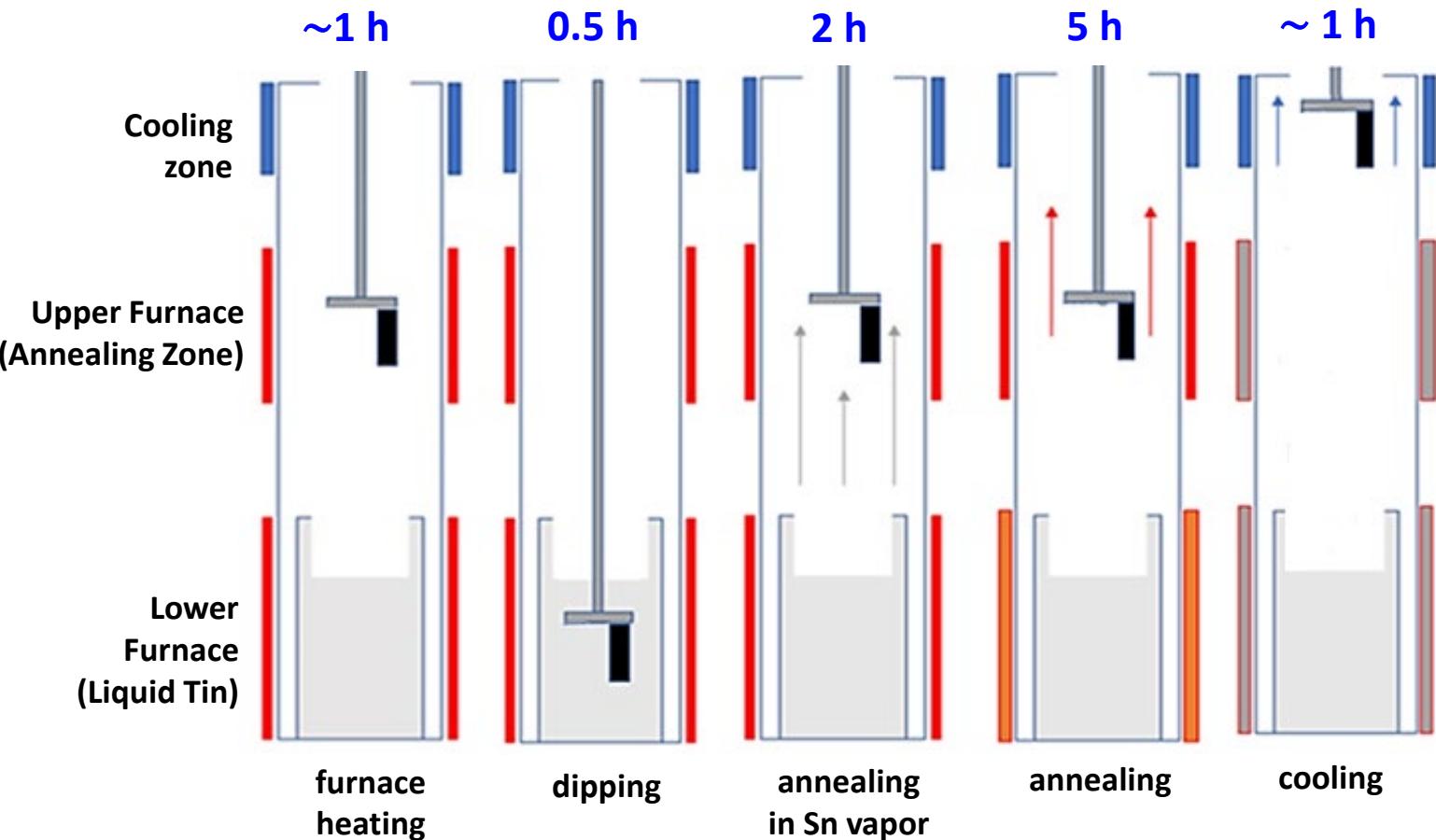


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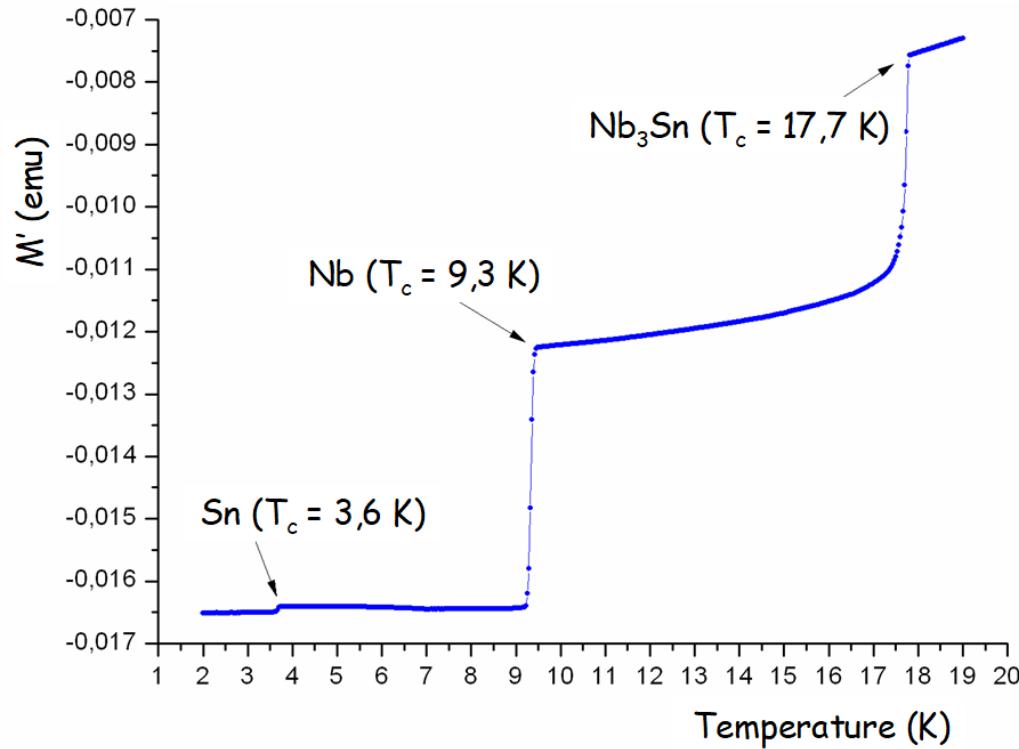


Temperature furnaces profile

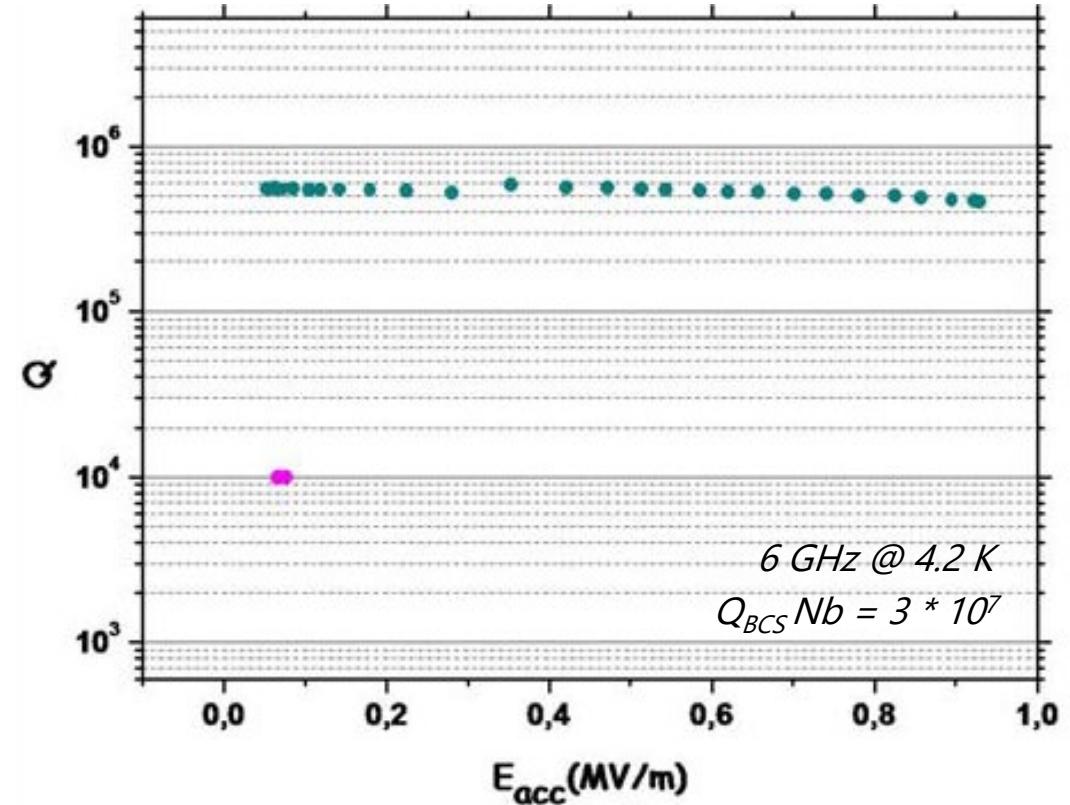


# Results (LNL 2006)

S. M. Deambrosis et al., "Al5 superconductors: An alternative to niobium for RF cavities," *Physica C*, 2006  
S. M. Deambrosis, "SRF2009"



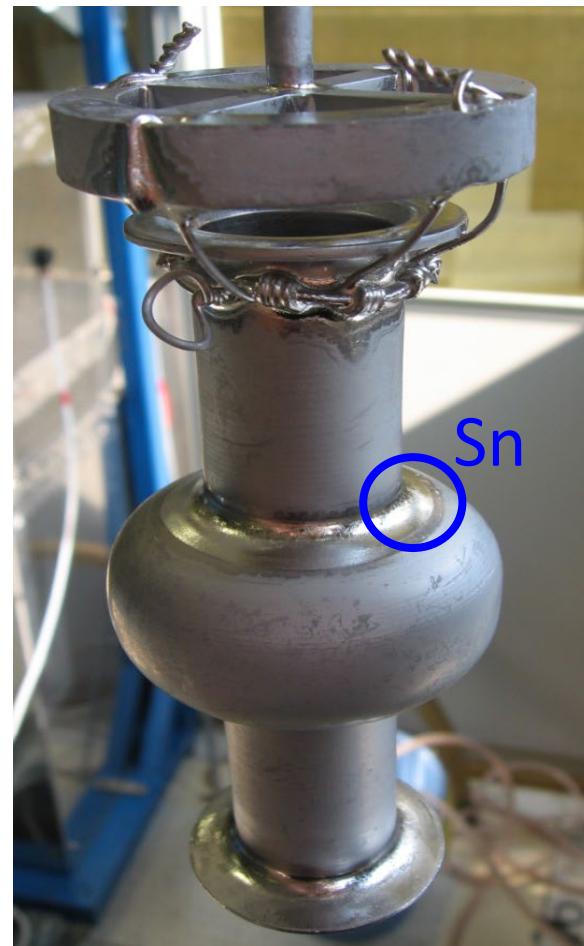
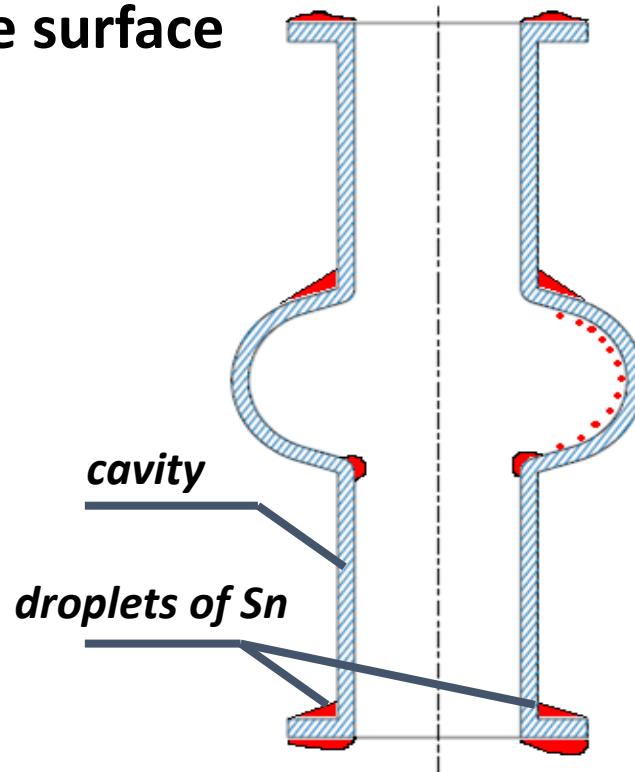
Good performance on planar samples



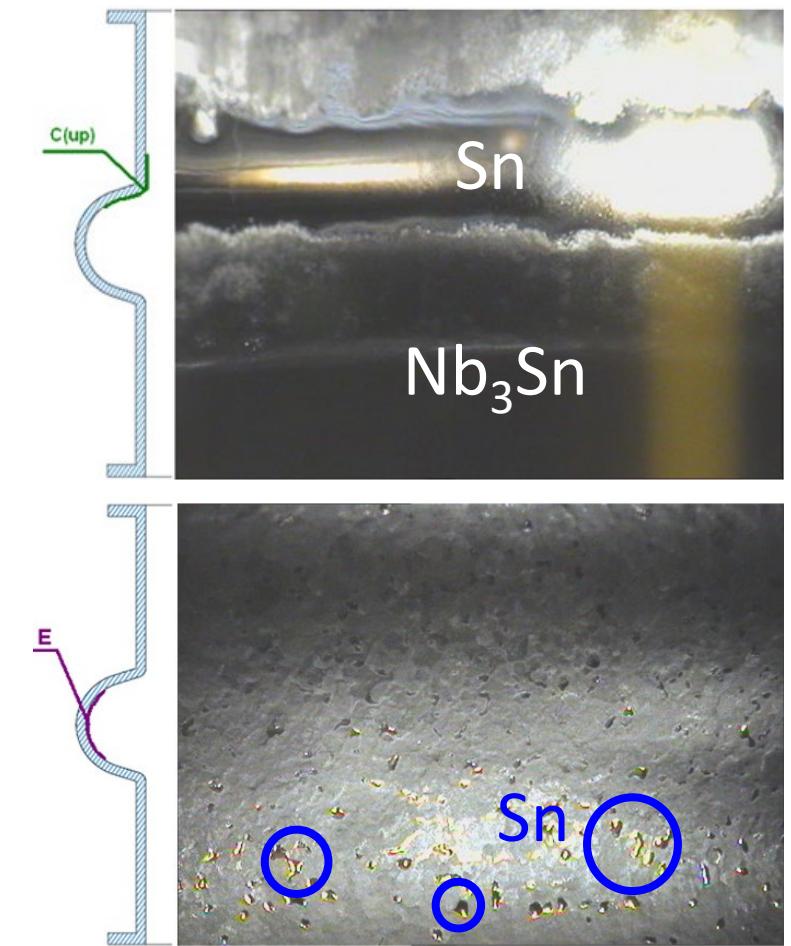
Poor performance on 6 GHz cavity

# Results (LNL 2006)

Drops of  
**unreacted Tin**  
on the surface

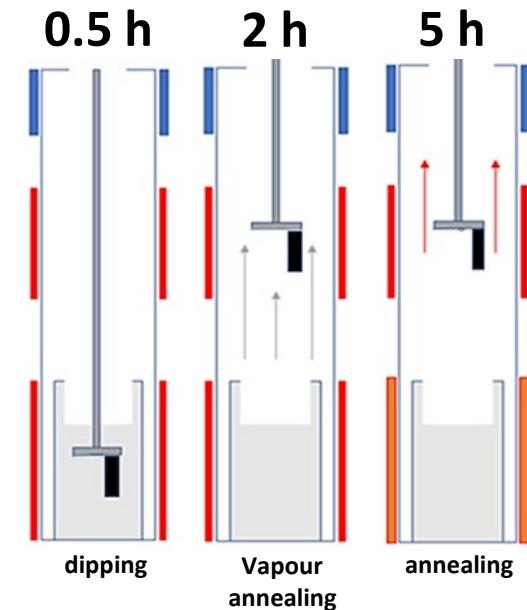
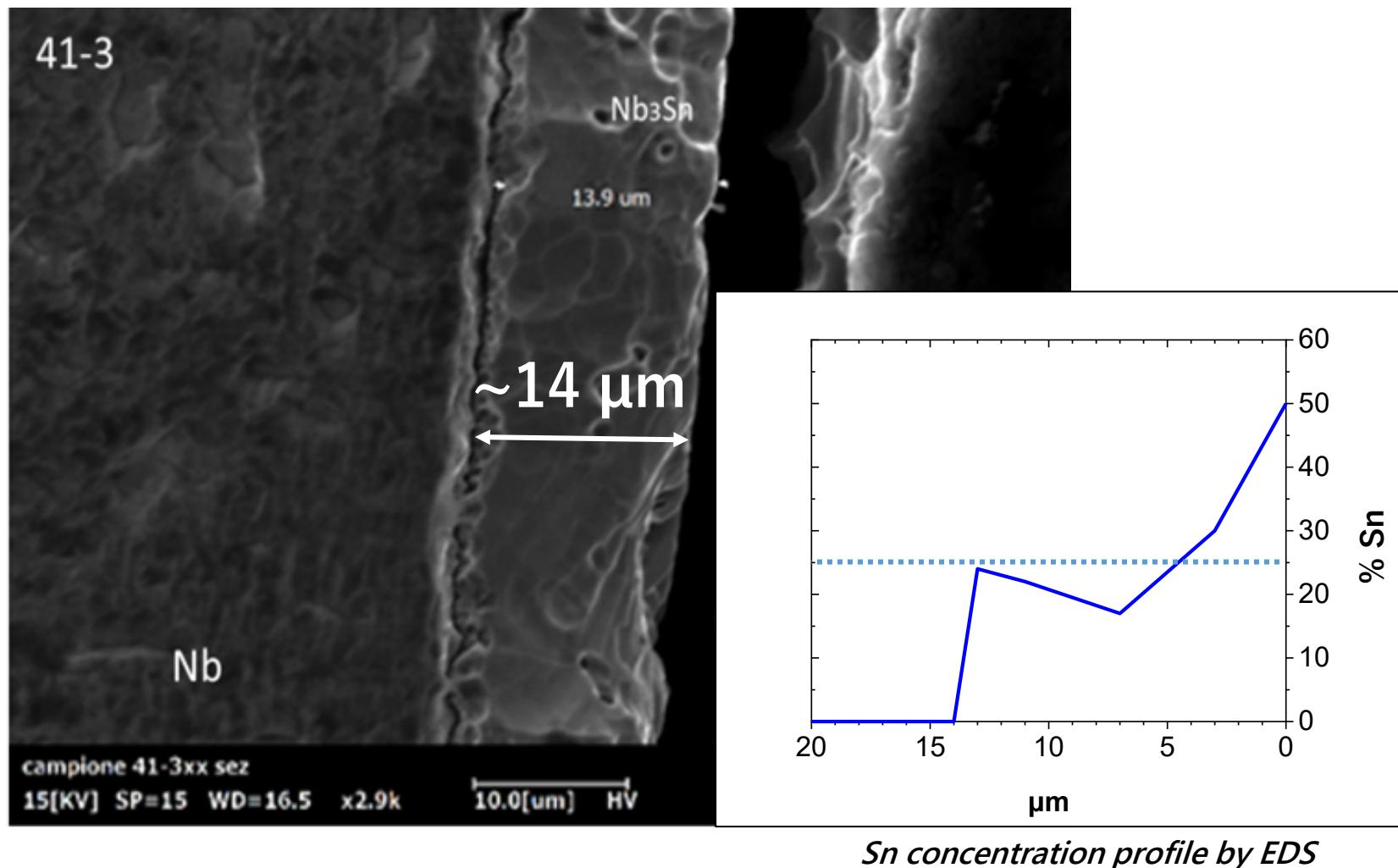


External surface

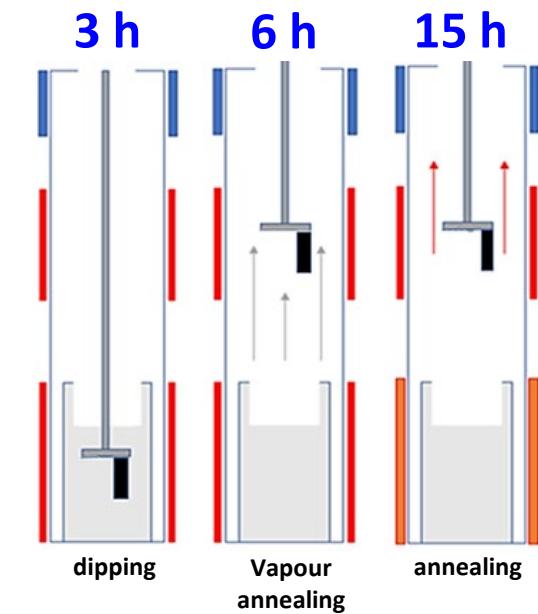
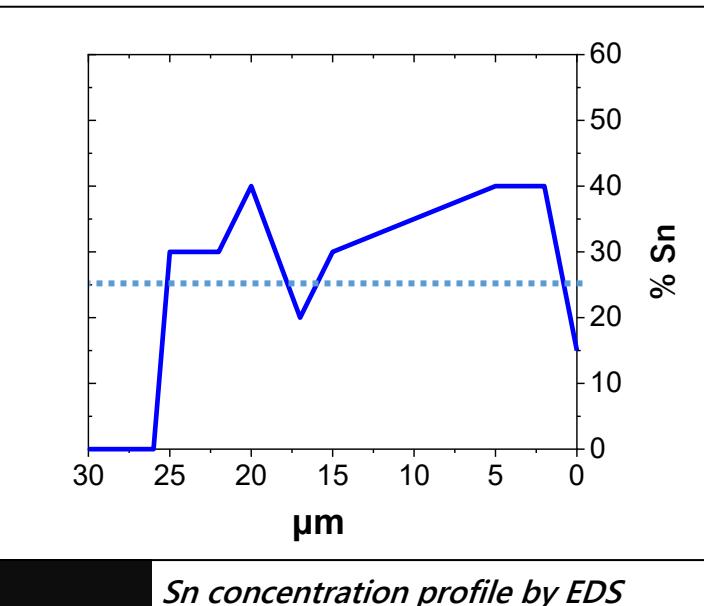
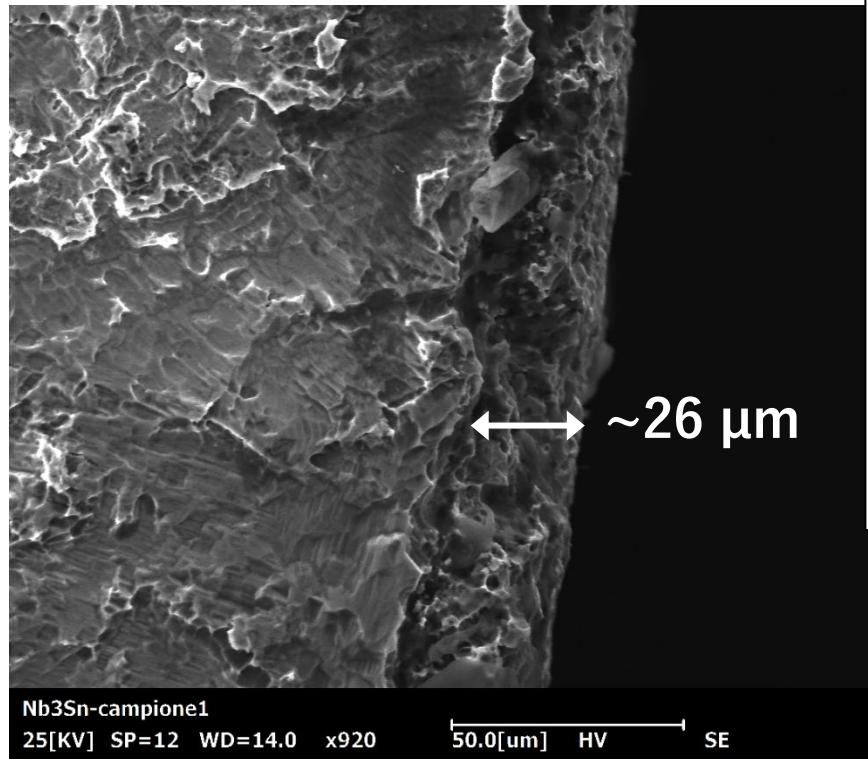


Internal surface

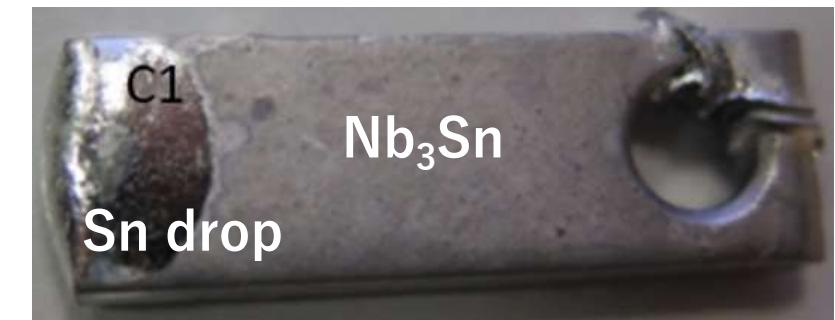
# Coating thickness (old process, 2006)



# Increasing coating thickness (old process, 2021)

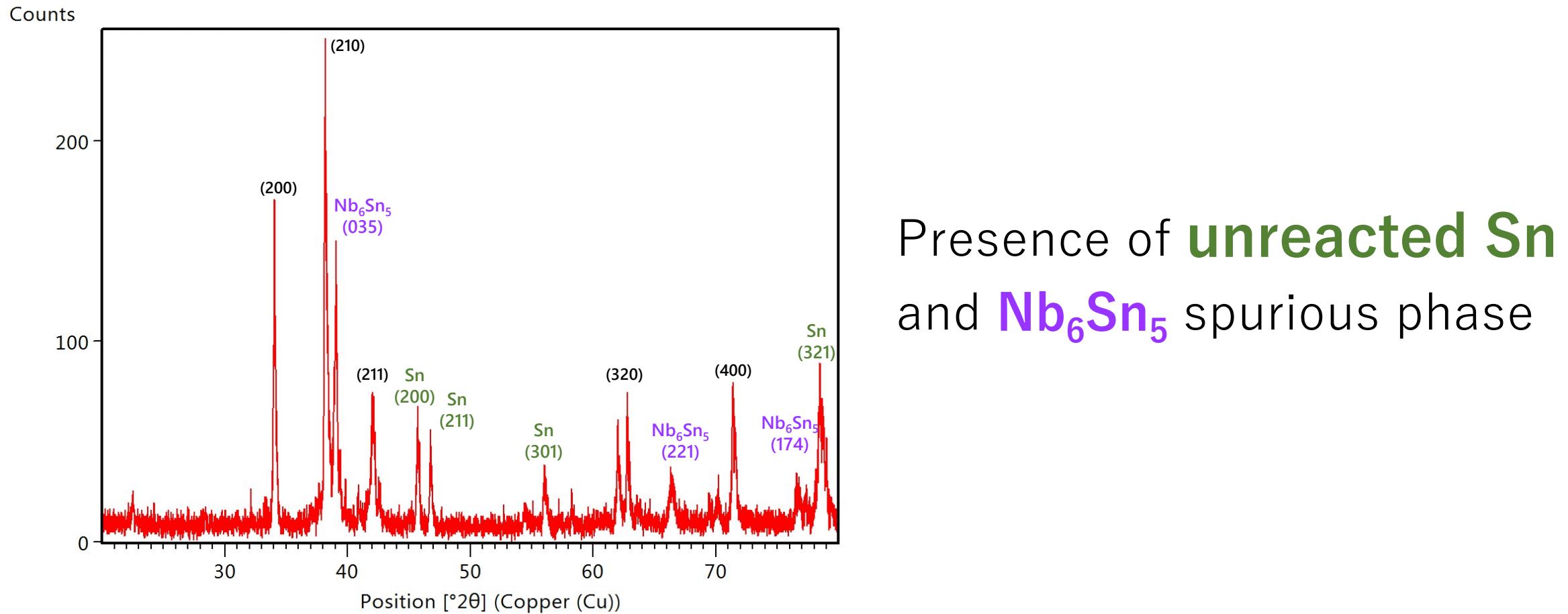


Process time increased



26  $\mu\text{m}$  thick!!! → Thickness can be modulated

# XRD (old process, 2021)



# Morphology (old process, 2021)

Same process, different substrate



Nb substrate (BCP)

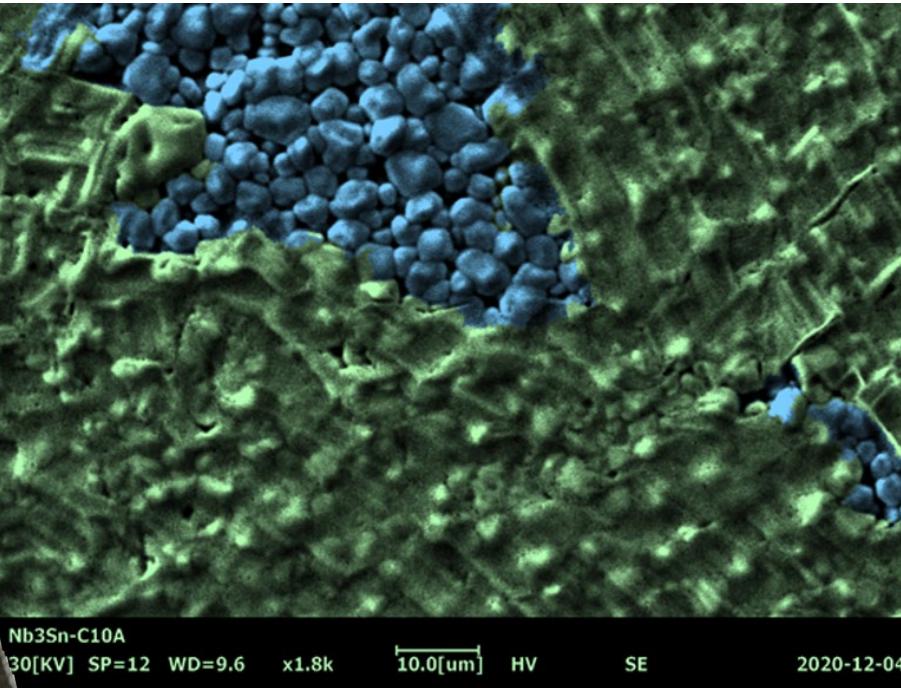
$\text{Nb}_2\text{O}_5$  substrate (Nb BCP anodized)

# Morphology (old process, 2021)

Sn 64% Nb 36%  
Sn 26% Nb 74%

Same process, different substrate

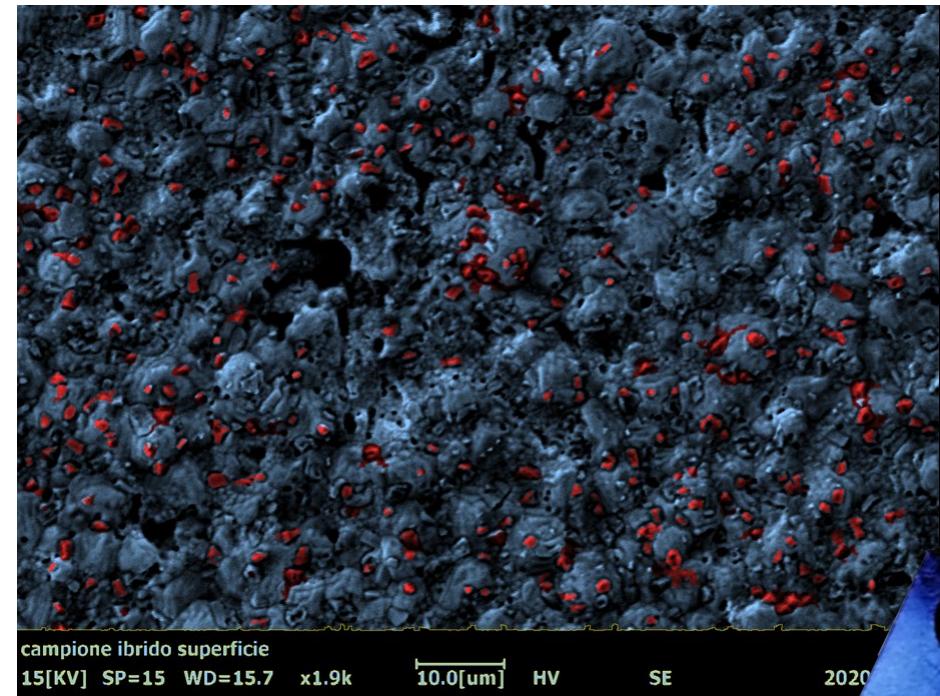
Sn 37% Nb 63%  
Sn 23% Nb 77%



Nb substrate (BCP)

*False color SEM*

Nb<sub>3</sub>Sn  
Sn  
Nb<sub>6</sub>Sn<sub>5</sub>



Nb<sub>2</sub>O<sub>5</sub> substrate (Nb BCP anodized)

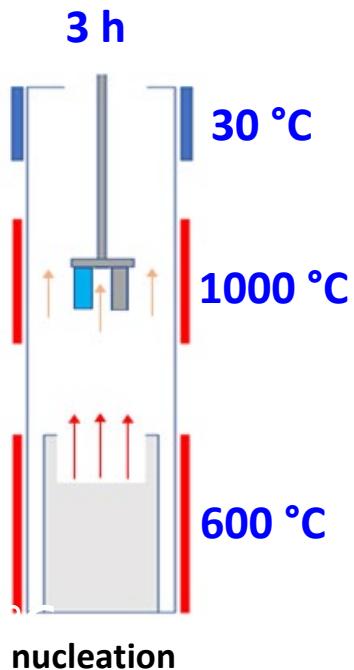
**Nucleation is important!**

(Already known in Vapor Diffusion)

# New process

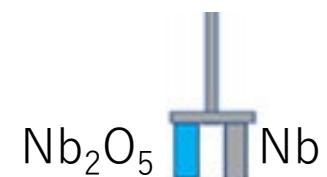
Analogous to Vapor Diffusion:

- Added an **initial Nucleation step** (without  $\text{SnCl}_2$ ) @  $600^\circ\text{C}$  for 3 h



NOTE:

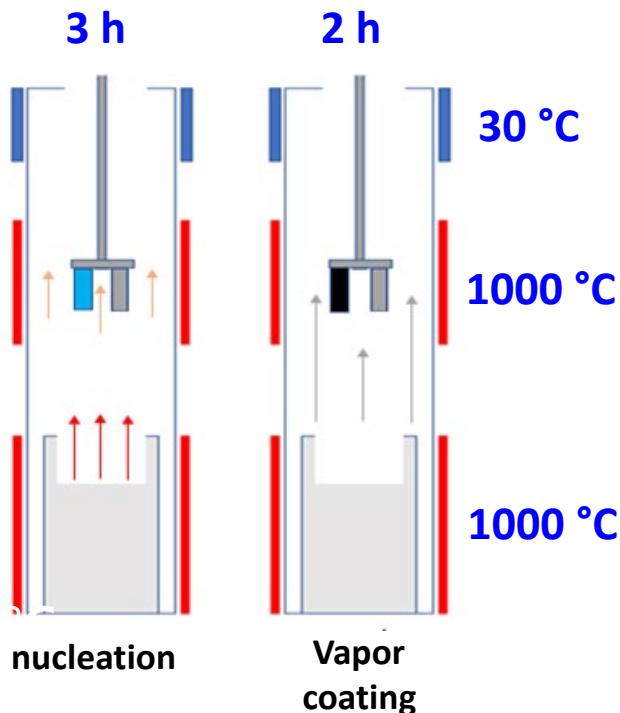
time process steps longer than vapor diffusion due to lower  $T$



# New process

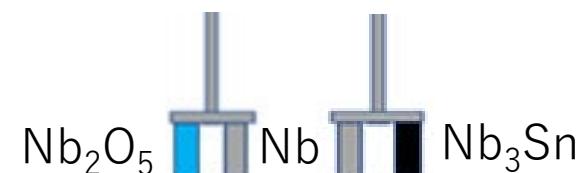
Analogous to Vapor Diffusion:

- Added an **initial Nucleation step** (without  $\text{SnCl}_2$ ) @  $600^\circ\text{C}$  for 3 h
- Added a **Vapor coating step** @  $1000^\circ\text{C}$  for 2 h



NOTE:

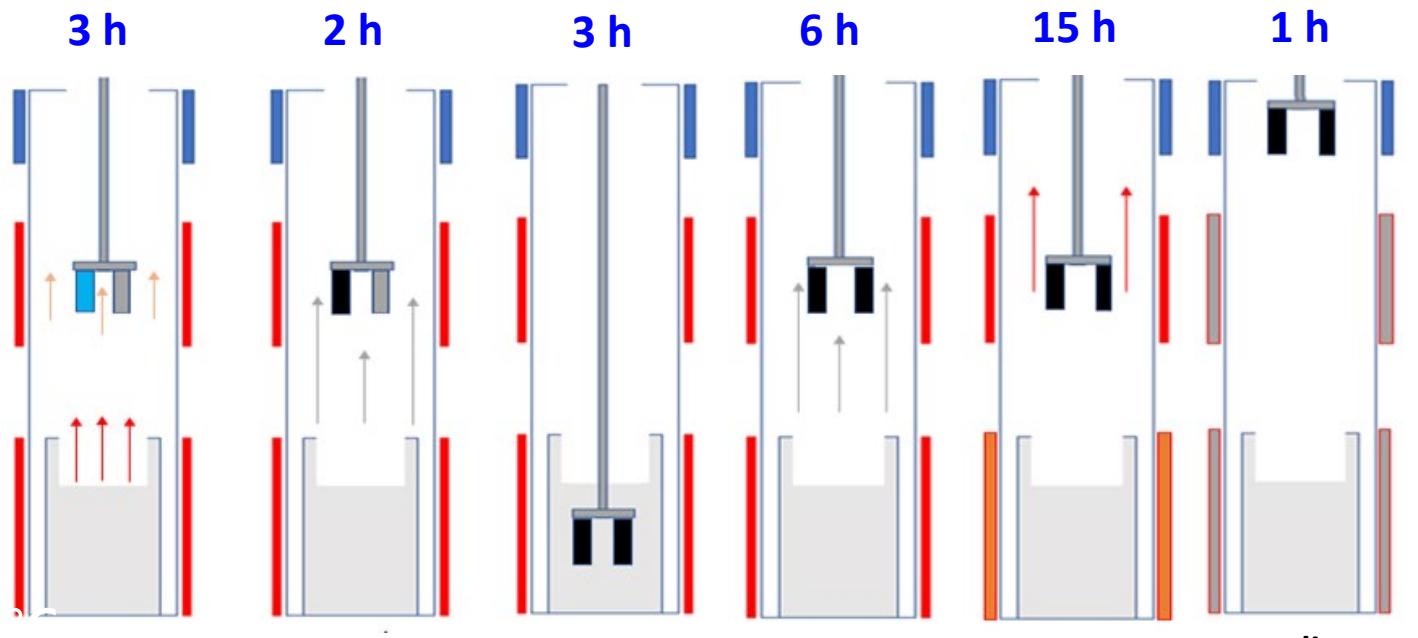
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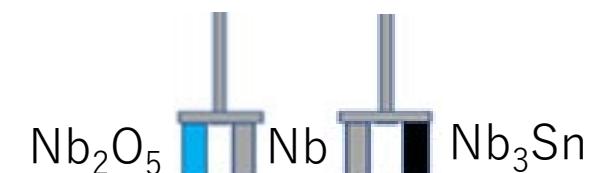
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NOTE:

time process steps longer than vapor diffusion due to lower  $T$



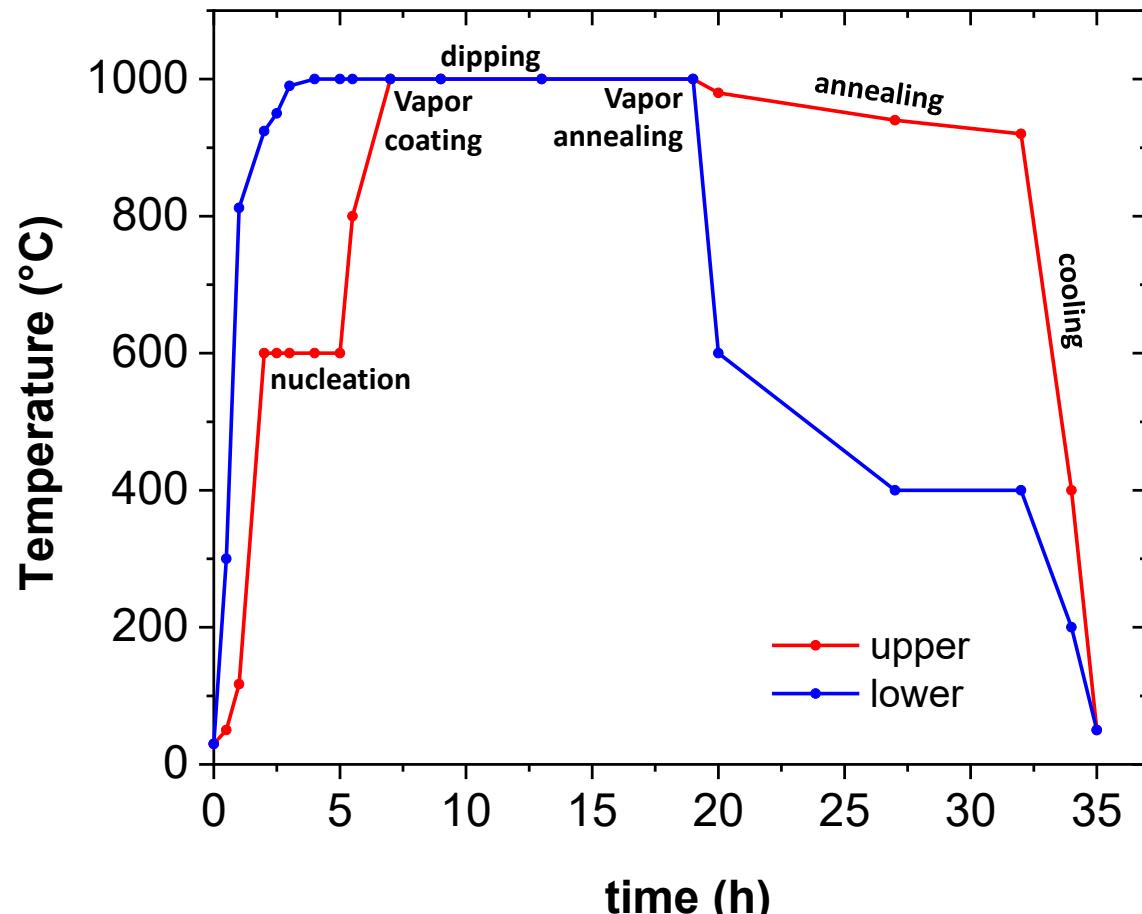
# New process

Analogous to Vapor Diffusion:

- Added an **initial Nucleation step** (without SnCl<sub>2</sub>) @ 600 °C for 3 h
- Added a **Vapor coating step** @ 1000 °C for 2 h
- During the **Annealing step**  
Temperature decreased to ~ 920-940 °C

NOTE:

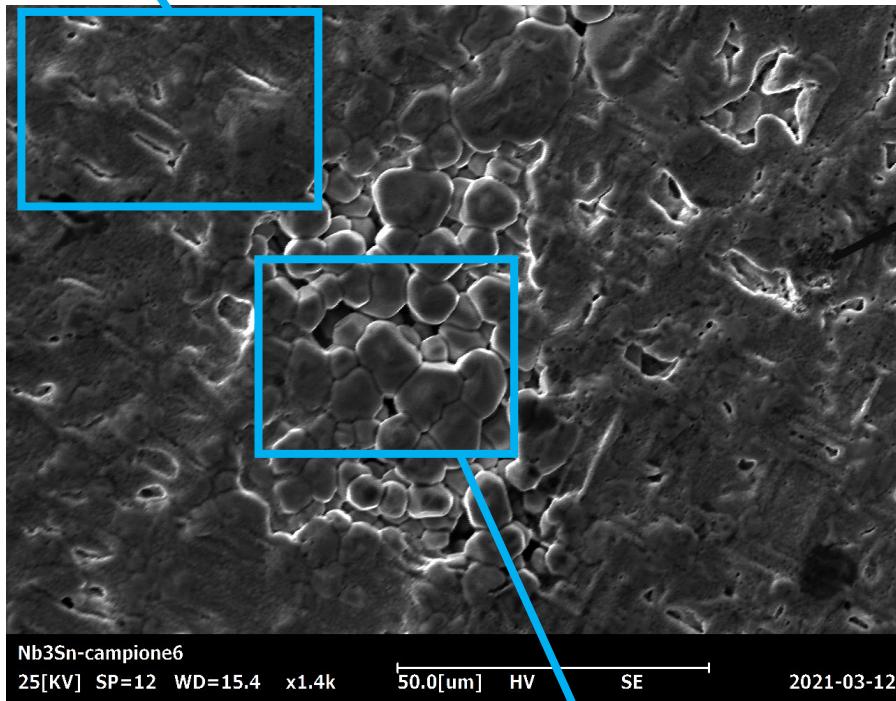
time process steps longer than vapor diffusion due to lower T



Temperature furnaces profile

# Morphology (new process, 2021)

Grains + GB  
Nb 70 %  
Sn 30 %  
Cr 0 %

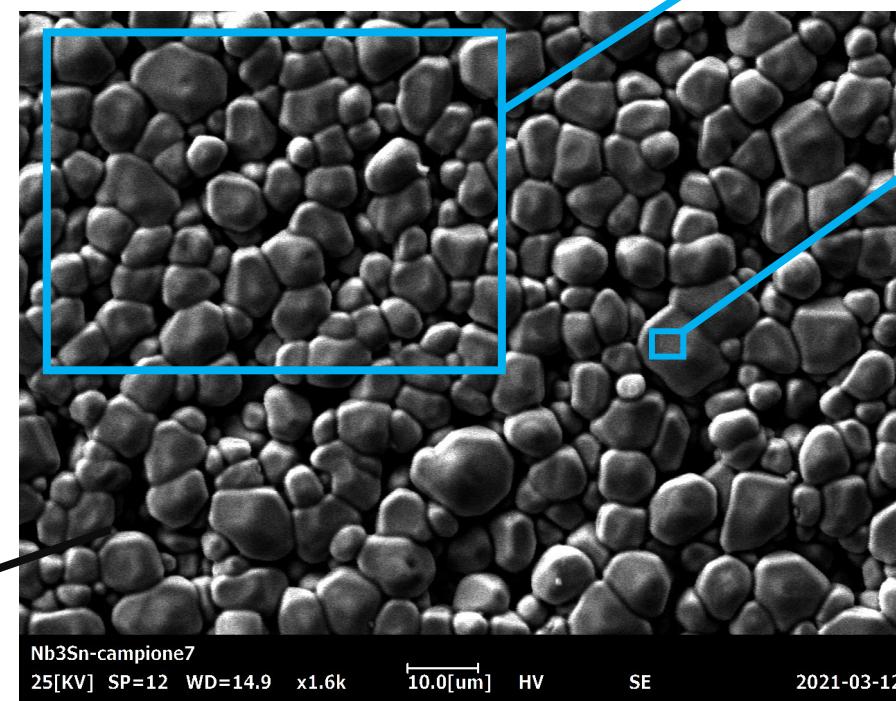


Nb substrate (BCP)

Grains + GB  
Nb 50 %  
Sn 29 %  
Cr 20 %



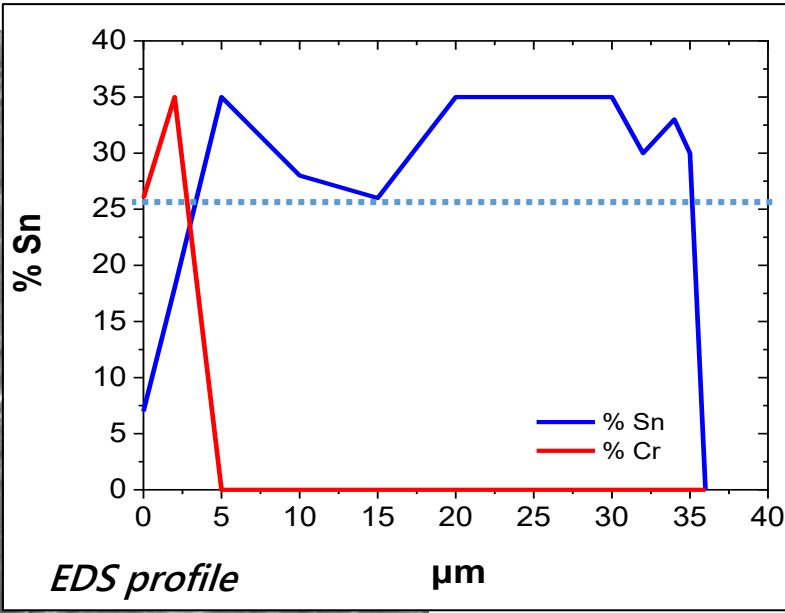
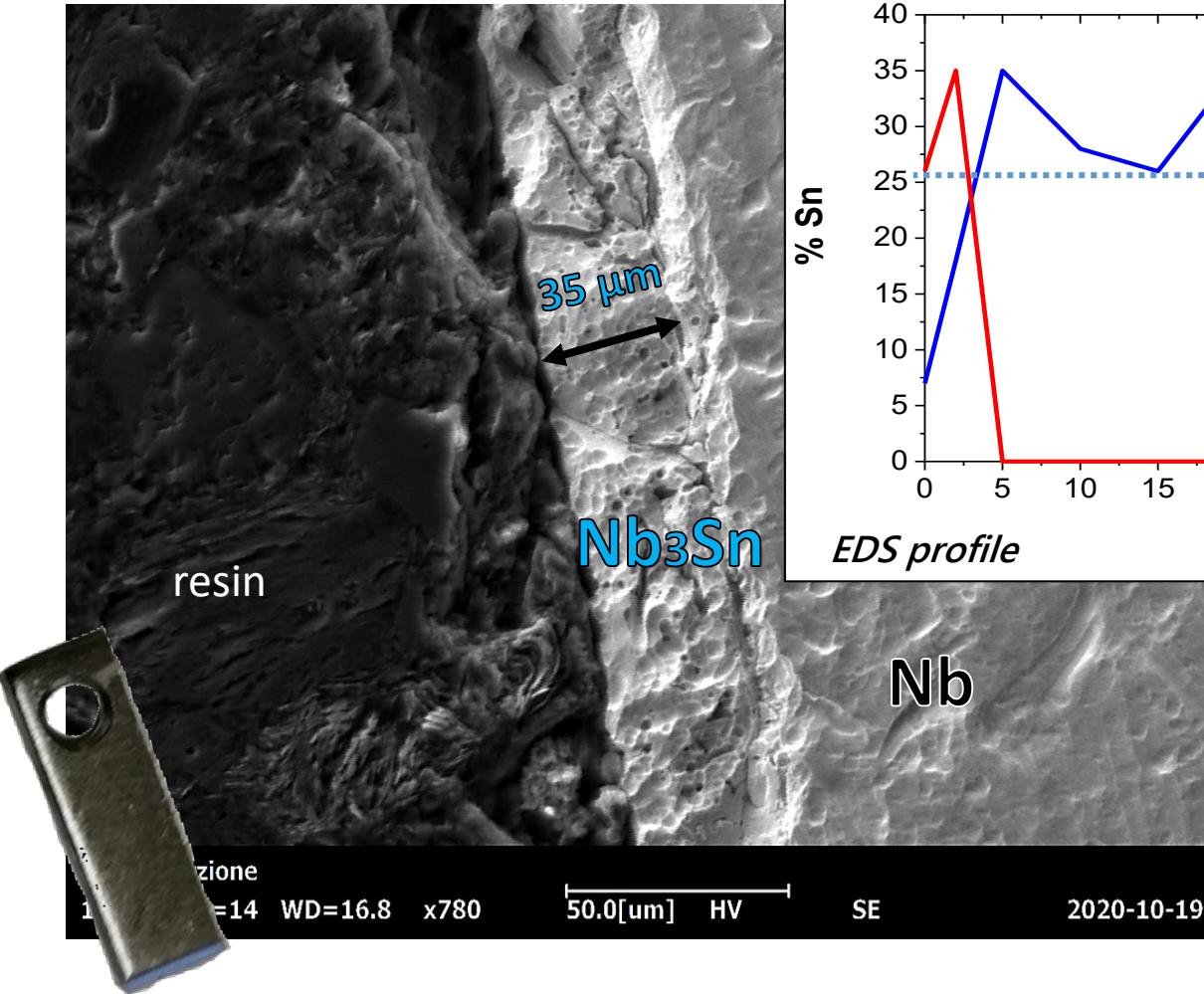
Grains + GB  
Nb 58 %  
Sn 30 %  
Cr 10 %



Nb<sub>2</sub>O<sub>5</sub> substrate (Nb BCP anodized)

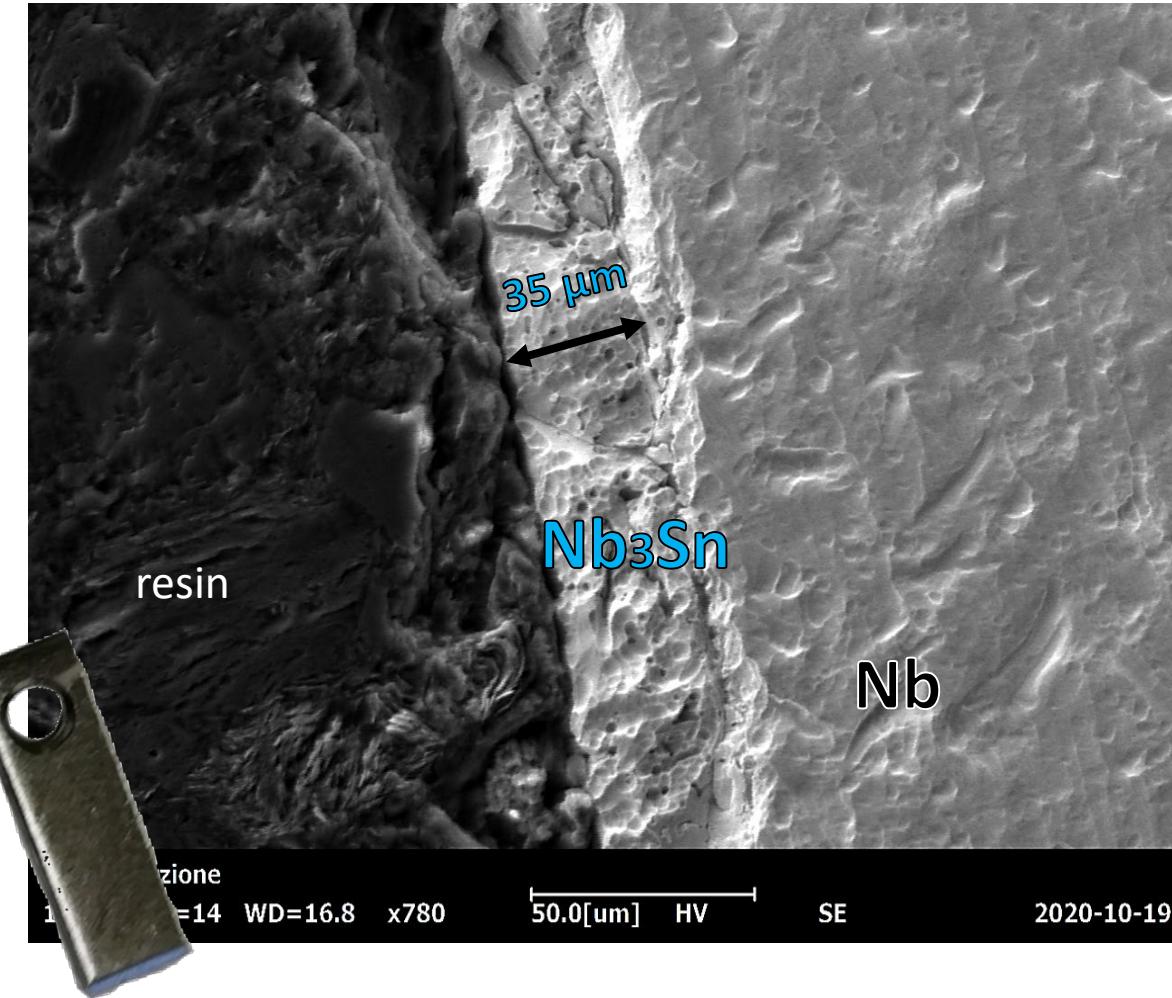


# Chromium contamination



- Contamination is **superficial**
- No contamination in **Sn** crucible
- Presence of Ni and Fe as well → **Inconel Alloy** of vacuum chamber
- Contamination present also in **old samples**
- A problem to face and solve

# Dipping time - thickness relationship



7 hours of dipping → 35 μm

# Dipping time - thickness relationship

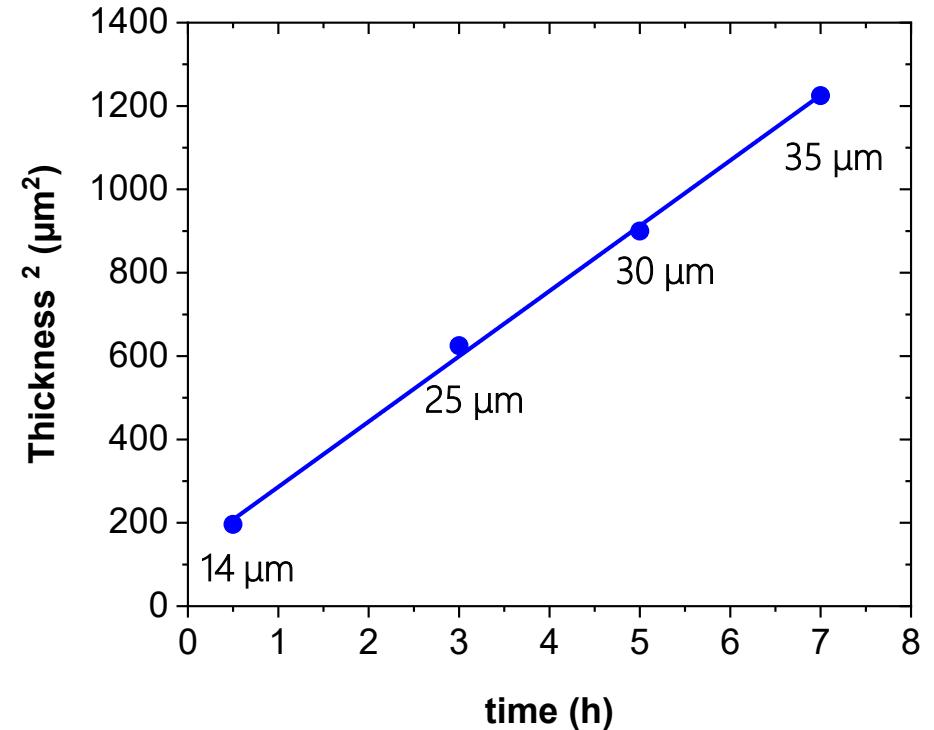
We suppose that the  $\text{Nb}_3\text{Sn}$  growing rate by dipping  
is related to the Sn diffusion in Nb

If it is true, from the Fick's laws of diffusion we have that:

$$x^2 = kt$$

*x:  $\text{Nb}_3\text{Sn}$  thickness*  
*t: dipping time*

*k: a constant that takes in account the diffusion coefficient D*



*Good agreement with the experimental data*

$100 \mu\text{m} \rightarrow \sim 63 \text{ hours of dipping!}$

$24 \text{ hours of dipping} \rightarrow \sim 62 \mu\text{m}$

$$x^2(\mu\text{m}^2) = 157 \cdot t(\text{h}) + 130$$

$$R^2 = 0.9983$$

# Proof of concept

- Preparation of a 1" target (30 microns thick)
- Coating on quartz samples

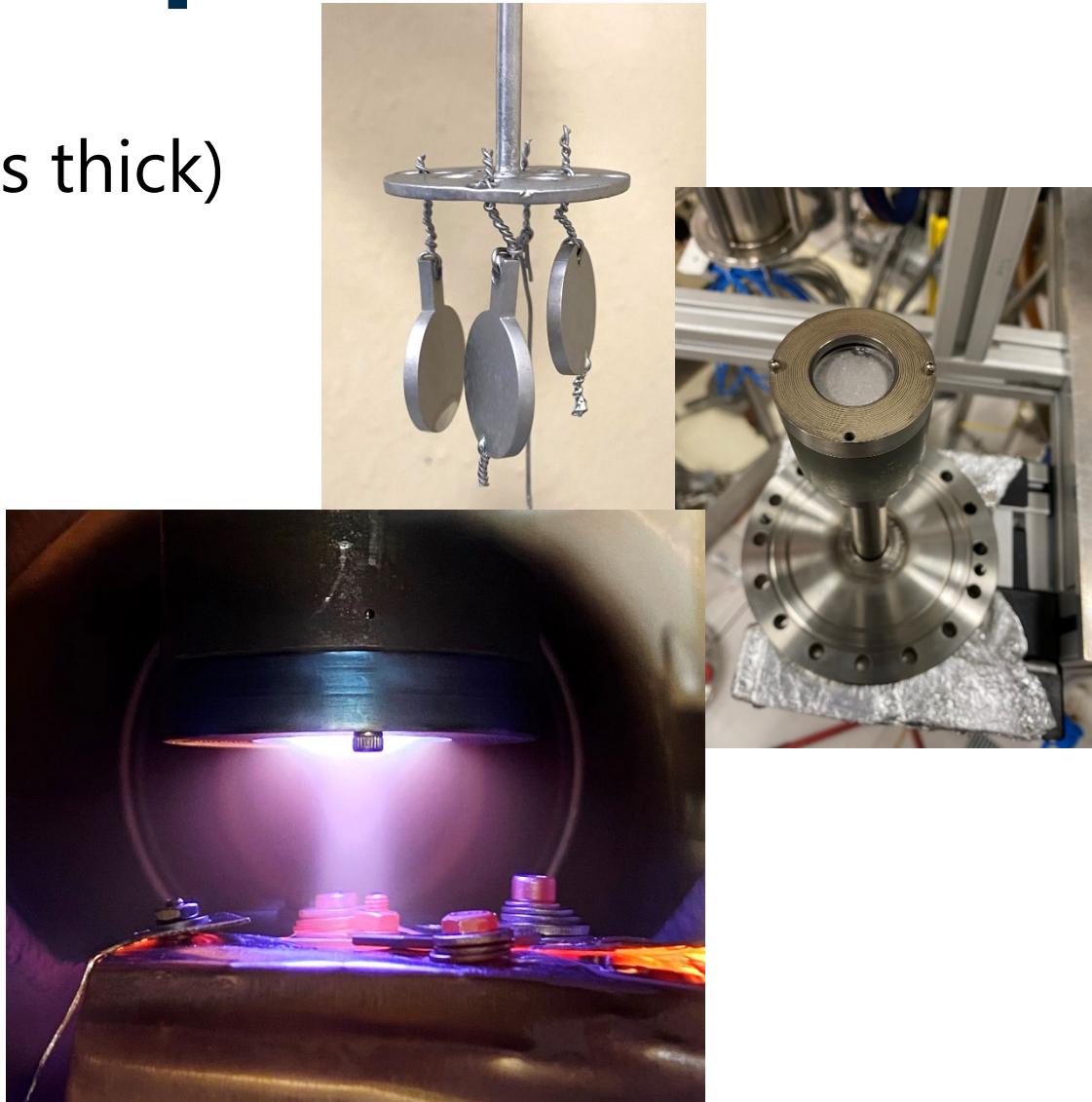
$I = 0.1 \text{ A}$  ( $5 \text{ mA/cm}^2$ )

$t = 30 \text{ min}$

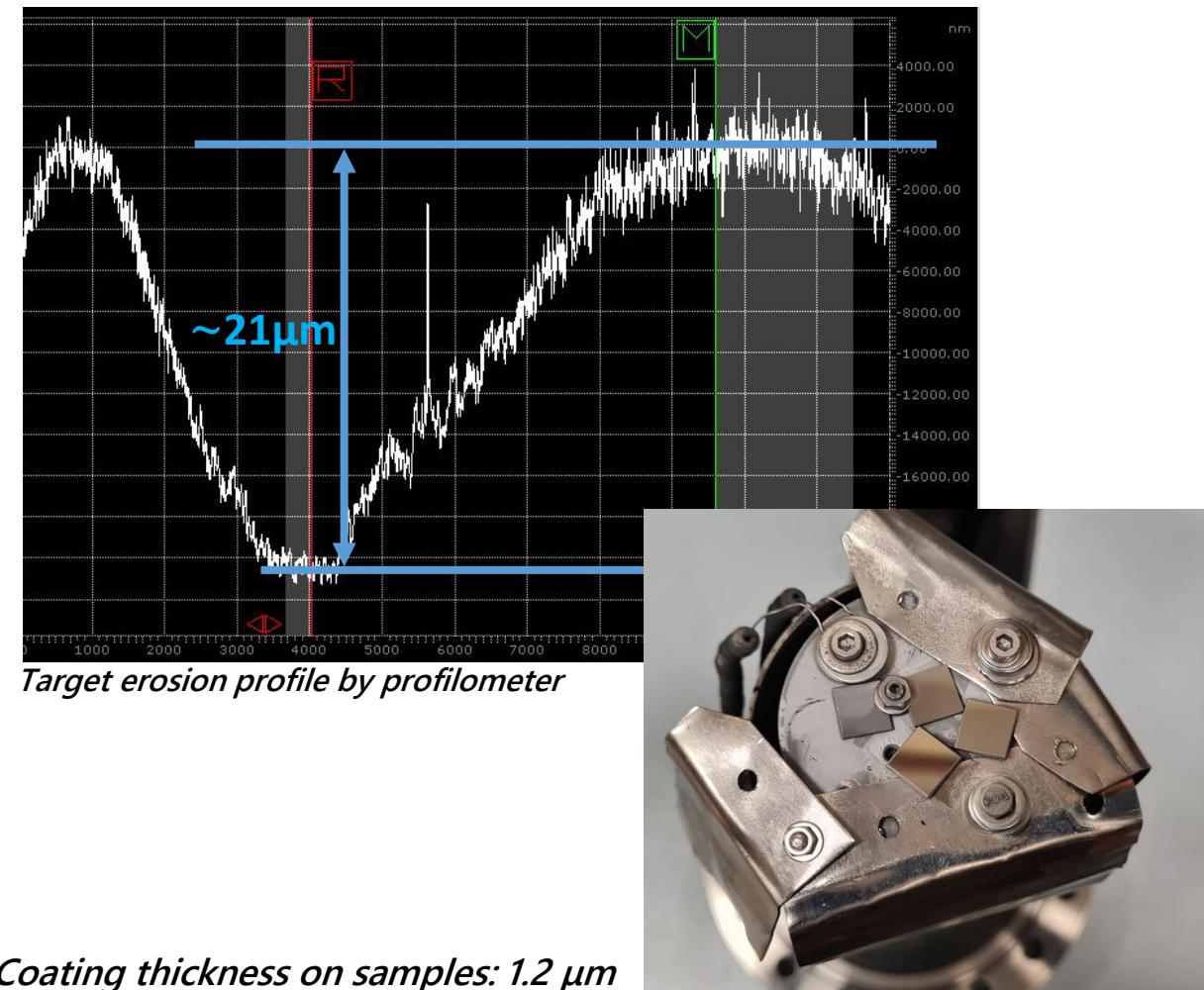
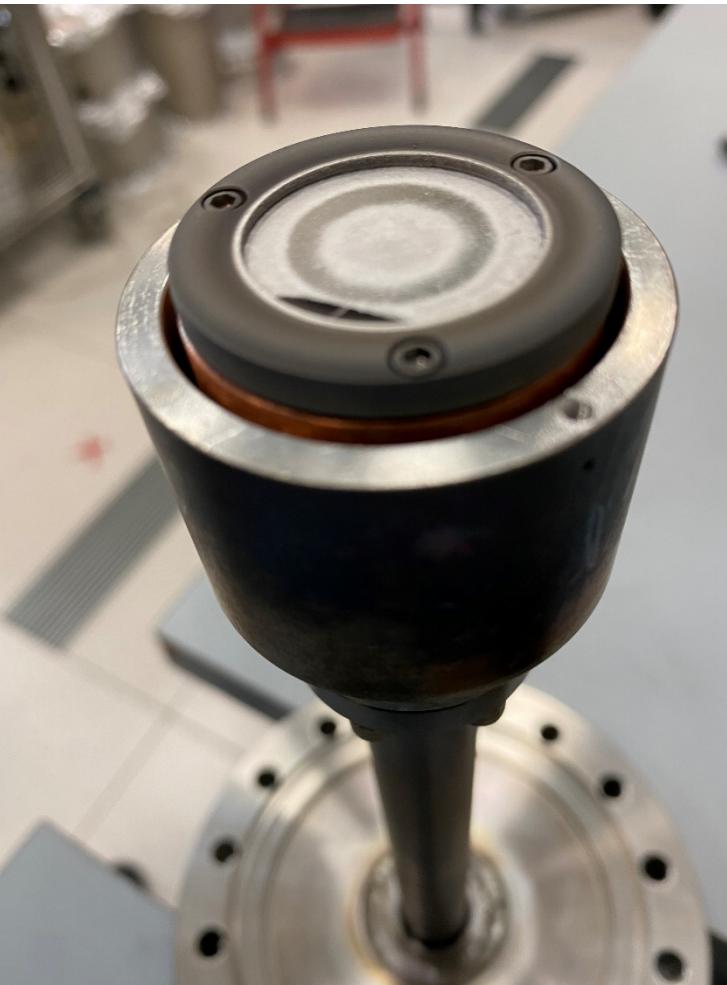
$T = 750 \text{ }^\circ\text{C}$

$P_{\text{base}} = 2 \cdot 10^5 \text{ mbar}$

- Process stopped when V decreased

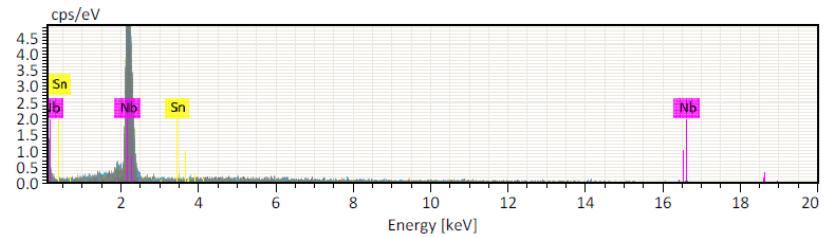
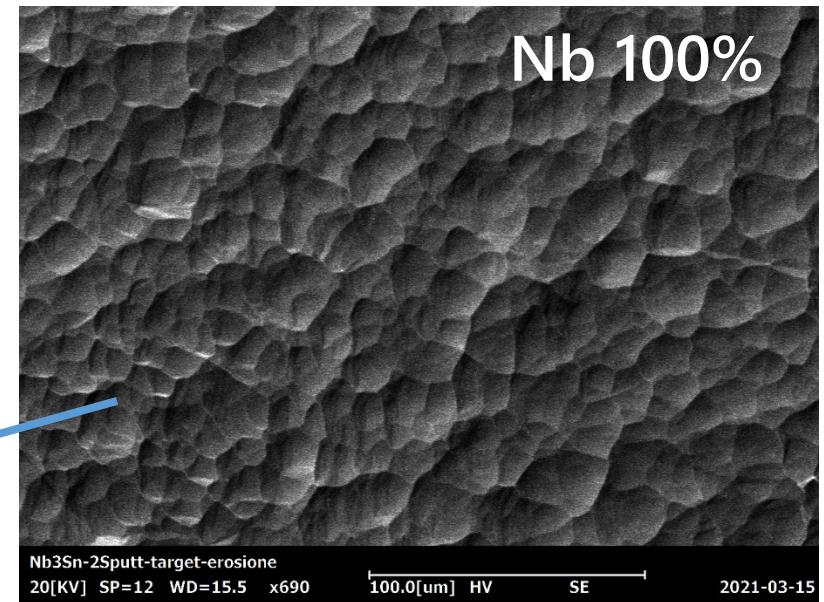
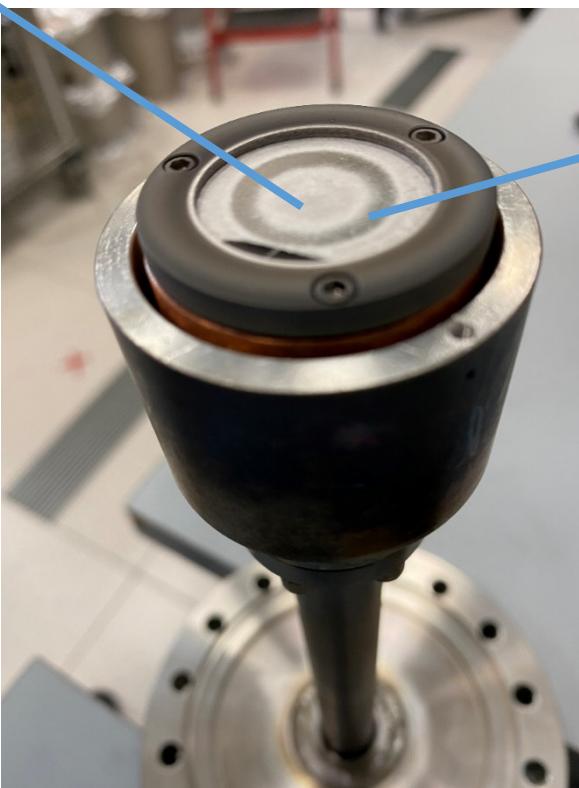
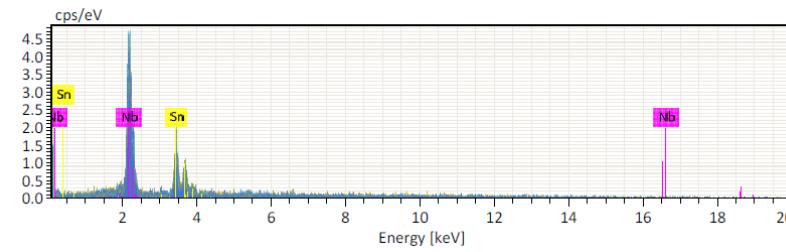
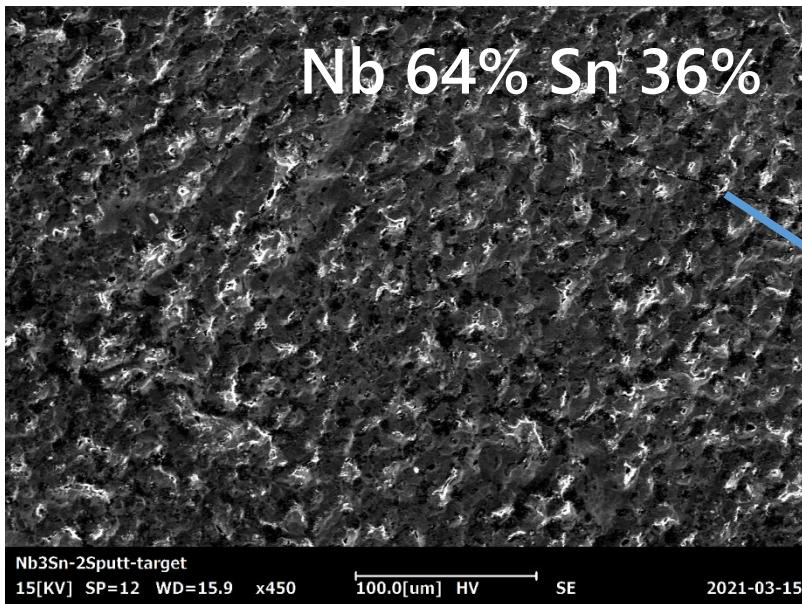


# Results - Target erosion

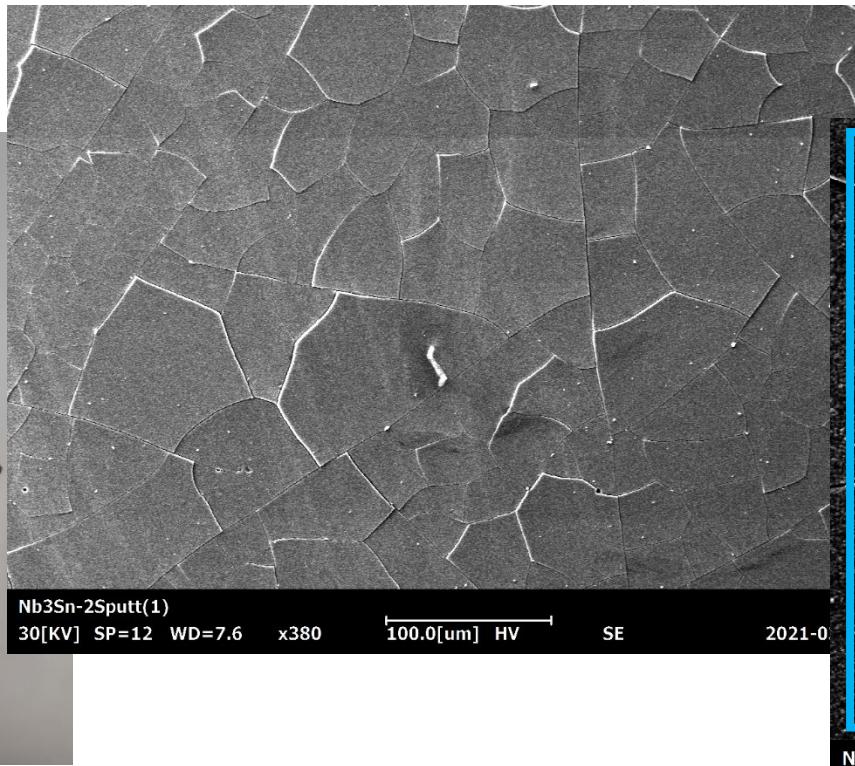
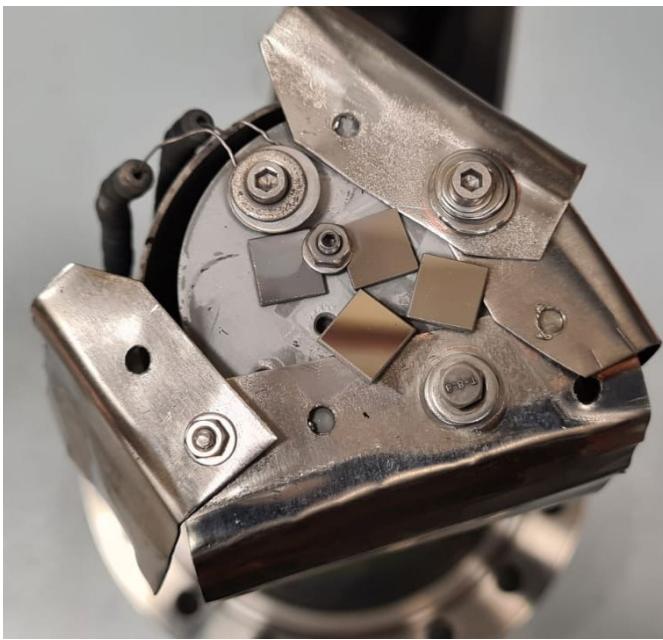


*Coating thickness on samples: 1.2  $\mu$ m*

# Results target



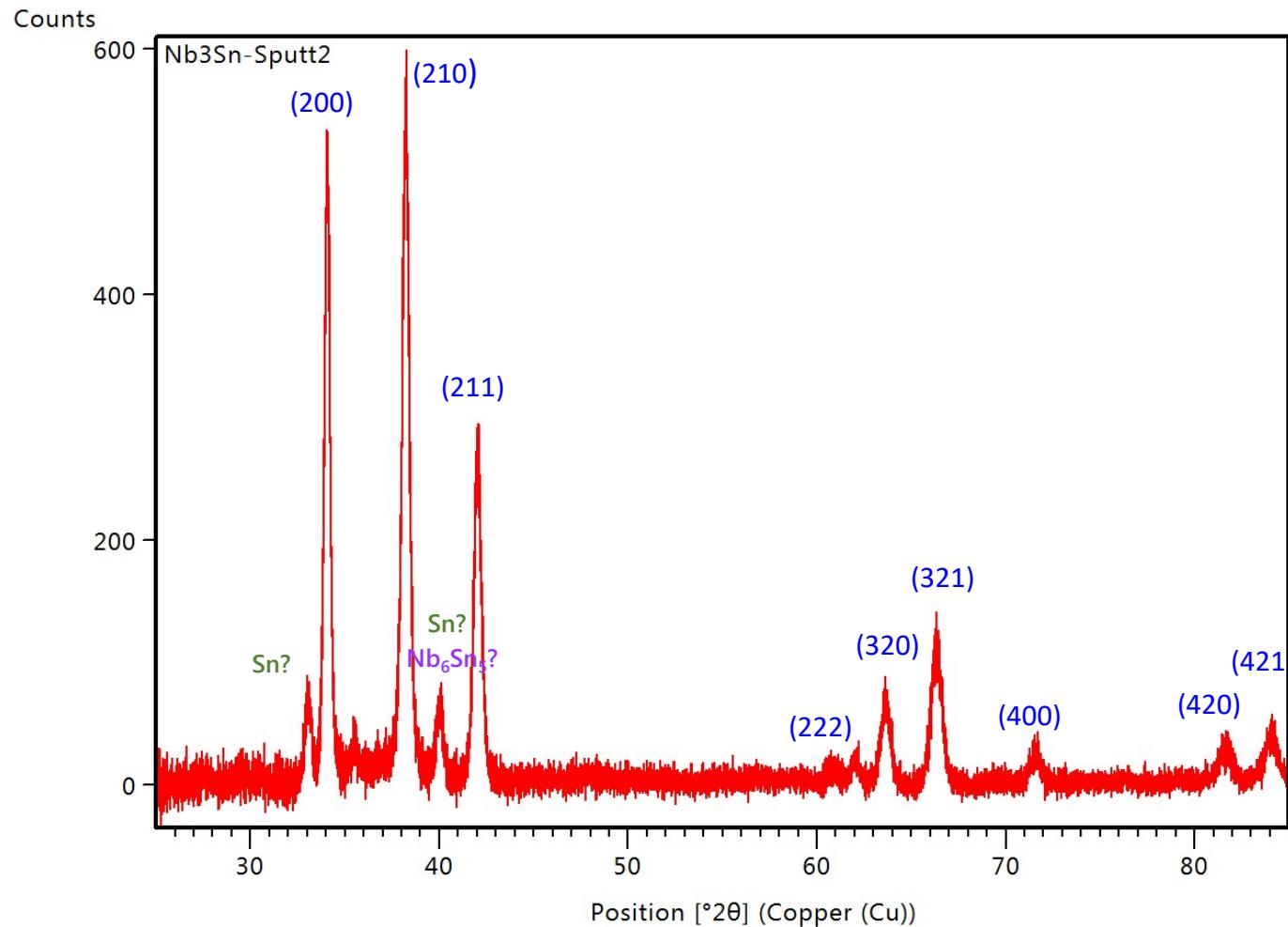
# Sputtered samples



Strange behaviour R vs T → not possible measure Tc  
No optimized deposition conditions

Nb 74% Sn 26%

# Sputtered sample XRD



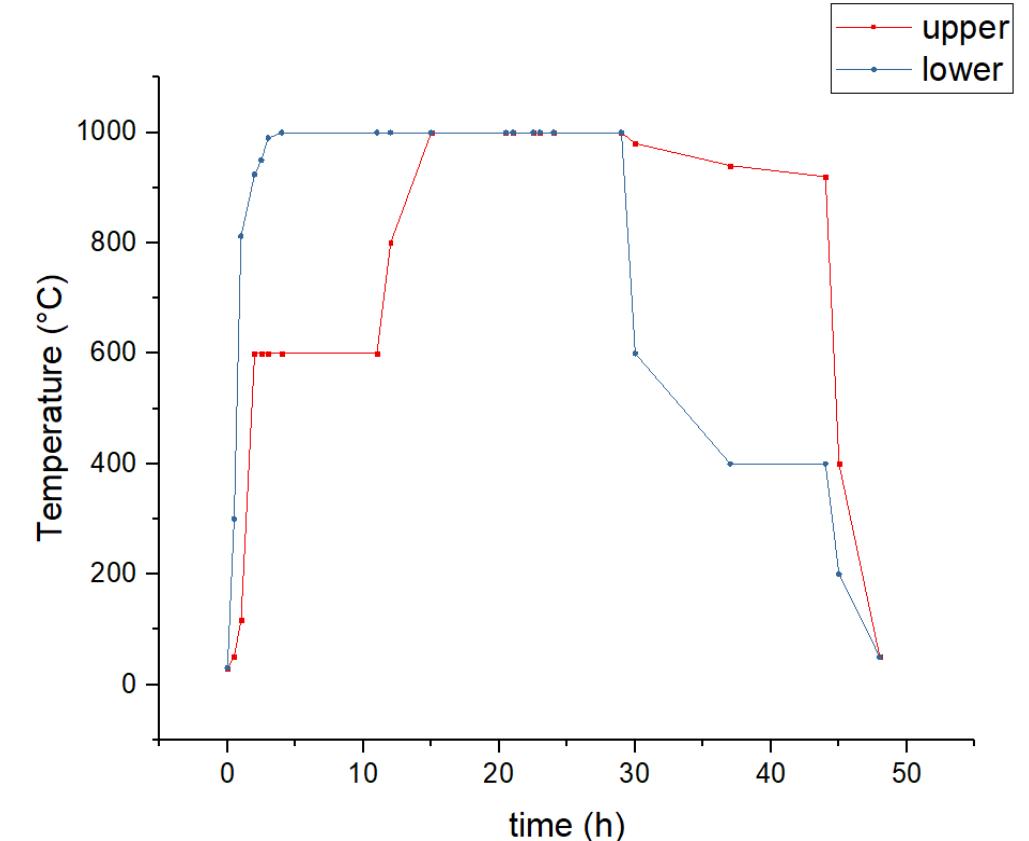
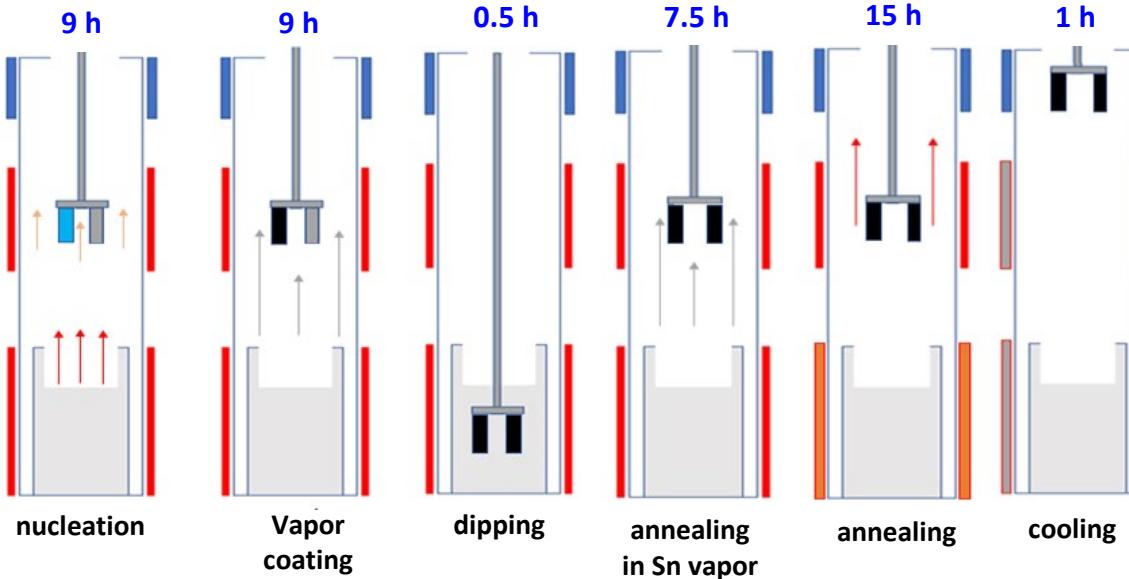
XRD shows  $\text{Nb}_3\text{Sn}$  phase  
Presence of unreacted Sn  
and/or  $\text{Nb}_6\text{Sn}_5$  spurious phase

# How does the new process work in RF?

# How does the new process work in RF?

Process has been adapted to reduce  $\text{Nb}_3\text{Sn}$  thickness

(5-10 microns as the old cavities)



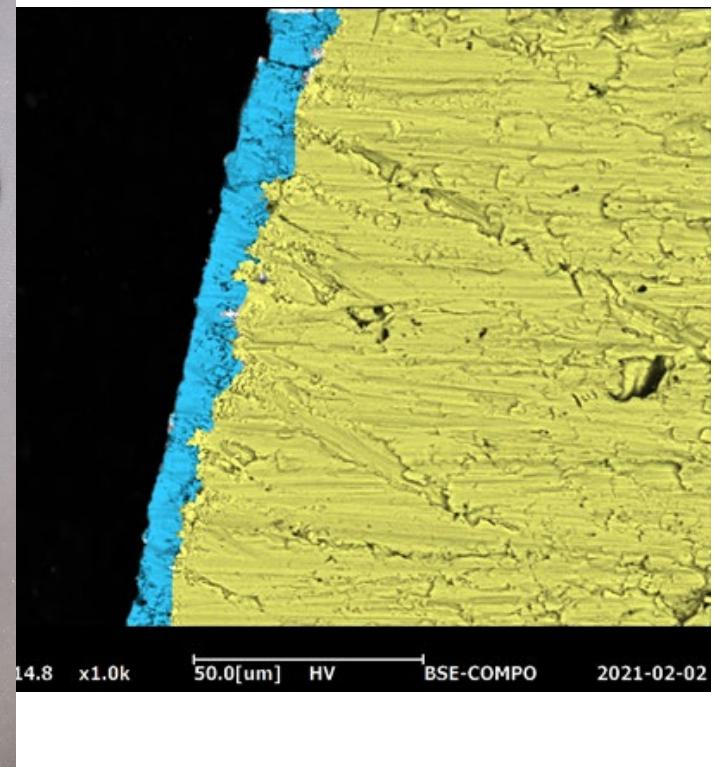
# 6 GHz cavity #1 (only geometry test)

First test in a broken cavity to  
test possible geometry effect

No tin drops!

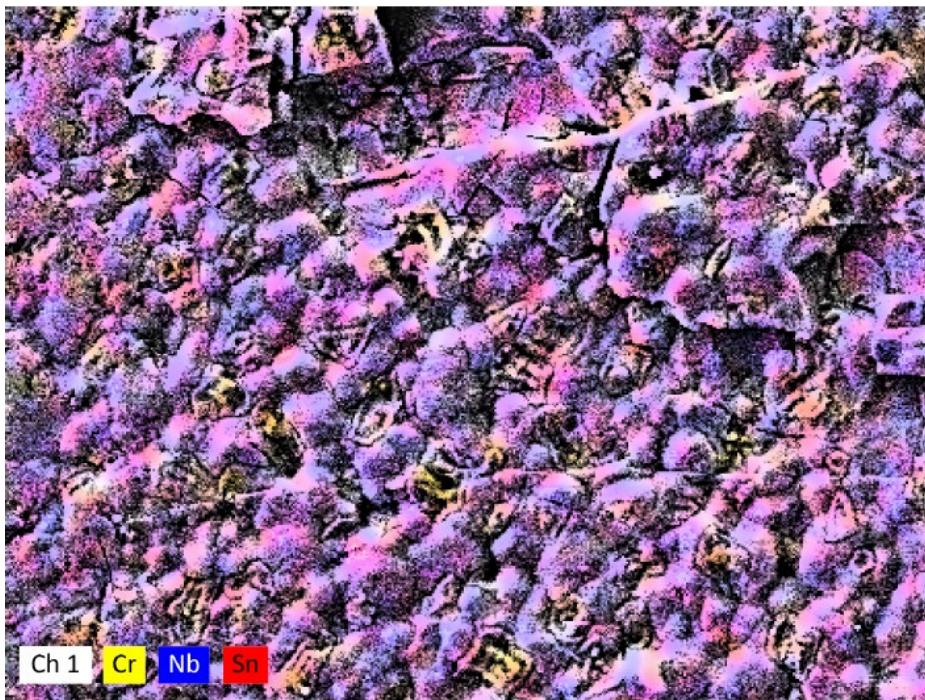
Very good surface

5-10 micron  $\text{Nb}_3\text{Sn}$  thickness



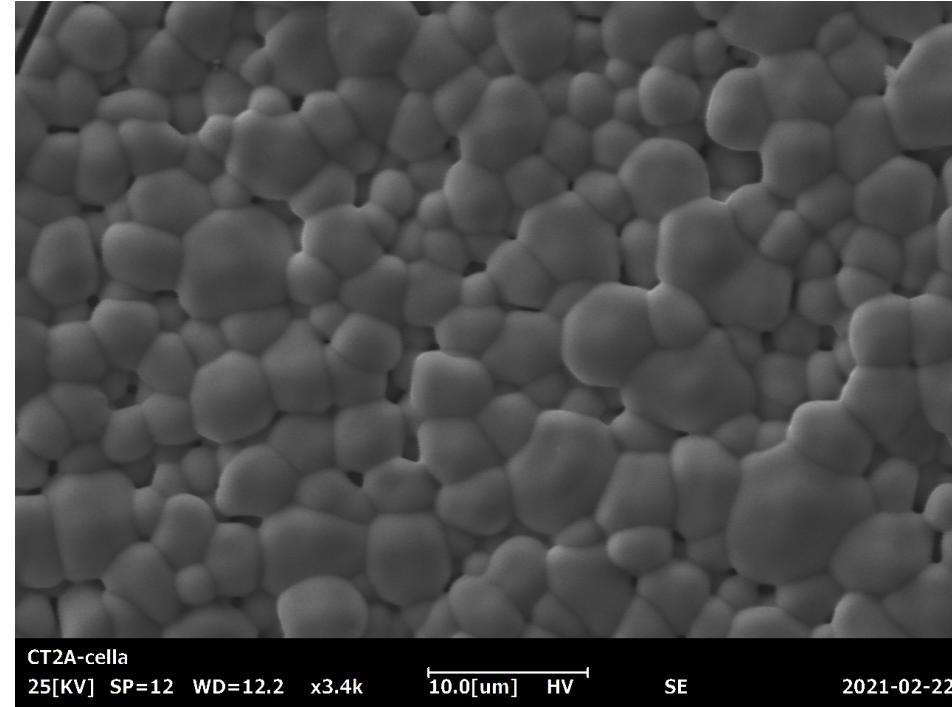
# Cr-Ni contamination in 6 GHz cavity

External Surface of a 6 GHz elliptical cavity



Nb 42% Sn 13% Cr 38% Ni 7.2%

Inner Surface of a 6 GHz elliptical cavity

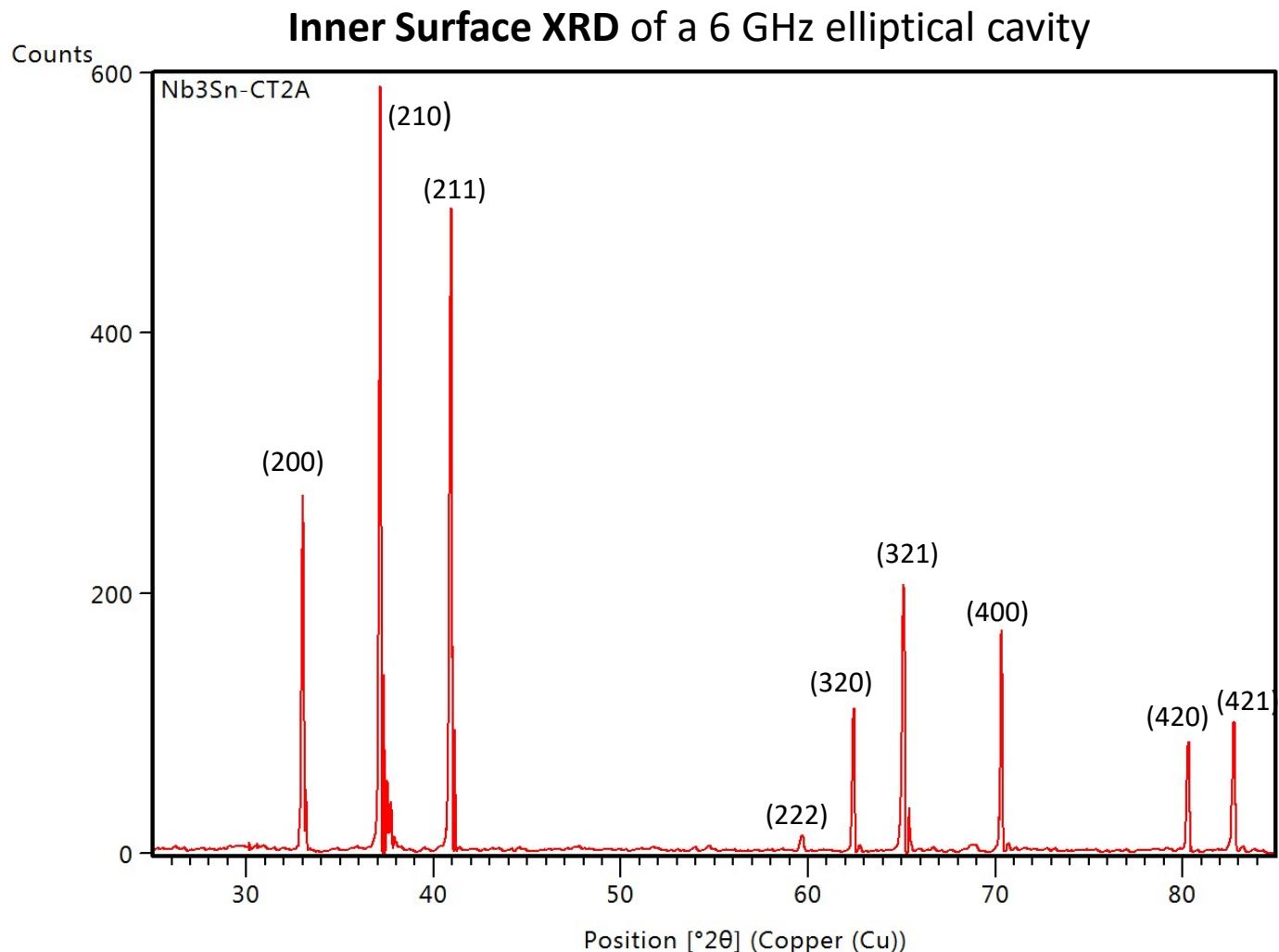


Nb 65% Sn 35% Cr 0%

The cavity **geometry preserves** the **inner surface** from **contamination**

# 6 GHz cavity #1 (only geometry test)

XRD shows only the  
**Nb<sub>3</sub>Sn phase**



# 6 GHz cavity #2

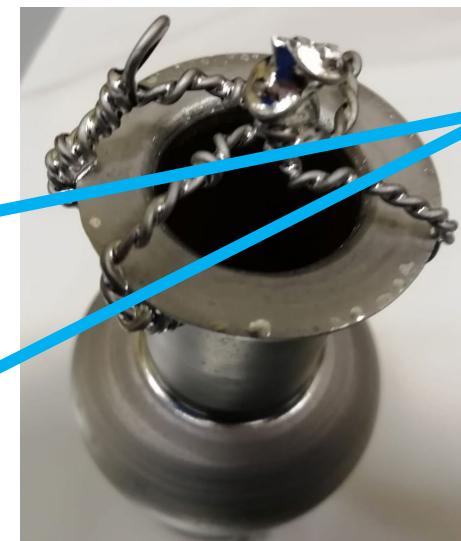


Same process of cavity #1

Problem with upper furnace: annealing at 920 °C instead of 940 °C



45 h of annealing instead of 15 h

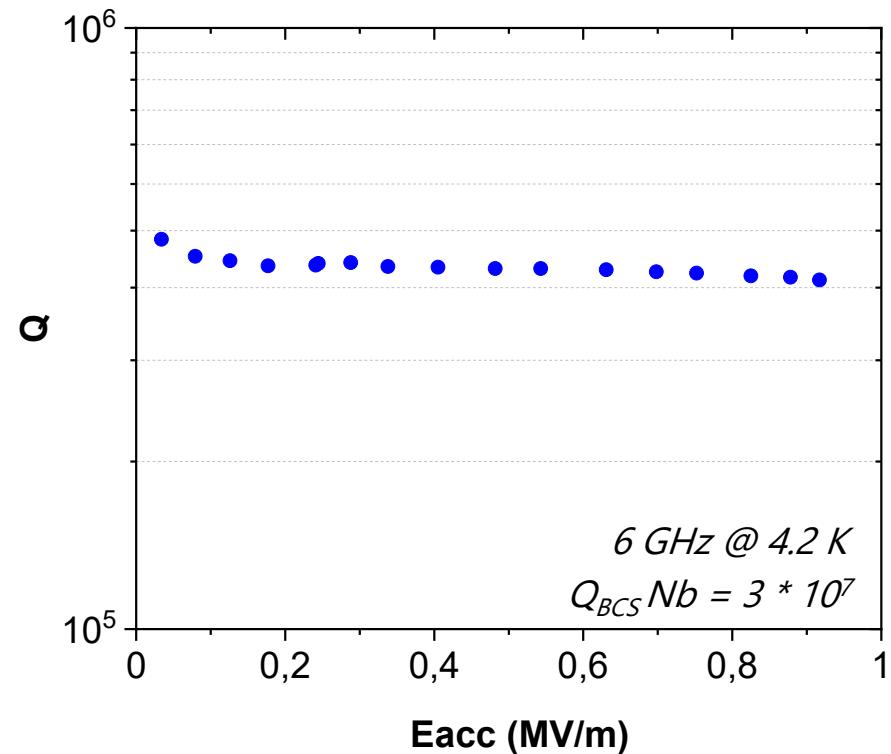
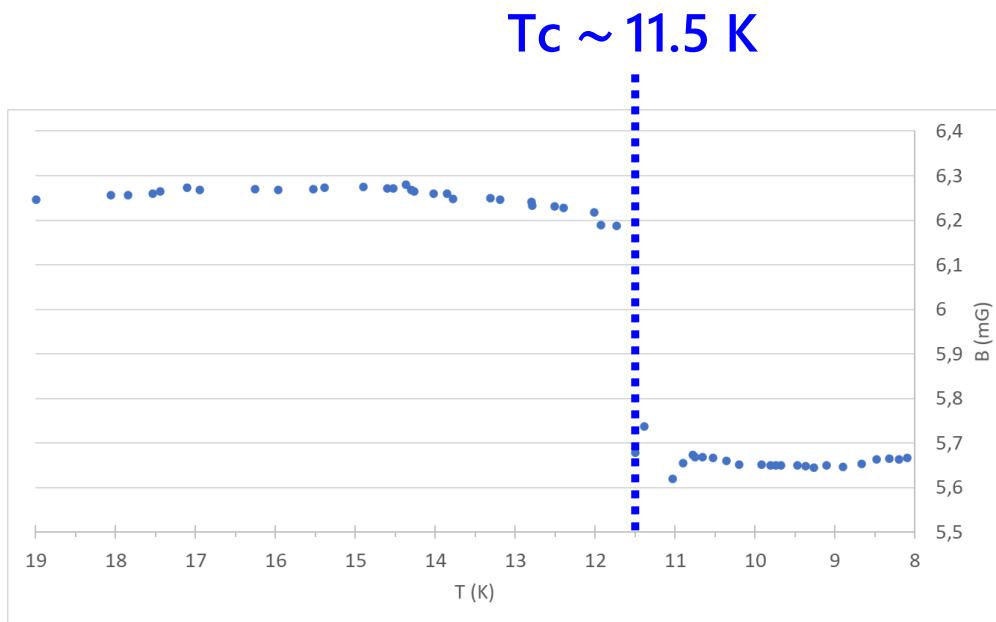


Metallic tin outside

Dark spots in the inner surface

(signs of tin excess)

# Tc and RF Test



Poor RF performances comparable to old process  
(Sn excess? Cr-Ni contamination? Low T? Thick film?)

# Conclusions

- Possibility to grow thick film → target production
- Good Stechiometry
- Proof of concept passed (*sputtering process must be optimized*)
- Cr-Ni contamination to solve (*Nb screen or New Nb chamber*)
- Poor SRF performance  
(Sn excess? Cr-Ni contamination? Low T? Thick film?)

# Thank you for the attention!



Material Science and Technology Service for Nuclear Physics crew

*Special thanks to*  
Matteo Zanierato, Vanessa Garcia, Eduard  
Chyhyrynets, Roberta Caforio, Fabrizio Stivanello

[cristian.pira@lnl.infn.it](mailto:cristian.pira@lnl.infn.it)

